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A fast method to prepare mechanically strong and water resistant lignocellulosic nanopapers

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Abstract

This study covers a green method to prepare hybrid lignocellulosic nanopapers by combining wood nanofibres (WNFs) and cellulose nanofibres (CNFs). The WNFs and CNFs behave synergistically to compensate for the drawbacks of each other resulting in enhanced hybrid nanopapers. The draining time of hybrid nanopapers was improved by up to 75% over CNF nanopaper, and the mechanical properties (modulus, strength and elongation) were respectively improved up to 35%, 90% and 180% over WNF nanopaper. Additionally, the water resistance of hybrid nanopapers was considerably improved with a water contact angle of 95°; the neat CNF nanopaper had a contact angle of 52°. The morphology of nanopapers, studied by electron microscopy, indicated that lignin acts as a matrix, which binds the nanofibres together and makes them impervious to external environmental factors, such as high humidity. The reported hybrid nanopapers are 100% bio-based, prepared by a simple and environmentally friendly processing route. Reported hybrid nanopapers can be used in novel applications such as gas barrier membranes and printable electronics.

Keywords

Hybrid nanopapers; Wood nanofibres; Cellulose nanofibres; Dewatering; Water resistance; Thermal properties.
1 Introduction

With the advent of nanotechnology, novel and intriguing functional materials have been coming into the spotlight. A thin, freestanding film called as nanopaper is one such material. Nanopapers are formed from highly concentrated, tightly packed nanoparticles, such as carbon nanotubes (CNTs), carbon nanofibres, graphene, clay nanoplatelets or cellulose nanofibres (CNFs) (Huang, 2017). Their unique and distinctive properties make them stand out from other conventional material. Nanopapers prepared from carbon-based nanomaterials have excellent electrical conductivity (Salajkova, Valentini, Zhou, & Berglund, 2013). Nanopapers based on clay nanoplatelets (CNPs) result in excellent barrier properties and thermal stability (Sehaqui, Kochumalayil, Liu, Zimmermann, & Berglund, 2013). CNF nanopapers, on the other hand, are known for their extreme toughness (M. Henriksson, Berglund, Isaksson, Lindström, & Nishino, 2008)(Deng, Huang, Zhou, Chen, & Fu, 2016). Additionally, CNF nanopapers have a low thermal expansion coefficient (similar to glass) and a higher dielectric constant and thermal durability than most plastics; yet, it is foldable like paper (Yagyu, Ifuku, & Nogi, 2017). Such a variety of functional properties make them suitable for multifunctional and high-end applications such as electronic...
displays (Sehaqui, Zimmermann, & Tingaut, 2014), flexible electronics (Koga et al., 2014), organic
solar cells (Yagyu et al., 2017) and lithium ion batteries (Chun, Lee, Doh, Lee, & Kim, 2011).

CNF nanopaper has been a centre of attention for its simple preparation and lightweight structure
and due to the renewable nature of the raw material. CNFs are usually prepared by disintegrating
bleached pulp in stone grinders or microfluidisers. They usually have a thickness of a few
nanometers and a length of a few micrometres (Kargarzadeh, Ahmad, Thomas, & Dufresne, 2017).

CNF nanopapers are prepared from CNF suspensions by using vacuum filtration for water removal
through vacuum filtration, followed by drying the wet CNF mat. They are transparent (partially or
fully) with ultrastrong mechanical characteristics. They can have an elastic modulus of tens of GPa
and tensile strength of few hundred MPa (M. Henriksson et al., 2008; Sehaqui et al., 2012). Such a
high degree of mechanical stiffness combined with the ease of preparation is unheard of in the
case of an artificial polymeric material, and a green renewable material providing such properties is
an unprecedented advantage. However, despite their numerous advantages, CNF nanopapers are
not problem-free. The primary challenge is their affinity for water, which affects both the
preparation and long-term use of the nanopapers. Usually, cellulose nanofibres have the allomorph
of cellulose which mostly contains (110) surfaces that are highly decorated with protruding hydroxyl
groups (Mazeau & Rivet, 2008). This abundance of hydroxyl groups makes them hydrophilic and,
as a result, CNFs retain a lot of water, which is hard to remove during filtration. The dewatering
time usually lasts up to a few hours (Iwamoto, Nakagaito, Yano, & Nogi, 2005), which makes them
difficult to commercialise. Advanced methods that drain water in 0.5–1 hours have been reported
but they are few in number (Österberg et al., 2013; Sehaqui, Liu, Zhou, & Berglund, 2010). In
addition, water causes problems with the functionality of CNF nanopapers as it weakens the
finished product. A loss of 95% in the modulus was reported when the nanopaper was in contact
with water (Sehaqui et al., 2014). Lucenius et al. also reported a loss of 80–90% in the mechanical
properties of wet CNF nanopapers (Lucenius, Parikka, & Österberg, 2014). Even in the presence
of high relative humidity (or RH; specifically, above 70% RH), CNF nanopapers’ properties
deteriorate rapidly (Benitez, Torres-Rendon, Poutanen, & Walther, 2013; Sethi et al., 2018).
Although some post-chemical modification has been proposed to improve the tolerance of nanopapers towards the water and humidity (Lucenius et al., 2014; Sehaqui et al., 2014), simplified and eco-friendly methods are desired to deal with the dewatering time and water and humidity tolerance of CNF nanopapers before they can be mass-produced.

