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Simulations of diamond nucleation in carbon fullerene cores

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Recent experiments have shown that heavy ion or electron irradiation induces the nucleation of diamond crystallites inside concentric nested carbon fullerenes, i.e., bucky onions. This suggests that the fullerene acts as a nanoscopic pressure shell. In this paper we study the formation of tetrahedrally bonded carbon inside a prototype icosahedral two-shell fullerene by means of atomic-scale computer simulations. After the simulated irradiation, we can identify regions in which almost all carbon atoms become $sp^3$ bonded. Additionally, we observe a counteracting tendency for the carbon atoms to form shell-like substructures. To shift the balance between these two processes towards diamond nucleation strongly nonequilibrium conditions are required.

It has been discovered that fullerenes can form multilayered onion structures, when carbon soot is irradiated by electrons. Further electron or heavy ion irradiation causes the fullerenes to transform into diamond near their centers. The irradiation process is viewed to induce knockon displacements, collision cascades, and flux of C atoms towards the core, resulting in increase in particle density, local pressure, and temperature. Finally, the conditions become preferable for the nucleation of diamond. It is necessary for the nucleation that the onion shells possess enough rigidity to withstand the increase in pressure and temperature in order to prevent a relaxation towards a graphitic core. Zaiser and Banhart have presented a thermodynamical quasiequilibrium theory to explain this irradiation-induced transformation to diamond. In this paper, we report results of atomic simulations of the collision-induced phenomena inside carbon fullerenes. These show the nucleation and growth of diamondlike structures.

The simulations are performed using the density-functional-based tight-binding method (DF-TB). The method has been successfully applied for studies of various systems in particular for carbon clusters such as small molecules, hydrocarbons, fullerenes, and extended allotropes such as amorphous and crystalline structures. Here we study the diamond nucleation process inside the smallest ideal carbon onion which consists of 300 atoms. This onion consists of two concentric fullerene shells, a $C_{60}$ ball and a surrounding icosahedral $C_{240}$ shell with average radii of 3.6 and 7.1 Å, respectively. We consider this low-energy allotrope as a likely core of real giant fullerenes and simulate the release of atoms due to knockon displacements in outer shells and their transport to this core by a sequence of random atom additions with zero initial velocities. The structural evolution is examined using a combination of molecular dynamics (MD) and a conjugate-gradient (CG) optimization technique and in dependence upon the number of (interstitial) atoms during an *ad hoc* injection.

To speed up the very demanding calculations, the fullerene core has originally been seeded with a ten-atomic cluster consisting of two interconnected five-membered rings. We find this atomic arrangement to be remarkably stable during an extensive phase-space search for likely atomic inclusions inside icosahedral $C_{60}$. We model the irradiation by randomly adding further interstitial carbon atoms to the fullerene. The number of added atoms are varied in order to minimize the influence of simulation details and initial settings upon the final result. We relax the growing structure between the additions to reflect that the relaxation rate is much higher than the damage creation rate. Finally, we refine our study by simulating the interstitial injection with an unseeded core. Note that the atom numbers given below always reflect the number of interstitials added to either of the starting structures (i.e., the $C_{10}$ seeded or the empty fullerene).

During simulation series A, 50 carbon atoms were initially placed *ad hoc* into random locations between the $C_{240}$ and seeded $C_{60}$ shells. The relaxation was at first carried out by microcanonical MD for 240 fs with vanishing initial velocities. After equilibration the system reached a temperature of about 2100 K due to the atomic interactions. The system was then brought into contact with a heat bath and linearly cooled down to 1300 K with a rate of $0.69 \times 10^{15}$ K/s, using a simple velocity-scaling technique. Ten more atoms were added to random positions inside the fullerene. The addition was completed by microcanonical MD equilibration and thermalization at 2000 K for another 240 fs and an annealing towards 1300 K using the same cooling rate as above. After another such injection cycle the total number of interstitials amounted to 70. To simulate more intense irradiation and very rapid thermal dissipation to the surroundings, we continued the filling by adding sets of 20 C atoms at random positions inside the structure. Again, we initially equilibrated the structure and thermalized it at 2000 K for 240 fs. Then, we CG relaxed the final snapshot structure of the high-temperature phase which yields a low-energy metastable configuration in phase space at 0 K. The convergence criterion for the CG relaxations was a maximum rms force of 0.001 atomic units. We repeated such filling cycles until 150 extra atoms were placed into the fullerene.