CNFs are extracted from lignocellulosic biomass in two principal steps. Firstly, a harsh chemical pre-treatment is done to remove the lignin and hemicellulose. The origin of this process comes from the paper industry, which requires bleaching (cleaning) of the pulp before papermaking. Secondly, the treated (bleached) pulp is then severely ground to separate individual CNFs from the pulp. In recent years, an argument has been made that it is not judicious to bleach the pulp before nanofibre preparation as it removes lignin. Lignin is the second most abundant polymer in nature, after cellulose, and comprises up to 40% of plant biomass (G. Henriksson, 2017). It is necessary for proper load distribution and for maintaining the integrity of the wood. Additionally, lignin provides water resistance to the wood. Finally, the removal of lignin is an expensive process as it is usually done by kraft pulping, which utilises expensive and hazardous chemicals. Therefore, it is of interest to research lignin-rich wood nanofibres (WNFs) as a potential raw material for making high-end products, such as nanopaper. It is a rather new area of research and only a handful of studies have been reported on the topic (Herrera et al., 2018; Nair & Yan, 2015; Rojo et al., 2015; Visanko et al., 2017). Nanopapers made from WNFs offer certain advantages such as quicker preparation time and improved water resistance. However, WNF nanopapers have lower mechanical properties (25% less modulus, 45% less yield strength) and are brittle (20% less elongation) when compared to CNF nanopaper (Visanko et al., 2017).

Nanopapers prepared from either WNF or CNF have distinct advantages, yet they suffer from inherent drawbacks, which need to be addressed. Interestingly, the advantages and drawbacks seem to be mutually exclusive. CNF nanopapers are tough but it takes hours to prepare them and they are extremely weak in the presence of moisture. WNF nanopapers are prepared quickly and they have high water resistance, but they are brittle. Therefore, it makes sense to prepare a hybrid nanopaper by combining WNF and CNF to achieve a synergic relationship. Combining two
different kinds of nanostructured particles to make a hybrid multiphase material is a well-known, fruitful way of improving the properties of nanopapers, where each ingredient complements the other. In CNT-CNf hybrid nanopaper, CNTs imparted the electrical conductivity, and CNFs provided a water-based processing route, along with a cheap, high strength matrix (Salajkova et al., 2013). Similar synergistic behaviour was observed between CNP-CNf nanopapers. CNPs imparted fire retardancy and barrier properties, and CNFs contributed to high elongation, which was missing in nanopaper made exclusively from CNPs (Liu, Walther, Ikkala, Belova, & Berglund, 2011). The current work is inspired by a similar principle. We hypothesised that an intricate WNF-CNf interpenetrating network in a nanopaper would allow each material to make up for other’s deficiencies. WNFs would benefit from the integral CNF network, which has extraordinary dry mechanical properties such as high modulus and elongation. In return, the CNF network would benefit from the binding properties of the WNF network, water resistance, and quicker draining time. We report a novel method to prepare lignin-rich hybrid nanopaper by using a mixture of CNF and lignin WNF suspensions in water.

This study provides a simplistic and industrially adaptable method of preparing lignin rich hybrid nanopaper and discusses the advantages of lignin-rich hybrid nanopapers. The primary aim of this study was to create and study a hybrid nanopaper prepared from WNFs and CNFs. WNFs with 27% lignin content were prepared through high temperature grinding and combined with CNFs in ratios of 9:1, 3:7, 5:5 and 7:3. The neat WNF and neat CNF films were also prepared as references. The morphology was studied with the help of scanning electron microscopy. Mechanical properties were characterised by the tensile testing (dry and wet). The effect of humidity on mechanical properties was studied by dynamic mechanical analysis. Thermal gravimetric analysis was used to study thermal properties, and contact angle measurements were used for evaluating the water repellency of the nanopapers. This research, for the first time, reports a hybrid nanopaper prepared from WNFs and CNFs, which is 100% bio based. Additionally, this is a first attempt that mentions the drawbacks of WNF nanopaper and improves it with an eco-friendly method.
2 Materials and methods

2.1 Preparation of nanofibres

WNFs were prepared from unbleached spruce groundwood pulp (never dried), which was provided by Stora Enso (Veitsiluoto, Finland). To prepare WNFs, the high temperature thermomechanical grinding of wood pulp was done in Masuko super mass collider (MKCA6-2 J CE; Masuko Sangyo, Japan) according to the method reported by (Visanko et al., 2017). Briefly, 1.5 wt.-% of wood pulp (4kg total weight) was heated until the temperature reached 95°C. The heated pulp was repeatedly fed in grinding disks (10-inch diameter). The outcome from the grinder was collected, heated back to 95°C and fed back into the grinder. The distance between the grinding disks was gradually decreased. The pulp was ground once through 0 µm and once through -50 µm, and fifteen times through -100 µm disk clearance. On the completion of grinding, the WNF suspension was collected and stored at a temperature of 4°C. For more details about WNFs and its characteristics, the readers are advised to follow the earlier publication (Visanko et al., 2017).

Cellulose nanofibres were prepared from softwood sulphite pulp, also provided by Stora Enso (Veitsiluoto, Finland). For grinding, they were also repeatedly fed in the Masuko super mass Collider (MKCA6-2 J CE; Masuko Sangyo, Japan). The pulp was passed three times from -20 µm, four times from -40 µm, five times from -60 µm, and seven times from -90 µm clearances. On the completion of grinding, the WNF suspension was collected and stored at a temperature of 4°C.

2.2 Preparation of nanopapers

To prepare the hybrid nanopapers, the WNF and CNF suspensions were added to a glass beaker and diluted to a concentration of 0.2 wt.-%, according to the formulations presented in Table 1. The WNF-CNF suspension was mixed using an Ultra-Turrax homogeniser, at 10000 rpm for 5 min, and degassed under a vacuum of -70 kPa for 15 min to remove air bubbles. Finally, the water was drained by vacuum filtering the WNF-CNF suspension through a Durapore 0.65 µm polyvinylidene fluoride membrane (M. Henriksson et al., 2008; Sethi et al., 2018). The vacuum applied was -70 ±
5 kPa. The draining was assumed complete when the time difference between the fall of two consecutive water drops from the funnel was 30 seconds. After the draining was complete, the WNF-CNF wet mat was carefully peeled from the membrane and stacked between two steel meshes (70 µm pore size) and blotting paperboards. The whole assembly was compression moulded between hotplates at 150°C for 3 min at 2 MPa and 15 min at 50 MPa. The samples are coded according to a ratio of WNF to CNF. For example, 70WNF30CNF represents the sample that has 70 wt.-% WNF and 30 wt.-% CNF. 100WNF and 100CNF represents the neat WNF and CNF nanopapers respectively. The grammage of nanopapers was 100-120 grams per square meter.

The cross-section and surface morphology of reference and hybrid nanopapers was studied by field emission scanning electron microscopy (FE-SEM) using a Zeiss Ultra Plus electron microscope (Oberkochen, Germany). Before imaging, the samples were coated with a thin layer of platinum. The samples were scanned with an electron beam accelerated at a voltage of 5 kV, and an in-lens detector was used to collect the signals.

<table>
<thead>
<tr>
<th>Samples</th>
<th>WNF suspensions</th>
<th>CNF suspension</th>
<th>Deionized water**</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>WNF</td>
<td>CNF</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Water*</td>
<td>Water*</td>
<td></td>
</tr>
<tr>
<td></td>
<td>0.20</td>
<td>0.02</td>
<td>92.0</td>
</tr>
<tr>
<td></td>
<td>0.18</td>
<td>0.06</td>
<td>91.8</td>
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<tr>
<td></td>
<td>0.14</td>
<td>0.06</td>
<td>91.2</td>
</tr>
<tr>
<td></td>
<td>0.10</td>
<td>0.1</td>
<td>90.7</td>
</tr>
<tr>
<td></td>
<td>0.06</td>
<td>0.14</td>
<td>90.2</td>
</tr>
<tr>
<td></td>
<td>-</td>
<td>0.2</td>
<td>89.4</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>7.8</td>
<td>1.0</td>
<td>100</td>
</tr>
<tr>
<td></td>
<td>7.0</td>
<td>3.1</td>
<td>100</td>
</tr>
<tr>
<td></td>
<td>5.4</td>
<td>5.2</td>
<td>100</td>
</tr>
<tr>
<td></td>
<td>3.9</td>
<td>7.2</td>
<td>100</td>
</tr>
<tr>
<td></td>
<td>2.3</td>
<td>10.4</td>
<td>100</td>
</tr>
<tr>
<td></td>
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<td></td>
</tr>
<tr>
<td></td>
<td>Total</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>100WNF</td>
<td>90WNF</td>
<td>70 WNF</td>
</tr>
<tr>
<td></td>
<td>10CNF</td>
<td>30CNF</td>
<td>50CNF</td>
</tr>
<tr>
<td></td>
<td>50CNF</td>
<td>30WNF</td>
<td>100CNF</td>
</tr>
</tbody>
</table>

* Water as a part of never-dried pulp
** Deionised water added to dilute the CNF suspension

2.3 Characterisation of nanopapers

Water contact angles were determined using the sessile drop method. A 6.5 µL Milli-Q water droplet was placed on the surface of the samples, and 60 images of the droplet at the rate of 1 image per second were collected using a CAM 200 contact angle meter (KSC instruments,
Finland). The static contact angle of each image was determined by a Young-Laplace equation estimation of the drop shape using the software provided by the manufacturer. An average of all 60 values was calculated and assumed as the contact angle of the surface. Three measurements at different positions on each sample were conducted, and the average value along with the standard deviation is reported.

To measure water absorption, the reference and hybrid nanopapers were dipped in distilled water for a prolonged period. The wet sample was periodically removed from the water after 10 min, 1 hour, and 24 hours and then weighed on an analytical balance after blotting the excess water with a dry tissue paper. The results are expressed by an increase in weight as a percentage gain.