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The C240 shell which takes up defects and/or becomes bonded is presented in Table I and in Fig. 1. The snapshots at the end of the the CG relaxation in simulation series A. The sp³ atoms are in dark color.

<table>
<thead>
<tr>
<th>Hybridization</th>
<th>Threefold nonplanar</th>
<th>sp³</th>
<th>Average coord.</th>
</tr>
</thead>
<tbody>
<tr>
<td>70 added atoms</td>
<td>Whole structure</td>
<td>63.8</td>
<td>20.4</td>
</tr>
<tr>
<td></td>
<td>Center (28 atoms)</td>
<td>50.0</td>
<td>17.9</td>
</tr>
<tr>
<td>90 added atoms</td>
<td>Whole structure</td>
<td>68.3</td>
<td>11.4</td>
</tr>
<tr>
<td></td>
<td>Center (30 atoms)</td>
<td>40.0</td>
<td>20.0</td>
</tr>
<tr>
<td>110 added atoms</td>
<td>Whole structure</td>
<td>55.8</td>
<td>15.5</td>
</tr>
<tr>
<td></td>
<td>Center (34 atoms)</td>
<td>11.8</td>
<td>20.6</td>
</tr>
<tr>
<td>130 added atoms</td>
<td>Whole structure</td>
<td>46.5</td>
<td>14.5</td>
</tr>
<tr>
<td></td>
<td>Center (39 atoms)</td>
<td>12.8</td>
<td>10.3</td>
</tr>
</tbody>
</table>

In the beginning of the simulation series A the C60 shell is destroyed and a larger shell-like structure consisting of ~100 atoms forms. This shell has a radius of about 4–5 Å and initially leaves the outer C240 shell intact. This special atomic arrangement in the center appears to be quite stable under further injections. Clearly, a few atoms added avoid a location near the outermost C240 shell. The intershell distance between C240 and the emerging 100-atom shell is ~2.5 Å, which could explain the experimentally observed intershell distance of 2.2 Å in fullerene cores during early stages of irradiation. Upon increase of the atom number condensation takes not only place in the outermost C240 shell. However, by the same time most of the fourfold-coordinated atoms is about 11%. Upon increase of the atom number condensation takes not only place in the shell but also in the very core of the structure. The sp³-bonded atoms appear at this stage rarely correlated at various places in the carbon network.

Continuing the filling with sets of 20 atoms leads to a breakdown of the interior shell structure. Since atoms added during these cycles do not have enough time and space to relax into fullerene-type arrangements, the condensation of fourfold-coordinated atoms is triggered. Interestingly, we can now identify an sp³-rich cluster which binds to the outermost C240 shell. However, by the same time most of the inner atoms avoid the region near the outer shell indicating its stability due to π bonding. The evolution of the model and its core cluster in terms of the number of three- and fourfold-coordinated atoms is presented in Table I and in Fig. 1. For an inner region with 8 Å diameter, the sp³ fraction appears to increase sharply for 90–110 atoms added.

As mentioned above, the diamond nucleation starts near the C240 shell which takes up defects and/or becomes bonded to the atomic arrangements in the core. As more atoms are added, sp³-rich domains emerge and grow towards the center. There they fuse and form a highly fourfold-coordinated condensate as is indicated in Table I by the large fraction of sp³ atoms in the central areas of series A structures, see also Fig. 1. Figure 3 shows the total number of sp³-hybridized atoms in the largest sp³-bonded cluster during the addition. The neighboring atoms are considered to belong to a same cluster if they share a sp³ bond.

While the total number of sp³ bonded atoms declines slightly by about 5% when the filling is continued beyond 130 added atoms, the sp³ concentration near the core region becomes saturated. This implies that the diamondlike structure, once nucleated, remains fairly stable under further injection of atoms even if the outer shell becomes disrupted.

The outermost shell of the fullerene develops a considerable number of defects during the irradiation simulation. These locally cause outward expansion of the shell, leading to pressure relief. Another mechanism to reduce strain in the core is emission of carbon dimers after their thermal activation. Both effects are contrary to diamond nucleation. However, in a real giant fullerene we expect that the manifold of fullerene layers supports the structural confinement of the core. Hence, the undesired stress relief effects become less important and the balance is shifted towards increase of the diamond yield.