The wet and dry mechanical properties were evaluated with tensile testing using Zwick Roell universal test machine (Ulm, Germany). The samples were cut with scissors into 50 mm x 5 mm rectangles, and stored in ambient conditions (23°C and 55% RH) for at least 48 hours before testing. Stress-strain curves were recorded using a load cell 1 kN at the crosshead speed of 5 mm/min. The initial separation between the grips was 20 mm. The elastic modulus was determined by calculating the slope of the stress-strain curve in the linear region and the yield strength was determined as the point of intersection of the stress-strain curve and offset line originating from a 0.2% co-ordinate. The results are reported as an average of at least 5 strips per sample. For wet tensile tests, the method reported by Sehaqui et al was used (Sehaqui et al., 2014). In short, a 50 μl drop was gently placed in the middle of strips. The drop was gently wiped after 60 seconds and the stress strain curves were recorded. The results are reported for an average of at least 3 samples.

The effect of humidity on the nanopapers’ mechanical properties was determined using a dynamic mechanical analysis (DMA) Q800 equipped with a RH accessory (TA Instruments, New Castle, USA) in strain mode (amplitude 10 μm and frequency 1 Hz). The samples were first equilibrated at 0% RH for 1 hour to remove the absorbed moisture. Thereafter, RH was increased from 0% to 95% at a rate of 1%/minute. Finally, the RH was maintained at 95% for 1 hour to determine the
equilibrium value of the storage modulus. The temperature was maintained at 30°C for the entire duration of the test.

The thermal stability of the nanopapers was determined by a Netsch STA 409 PC (Germany). The samples were heated from room temperature to 800°C (heating rate 10°C/min) under a nitrogen atmosphere (flow rate: 60 ml/min), and the weight loss was recorded.

The crystalline structure of the sample was determined by Wide-angle X-ray diffraction (XRD). The measurements were recorded using a Rigaku SmartLab 9 kW rotating anode diffractometer (Japan) using a Co Kα radiation (40 kV, 135 mA; k = 1.79030 Å. Bragg’s angle (2θ) was varied from 10° to 50°, with a step width of 0.02° (scanning speed = 2° min⁻¹). The degree of crystallinity (CrI) was calculated from the peak intensity of the main crystalline plane (200) diffraction (I₂₀₀) which was at 26.2° and from peak intensity at 22° C, which is associated with amorphous fraction of cellulose (Iₐₐₘ) (French, 2014), according to Equation 1:

\[ CrI = \left( \frac{I_{200} - I_{am}}{I_{200}} \right) \quad \text{Equation 1} \]

3 Results and discussions

3.1 Draining time and ease of nanopaper preparation

The 100WNF nanopaper was the fastest to drain (in 15 min). On the other hand, 100CNF took almost 120 min to drain. The draining time is one of the most significant challenges hindering the use of CNFs in commercial applications. Clearly, if an intermediate process takes 120 min, nanopapers are unlikely to be produced on an industrial scale. It was found that by replacing a part of the hydrophilic CNFs with hydrophobic WNFs, the draining time can be considerably decreased. The hybrid nanopapers drained in a time that was in-between the draining time of 100WNF and 100CNF, closer to 100WNF. The draining times for nanopapers are presented in Figure 1. To summarise, 70WNF30CNF took 35 min, 50WNF50CNF took 55 min, and 30WNF70CNF took 65 min. Interestingly, the partial replacement of CNF with WNF improved the draining time in contrast to the neat CNF nanopaper. For example, 30WNF70CNF should have theoretically drained in 90
min (determined by rule of mixtures). Experimentally, it drained in 65 min, which is an improvement of 28%. In the case of others (70WNF30CNF and 50WNF50CNF), the improvement was around 20–25%. The reason behind this needs to be studied further.

![Draining time of references and hybrid nanopapers; 100WNF drained in 12% of time taken by 100CNF.](image)

However, 100WNF, after draining swiftly, performed poorly in preparation ease. Wet 100WNF film disintegrated easily while peeling from the polyvinylidene fluoride membrane after draining, making the job tedious. The photographic image of 100WNF is shown in Figure S1 Supplementary file, which have cracks on the edges due to disintegration while peeling. The reason behind this is the lack of cohesion due to the deterioration of hydrogen bonding among the wet WNFs. It has been reported that, when wet, hydrogen bonding plays a role in strength (Tenhunen et al., 2016). WNFs are covered by lignin, which can form hydrogen bonds (Kubo & Kadla, 2005), but not as extensively as CNFs which have a large number of hydroxyl groups. The addition of CNFs to the hybrid nanopapers strengthened the wet film, and it maintained its structure while peeling. The photographic image of a hybrid nanopaper (50WNF50CNF) is provided in Figure S1 of the supplementary file, which does not show any cracking. This difference is an added advantage of hybridization.
3.2 Morphology of nanopapers

When observed under an electron microscope, 100WNF has a levelled surface (Figure 2a). No fibrillar entities are visible showing that lignin has melted and fused to form a uniform film under the influence of high pressure and temperature (Wang, Hu, & Zeng, 2012). On the other hand, the surface of 100CNF has an intertangled web like structure, where fibres are visible as individual entities (Figure 2c), which is common for nanopaper surfaces. Finally, 50WNF50CNF has characteristics of both; nanofibres are visible yet embedded in the lignin matrix (Figure 2b). The structure can be assumed to be an anisotropic composite where CNFs are randomly distributed within the lignin matrix. Additionally, it is apparent that the dispersion of CNFs within WNFs is uniform, validating the preparation method. The presence of lignin in hybrid nanopapers are is confirmed by FTIR, which are presented in Figure S2. A peak at 1510 cm⁻¹, which represents the aromatic rings of lignin, can be clearly observed in 100WNF and 50WNF50CNF. On the other hand, no such peak is present in 100CNF nanopaper.

Figure 2. Surface micrographs from FESEM imaging of surface of 100WNF, 50WNF50CNF, and 100CNF. No fibres are visible in 100WNF indicating the fusion of lignin. 50WNF50CNF have fibres partially embedded in lignin, and 100CNF fibres are clearly visible on the surface.

The effect of lignin was more pronounced in the images of fractured cross-sections (Figure 3). As shown in Figure 3b and 3c, 100CNF and 50WNF50CNF have a layered structure, which is a known characteristic of cellulosic nanopaper (M. Henriksson et al., 2008). On the other hand, in 100WNF, the layers are not plainly noticeable, and the entire structure appears to be tightly glued (Figure 3a), resulting in the brittle nature of 100WNF (Figure 4). Additionally, no visible nanofibres that would be protruding from the fractured surface were seen in 100WNF (Figure 3a magnified). However, in 100CNF, individual CNFs can be clearly seen, indicating sliding and separation across
the layer during the fracture (Figure 3c magnified). This sliding is responsible for the ductile fracture and toughness (M. Henriksson et al., 2008). The slippage of individual fibres can also be observed in 50WNF50CNF (Figure 3b), which contributed to its ductility (Figure 4). Additionally, in 50WNF50CNF, the CNFs seem to be embedded well in the lignin matrix and appear to be uniformly distributed. The entire composite-like morphology of 50WNF50CNF can be imagined as an interpenetrating network of CNFs and nanocellulosic entities from WNFs, with lignin acting as a matrix. The benefits of this morphology are discussed in more detail in the mechanical properties section.

The (Crl) of the nanopapers decreased when the WNFs were added. The XRD diffractograms used to calculate the Crl are presented in Figure S3a. 100CNF has a Crl of 71%; on the other hand, 100WNF has Crl of 56%. The hybrid nanopapers have a Crl value between of 100CNF and 100WNF, as expected (Figure S3b). The presence of amorphous lignin and hemicelluloses in 100WNF is the reason for the observed low Crl of WNF. Also, it has been reported that delignified CNFs have higher crystallinity (Kaushik & Singh, 2011).
3.3 Mechanical properties of nanopapers

Adding the CNFs to the WNFs gave surprisingly good results. Testing found that 100WNF had the lowest mechanical properties (Figure 4); it was highly brittle with only 2.6% of elongation, a modulus of 4.3 GPa, and a tensile strength of 68 MPa. Hybrid nanopapers have higher modulus, yield strength, tensile strength and elongation, compared to to 100WNF (Figure 4). The quantitative results are presented in Table 2. Even at 30 wt.-% of added CNFs, the modulus of hybrid nanopaper increased from 4.3 GPa to 5.4 GPa, an improvement of approximately 25%. The value of 5.4 GPa is even higher than the theoretical value determined by the rule of mixtures (4.8 GPa). Additionally, both the elongation and tensile strength improved by 80% and 50%, respectively, and an improvement of 33% in yield strength was also observed.
The lower mechanical properties of 100WNF are due to the high concentration of lignin (27 wt.-%).

It has been reported that lignin has a modulus of 3 GPa and strength of 50 MPa (Gibson, 2012), which is significantly less than the properties of neat CNF networks (Figure 4). Another reason for the reduced mechanical properties of 100WNF is the decrease in the extent of hydrogen bonding. It has been reported that lignin interferes with hydrogen bonding (Horseman, Tajvidi, Diop, & Gardner, 2017; Rojo et al., 2015). WNFs are covered with lignin (Figure 2); hence, cellulose entities are not free to form hydrogen bonds among each other. Hydrogen bonds are a strong bond and the primary reason for the excellent dry mechanical properties of cellulosic nanopapers.

Additionally, hydrogen bonding also determines the yield strength (Benítez et al., 2013). Due to lack of hydrogen bonding and tightly glueing of nanofibres by lignin, 100WNF yielded to 20 MPa before 100CNF. Finally, after yielding, 100WNF fractured abruptly, but 100CNF (and hybrid nanopapers) exhibited a secondary inelastic elongation phase, where the deformation was extended without fracturing due to interfibrillar sliding.