The validity of this expectation was checked in a second simulation series B, in which 70 and 110 carbon atoms, respectively, were introduced to random locations at once inside the seeded C300 onion structure. During this simulation...
In order to rule out a possible bias in the results due to the initial seeding or the setwise addition of atoms we, finally, performed another simulation series C, in which the extra atoms were added one after another at random positions inside the unseeded C$_{300}$. The whole C$_{300+2}$ system was CG relaxed after each addition without any constraints. The number of fourfold-coordinated atoms saturates at about the same levels as for simulation series A ($\sim$33% in total and $\sim$70% in the center). This indicates that the final results of the simulations remain nearly the same while using different simulation procedures (see also Table III).

However, at the early stages of the fullerene filling the structures differ in the simulations A, B, and C. During the first steps of the simulation series C the inner C$_{60}$ shell becomes filled with chainlike structures rather than the more...
stable cluster arrangement assumed in series A. Furthermore, the nucleation positions and growth directions of the \(sp^3\) regions differ between simulation series A and C. The evolution of the \(sp^3\) region in simulation C is depicted in Fig. 2. The \(sp^3\) atoms tend to aggregate as soon as they emerge. They start to form on the surface of the inner C\(_{60}\) shell by transformation of the \(sp^2\)-type atoms in the curved sub-system. Between 30 and 50 added atoms the \(sp^3\) region has penetrated through the C\(_{60}\) fullerene. Between 50 and 70 added atoms the \(sp^3\) cluster reaches to the surface of the C\(_{240}\) shell. After this more \(sp^3\) atoms form in the middle of the cluster and between the shells and to some extent on surface of the C\(_{240}\). However, a major fraction of the C\(_{240}\) atoms remain \(sp^2\) bonded. The evolution of the size of the largest \(sp^3\) cluster is presented in Fig. 3. After the addition of 110 atoms the outermost C\(_{240}\) shell starts to break and can no longer maintain the pressure required for the diamond nucleation, resulting in decrease in the \(sp^3\) contents.

Finally, we analyzed the bonding geometries of the central cores of the final snapshot CG relaxed system for the case of 110 atoms added from simulation series A and C, and the CG relaxed systems with 70 and 110 atoms added from series B. Bond angles and distances are characteristic of the diamond structure. For 70 added atoms in simulation series B, the bond lengths are 1.48±0.05 Å. For 110 added atoms, the average bond lengths are 1.49±0.07 Å, 1.52±0.07 Å, and 1.51±0.08 for the simulation series A, B, and C, respectively. The average angles are about 111±12 degrees for the CG relaxed structures of series A and 112±13 degrees for series C but 109±12 degrees for the structures from series B. However, the large deviations from the ideal diamond binding indicate strong disorder which contradicts the experimental findings, where seemingly highly ordered diamond samples have been produced by the irradiation.3 We believe, however, that a transformation of the amorphous structure to a ordered crystal takes place on a longer timescale than can be studied with our present simulation techniques. The flux of incoming particles during the irradiation, which causes the temperature to be high in the core region of the fullerenes should cause the atoms to overcome the energy barriers associated with this transformation, while the confinement by the graphitic shells stabilizes the crystalline. We support the picture provided by Banhart and Ajayan, who propose an interstitial-vacancy self-annealing of the diamond nucleus and observe a very high stability of the emerging diamond crystallites.9

Ordered icosahedral molecules purely made of fourfold-coordinated carbon have been discussed in the past as possible seeds for a diamond deposition, too.10 While free clusters of this type which include only 100 atoms can easily delaminate into concentric fullerenes, such structures appear to become stable against MD relaxation at 2700 K for atom numbers greater than 300.11 This together with our current results seems to indicate that there is a critical size for spherical atomic arrangements below which threefold-coordinated atoms always dominate but above which diamondlike atoms can support themselves to bond into a completely fourfold-coordinated structure. The confinement by multilayer carbon fullerenes appears to lower this limit and to reduce the capabilities of \(sp^3\)-like atoms to relax back into a graphitic form. The nucleation in the carbon fullerene core involves a transition of quasi-two-dimensional curved shell structures into denser three-dimensionally packed diamond-like structures. Our simulation shows that this process requires rapid energy input by incoming particles along with ultrafast heat dissipation to the surroundings. We observe a transformation which is best described by a percolation of initially dispersed \(sp^3\)-bonded regions fusing after a certain threshold (the number of added atoms being between 50 and 100 atoms in our case) mostly as a one \(sp^3\) cluster. Our simulation shows that atomic-scale details of the nucleation processes inside carbon fullerenes can be studied to a good extent by computational modeling using density-functional-based interatomic forces.

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