The results from hybrid nanopapers show a remarkable reinforcing ability of CNFs when added to a WNF network. Another reason for such an improvement in mechanical properties is the formation of a hydrogen-bonded network among CNFs. Apparently, a critical concentration is needed for CNFs to percolate as no improvement was observed in hybrid nanopaper when the concentration of CNF was 10 wt.-%; however, at a concentration of 30 wt.-% and higher, the mechanical properties showed improvement. Perhaps, at 10 wt.-% CNFs were not able to make a network, but at 30 wt.-% the concentration was sufficient to make a stiff hydrogen-bonded network among WNFs. The FESEM images also confirm this premise. The hybrid nanopaper has separated individualised CNFs protruding out of the plane, interconnected to each other and forming a network (Figure 3b). This hydrogen-bonded network is primarily responsible for the excellent mechanical properties of nanopaper (Benítez et al., 2013) and interfibrillar sliding between CNFs is responsible for the higher elongation (Benítez et al., 2013) Finally, an interesting observation can be made from the stress-strain curves (Figure 4). The modulus and yield strength of hybrid
nanopapers plateaued when the CNF concentration was above 30 wt.-%. In contrast, the elongation and tensile strength increases with the increase in CNF concentration.

It can be concluded that entirely eliminating hydrogen bonding is not a wise decision and the results can be seen in the stress-strain curves of nanopapers. Peresin et al. also concluded that it is judicious to hydrophobise the surface of nanopaper, as modifying individual nanofibres interferes with the hydrogen bonding (Peresin et al., 2017). This step is critical to imparting the mechanical properties of nanopaper. The hydroxyl groups of cellulose, which provide excellent mechanical properties by bonding to each other, are also responsible for their poor mechanical properties in the presence of water (Benítez et al., 2013). Therefore, a balance is highly desirable where hydrogen bonding can be controlled.

Figure 4. Stress-strain curves of dry nanopapers; 100WNF has the poorest properties, which improved considerably when CNFs were added. The elastic modulus, elongation, tensile and yield strength were also improved.

<table>
<thead>
<tr>
<th>Sample</th>
<th>Elastic modulus (GPa)</th>
<th>Tensile Strength (MPa)</th>
<th>Elongation (%)</th>
<th>Density g/cm³</th>
</tr>
</thead>
<tbody>
<tr>
<td>100WNF</td>
<td>4.3 ± 0.1</td>
<td>68 ± 2</td>
<td>2.6 ± 0.1</td>
<td>1.03</td>
</tr>
<tr>
<td>90WNF10CNF</td>
<td>4.3 ± 0.1</td>
<td>75 ± 7</td>
<td>2.1 ± 0.3</td>
<td>1.04</td>
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<tr>
<td>70WNF30CNF</td>
<td>5.4 ± 0.2</td>
<td>100 ± 3</td>
<td>4.75 ± 0.3</td>
<td>1.10</td>
</tr>
<tr>
<td>50WNF50CNF</td>
<td>5.5 ± 0.2</td>
<td>107 ± 5</td>
<td>5.3 ± 0.6</td>
<td>1.13</td>
</tr>
<tr>
<td>30WNF70CNF</td>
<td>5.8 ± 0.1</td>
<td>131 ± 8</td>
<td>7.4 ± 1</td>
<td>1.18</td>
</tr>
<tr>
<td>100CNF</td>
<td>6 ± 0.1</td>
<td>138 ± 4</td>
<td>6.9 ± 0.6</td>
<td>1.22</td>
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</tbody>
</table>
The results of tensile testing are quite exciting, and better than expected. With just 30 wt.-% CNFs, the modulus and yield strength of nanopaper is at par with 100CNF. This result makes a compelling argument in support of the hybridisation of CNFs and WNFs, as using relatively fewer CNFs results in a disproportionate contribution to the enhancement of properties. After considering the draining time, 70WNF30CNF appeared to have good optimisation, comparable to the mechanical properties of 100CNF, at a draining time of only 35 min, in contrast with 100CNF, which took 120 min. This ratio maintains a balance between the draining time and mechanical properties of nanopapers, making it suitable for cost-effective production.

3.4 Contact angle and water absorption

Due to the presence of lignin, the hybrid nanopapers were more water repellent than 100CNF. The contact angle values, along with photographic images, are presented in Figure 5. Due to the high hydrophilicity of CNFs, 100CNF had a contact angle of 52° with water. On the other hand, 100WNF had an angle of 107°, indicating it is highly hydrophobic. Between them, 70WNF30CNF had a contact angle of 95°, making it hydrophobic, as a surface demonstrating a contact angle of more than 90° is considered hydrophobic. Clearly, lignin is effectively altering the surface, which can also be confirmed by FESEM images (Figure 2). The lignin has melted and fused, and, as a result, the hydrophilic CNFs are not in direct contact with the water droplet, causing the surface to become water repellent. It is worth mentioning that the contact angle of 30WNF70CNF remains 25° higher than 100CNF, indicating that, even in a small quantity (8 wt.-% in this case), the lignin, if uniformly distributed, can make a surface water repellent.

Making cellulosic materials hydrophobic is one of the biggest challenges for researchers. The most common technique is a surface modification using polymeric grafting, which is tedious and requires significant capital investment in reagents, solvents and chemical equipment. Paquet et al. grafted a high molecular weight polycaprolactone (M_w=42,500) on CNF surface, using isocyanate-based grafting in methylene chloride and anhydrous toluene as a reaction medium to obtain a contact
angle of 95° (Paquet, Krouit, Bras, Thielemans, & Belgacem, 2010). Other methods for improvement of water repellency have also been reported. A contact angle of 70° was achieved by silylating the nanopaper surface (Peresin et al., 2017), whereas contact angles as high as 140° have been reported using the layer by layer deposition of carnauba wax (Forsman et al., 2017). Our method gave a contact angle of 95° with a simplistic eco-friendly approach, using cheaper raw material and processing techniques known to the paper and pulp industry. This insight gives a rare, lucrative advantage to the reported material.

Figure 5. Water contact angle of reference and hybrid nanopapers; hybrid nanopapers are significantly more water repellent than 100CNF, with 70WNF30CNF demonstrating hydrophobicity.

Hybrid nanopapers not only had a higher contact angle, but they were also relatively impervious to water (Figure S4a); 100WNF absorbed the least amount of water. After soaking for 24 hours, 100CNF absorbed 110% of water; on the other hand, 100WNF absorbed 69% of water. The magnitude of water absorbed by hybrid nanopapers was intermediate to that of 100CNF and 100WNF. It can be concluded that the presence of lignin makes nanopaper resistant to water absorption. Interestingly, there appears to be a linear relationship between the amount of lignin and water absorption (Figure S4b). The higher the amount of lignin in nanopaper, the less water it absorbed when dipped in water. It has been reported that lignin fills the voids and crosslinks with the cellulose, which causes a decrease in water absorption (Nair & Yan, 2015).
3.5 Wet mechanical properties

When wet, 100CNF weakened the most and retained only 15% of its dry modulus (1 GPa when wet, from 6 GPa when dry; see Figure 6b). Additionally, it exhibited no yield point (Figure 6a), which indicates that it yielded right away to the inelastic region and permanently deformed as soon as a load was applied, demonstrating that the presence of water causes significant structural changes in nanopaper. Firstly, water seeps into the interfibrillar regions, diminishing the hydrogen bonding (Benítez et al., 2013), which is the primary factor responsible for stiffness. Once the hydrogen bonding is gone, the nanopaper permanently deforms under a considerably lower load than it would when dry (as seen in Figure 6). Secondly, water molecules plasticise the hemicelluloses, which are present in the non-crystalline part of CNFs (Cunha, Zhou, Larsson, & Berglund, 2014), and, hence, the modulus is decreased.

Lignin-rich nanopapers, on the other hand, were able to resist the water. As measured, 100WNF retained 75% of its modulus (3.2 GPa when wet from 4.3 GPa when dry), which was 220% more than wet 100CNF. Additionally, 100WNF showed elastic behaviour in the elastic region and yield, which are marks of a stiff material. Hybrid nanopaper also benefitted from the presence of lignin and showed improved mechanical properties when wet (Figure 6a and b). While wet, 70WNF30CNF retained 40% of its modulus (100CNF retained 15%) and showed elastic behaviour in the beginning. The presence of lignin restricts interfibrillar sliding, as lignin has melted and fused with fibres (Figure 2), which maintains the integrity of nanopaper even when damp. These results are considerably better than other reported studies (Lucenius et al., 2014; Sehaqui et al., 2014).

Another significant finding is that the mechanical properties of 100WNF are still in the range representing engineering polymers (modulus 3.5 GPa and strength 38 MPa). Water is an abundant material that frequently interacts with everyday materials, and a product such as nanopaper is expected to display some considerable amount of water resistance. Doing so is the only way to safeguard the commercialisation of nanopaper.
Figure 6. (a) Stress-strain curves after wetting the reference and hybrid nanopapers; 100CNF yielded right away and lost its stiffness, while 100WNF retained its stiffness. Hybrid nanopapers have better wet properties than 100CNFs. (b) Comparison of dry and wet modulus of reference and hybrid nanopapers; 100CNF lost 85% of its modulus, while 100WNF lost only 25%. Hybrid nanopapers displayed intermediate modulus to 100CNF and 100WNF.

3.6 Effect of Humidity

The moisture absorbed from humid conditions is equally damaging to the mechanical properties of 100CNF as soaking it in water; 100CNF lost 87% of its stiffness (storage modulus) in the presence of 95% RH, which is the humidity in the atmosphere on a rainy day. In contrast, the hybrid nanopapers had a higher tolerance towards humidity than both 100CNF and 100WNF (Figure 7). In the first stage of the experiment (when the humidity was steadily increased from 0% to 95% RH), all the nanopapers showed a similar response, which included an initial stiff phase where the modulus showed no sudden fluctuation. After the humidity reached around 60–70% RH, the modulus started dropping steadily. When the humidity reached 95%, 100CNF still had a higher modulus than 100WNF and hybrid nanopapers. It was only in the second stage, when the humidity was kept constant at 95% RH, that 100CNF lost its storage modulus in two sudden linear drops; no such behaviour was observed with 100WNF and hybrid nanopapers. In the end, the storage modulus of 100CNF reached a plateau value corresponding to a loss of 87% in its stiffness, which was, unexpectedly, the same drop as observed in the elastic modulus of wet tensile testing results (Figure 6b). On the other hand, 100WNF and hybrid nanopapers did not show any sudden drops and quickly reached a plateau value, which did not decrease any further, no matter how long the sample was exposed to the humidity. This result indicates that moisture weakens the materials a
bit, but the nanopapers were able to maintain their integrity. Finally, nanopapers containing WNFs lost around 60–65% of their properties (100CNF lost 87%). The advantage of hybridisation can be explained briefly by the results from 50WNF50CNF. After equilibrium, it had 2.3 times the modulus of 100CNF and 1.5 times the modulus of 100WNF. This difference shows the promising value of such materials, which have a high quantity of moisture-susceptible material, yet the hybrid nanopapers showed as much resistance as 100WNF. Apart from the magnitude of the storage modulus, it was found that incorporating lignin fibres delayed the onset of losing storage modulus. The onset point for 100WNF was 77% RH, while for 100CNF it was 70% RH, indicating the stable nature of 100WNF, which was transferred to hybrid nanopapers. The onset point for 70WNF30CNF was 75% RH and for 50WNF50CNF was 73%.

Figure 7 Variations in the storage modulus of nanopapers with the variations in relative humidity. Nanopapers with WNF resisted the humid conditions, while 100CNF lost most of its stiffness. The deterioration of storage modulus in the presence of moisture can be explained by the disruption of hydrogen bonding (Sethi et al., 2018) and plasticisation of amorphous regions (Benítez et al., 2013). It has been reported that nanopapers can absorb up to 30% water at 95%RH (Benítez et al., 2013). Such a large amount of water leads to dimensional changes and
swelling, and, as hydrogen bonds are weakened, the nanopapers lose their modulus. The presence of lignin makes the nanopaper impervious to the environment; thus, they resist the impact of humid environments. Moisture resistance is one of the mandatory properties for the success of commercial materials. If nanocellulosic materials are to be commercialised in the near future, this issue needs to be adequately addressed. Even in the presence of high humidity, the CNF nanopapers lose significant stiffness. Hybridisation with wood fibres can be one of the potential ways to improve it.

3.7 Thermal stability

The chemical composition of 100WNF caused it to have the poorest thermal stability out of all the nanopapers (until 350°C). WNF consists of many materials, such as lignin hemicellulose, which are extractives that are known to intensify degradation (Poletto, Zattera, Forte, & Santana, 2012). On the other hand, 100CNF has better thermal stability than 100WNF because it has higher crystallinity (Kim, Eom, & Wada, 2010). Interestingly, the hybrid nanopaper (50WNF50CNF) had a thermal stability superior to both 100CNF and 100WNF (Figure 8). By the time the temperature reached 340°C, 50WNF50CNF had degraded 29%, while 100WNF and 100CNF degraded 31% and 37% of their weight, respectively. Additionally, it is worth mentioning that the thermal stability of 100WNF was inferior only until 350°C. At temperatures higher than that, 100WNF became more stable than 50WNF50CNF (100CNF remained thermally inferior) (Figure S5). The reason is the slower degradation rate of lignin and stable by-products, which were also indicated by the amount of char left in the end. According to measurements, 100WNF left 24% char, 50WNF50CNF, left 22% char and 100CNF left 19% char.

The improved thermal stability of hybrid nanopapers should be credited to the interpenetrating morphology, which is known to impart thermal stability to materials (Sethi, Illikainen, Sain, & Oksman, 2017). Combining materials creates more interfaces, which delays the effective heat transfer and, in turn, delaying the degradation. It has been reported that increasing the interactions improves the thermal stability (Fu et al., 2008). Hybridisation conclusively benefits the thermal stability of nanopapers, which is necessary for high-temperature processing and long-term use in
flexible circuits, which can overheat due to electricity. Increased thermal stability is one advantage of the reported procedure, as it uses non-functionaliased fibres, due to which it is able to better maintain (and even enhance) thermal properties. It has been reported that chemical functionalisation leads to a decrease in thermal stability (Fukuzumi, Saito, Iwata, Kumamoto, & Isogai, 2009)(Eyholzer et al., 2010), and it uses expensive reagents and equipment, making it fiscally burdensome.

Figure 8 Degradation and derivative thermogram of 100WNF, 50WNF50CNF and 100CNF. 50WNF50CNF is most thermally stable.

4 Conclusions

In this study, we argue that the presence of lignin in nanopapers can help improve their properties and suggest a simple method to prepare lignin-rich hybrid nanopapers from WNF and CNF. It was found that in tandem, WNF and CNF behave synergistically to compensate for the drawbacks of each other. The synergistic effect was strong and many properties were better than what was expected by rule of mixture. The 70WNF30CNF hybrid nanopapers drained in 35 minutes, which was 85 minutes faster than 100CNF. Additionally, the hybrid nanopapers displayed significant improvement over the mechanical properties of 100WNF. Furthermore, due to the presence of lignin, the hybrid nanopaper showed improved mechanical properties under the influence water and humidity (in comparison to 100CNF). The reason for this improvement is that lignin melts and fuse with the nanocellulosic entities to bind them together. We present a method to obtain a high-end, lignin-rich product without arduous procedures. It is a laborious task to separate lignin for it to
be used in commercial products. (Yang, Zhong, Yuan, Peng, & Sun, 2013). Further work on using lignin-rich hybrid nanopaper for applications such as gas barrier membranes and other applications is in the planning stage.

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6 References


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