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Jukkola, Annamari; Hokkanen, Sanna; Kämäräinen, Tero; Partanen, Riitta; Heino, Antti; Rojas, Orlando J.

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1	Changes in milk fat globules and membrane lipids under the shear fields of microfiltration
2	and centrifugation
3	Annamari Jukkola ^{a*} , Sanna Hokkanen ^a , Tero Kämäräinen ^a , Riitta Partanen ^b , Antti Heino ^{b,}
4	Orlando J. Rojas ^{a*}
5	
6	^a Bio-Based Colloids and Materials (BiCMat), Department of Bioproducts and Biosystems,
7	School of Chemical Engineering, Aalto University, P.O. Box 16300, 00076 Espoo, Finland.
8	^b Valio Ltd., R&D Center, PO Box 30, 00039 Valio, Finland.
9	
10	
11	
12	*Corresponding author: annamari.jukkola@aalto.fi and orlando.rojas@aalto.fi

13 Abstract

14 This study compared the efficiency of centrifugal and microfiltration separation of milk fat 15 globules (MFG) from bovine cream and the changes that take place in the corresponding lipid 16 membranes (MFGM). Creams were washed with water (1:10) and subjected to either 17 centrifugation or microfiltration to fractionate proteins and other non-fat milk components. 18 Protein analyses of the obtained fractions were carried out by gel electrophoresis. Lipid 19 extraction and thin layer chromatography were also employed to separate lipid types and the 20 amount of polar lipids were determined by gas chromatography. The effect of flow conditions on 21 MFG's colloidal properties and MFGM components was evaluated based on estimates of the 22 average rate of energy dissipation in microfiltration and centrifugation processes. Both were 23 equally effective in removing the protein fraction (93% yield) as well as non-fat dry matter 24 (~100% removal). Microfiltration reduced the mean particle size by 0.3 μ m, whereas the 25 opposite was observed for centrifugal separation (average size increase by $0.8 \mu m$). The latter 26 process also induced a more significant reduction in the electrostatic charges (zeta potential) of 27 the colloids in the cream, which relates to the changes in the milk fat globule surface 28 composition and the release of MFGM components. The dissociated polar lipids amounted to 29 24% and 20% upon centrifugation and microfiltration, respectively. Overall, the results suggest 30 that MFG and MFGM are partially damaged under the shear forces typical of centrifugal and 31 microfiltration separation. A high separation efficiency, with minimal fat globule damage and 32 high MFGM yield is possible by adopting microfiltration under carefully optimized conditions. 33

34 Keywords: milk fat globules, microfiltration, centrifugation, milk fat globule membrane
35 (MFGM), shear

36

57

37 1 Introduction

38 The polar lipids of milk provide nutrients that support the growth, health and 39 development of neonates and infants [1-3]. The lipid composition in bovine milk resembles more 40 to that of human milk compared to plant-derived sources, which differ in chemical, physical and 41 structural properties, leading to different digestion pathways [4-6]. The polar lipids in milk 42 consist mainly of phospho- and sphingolipids that are located at the milk fat globule membrane 43 (MFGM) [7,8], which protects the fat globules, e.g., prevents coalescence and enzymatic 44 degradation. In the course of butter making, the polar lipid membrane is detached from the 45 surface of the fat globules and accumulates in the aqueous phases, such as buttermilk and butter 46 serum. Related compositions can be concentrated and incorporated in infant formulas. However, 47 any losses of polar lipid membranes, which typically occurs upon heating, agitation, pumping 48 and homogenization, will limit their otherwise potential nutritional value [9-12]. The separation 49 of polar lipids from dairy fractions has been pursued by using different approaches but their 50 efficiency and limited isolation yield remain challenging, mainly because process-induced 51 depletion and the presence of impurities in the respective complex systems [13-15]. 52 Recent studies have examined process-induced damages on MFGM upon separation [16-53 18]. For instance, Holzmuller et al. [17] reported centrifugal shear forces to cause a release in 54 MFGM components. Moreover, it has been noted that major losses, especially in components 55 located in the outer layers of the MFGM, occur upon increasing the number of centrifugal 56 processing cycles. In addition, cream washing may selectively remove MFGM components from

58 depend on the specific conditions applied during handling and processing. For instance, Brans et

the interface and decrease stability [16,19,20]. The extent of MFGM rupture is expected to

al. [21] have proposed that compared to traditional centrifugal separation, less MFGM damage
may occur during low-shear microfiltration. However, if applied for extended time,

microfiltration can still be detrimental to MFGM, as evidenced by the changes in the size of milk
fat globules [22].

63 Despite its importance, and to the best of our knowledge, the effect of microfiltration on 64 MFGM damage and the fate of polar lipids have not been assessed so far. Therefore, this study 65 examines the extent of MFGM rupture under the influence of shear and processing time during 66 microfiltration and centrifugation, two of the most typical operations used in dairy separation. 67

68 **2** Experimental

69 2.1 Materials

Bovine cream (fat 40.9±0.7%, protein 2.11±0.04%, dry matter 46.3±0.7%, pH 6.75±0.05)
was collected and pasteurized (75 °C and 20 s) in a dairy unit (Valio Ltd. Jyväskylä, Finland).
The samples were fresh and no more than one day under storage at 5 °C elapsed prior to each
experiment. All reagents used were analytical grade.

74 2.2 Microfiltration

Filtrations were carried out with a uniform trans-membrane pressure (UTP) filtration
system (Tetra Alcross®, Tetra Pak, Denmark). A multichannel tubular ceramic membrane was
adopted (1.4 µm pore size and 0.24 m² filtration surface area, Membralox®, Pall Corporation,
France). The filtration system was cleaned before and after each experiment according to [23].
Briefly, a three-step washing procedure was adopted, including base-acid-base steps at

temperatures 75 °C, 50 °C, 75 °C, respectively. Finally, the systems was rinsed and pure water
flux was determined at the end of the washing procedure.

82 A dilution-concentration procedure was employed in three cycles to study the effect of 83 processing on milk fat globules (MFG). In particular, 10 kg of cream was diluted (1:10 ratio) 84 with deionized water and heated to 50 $^{\circ}$ C in a water bath. The filtration system was heated to 50 85 °C with deionized water and drained prior addition of the diluted cream into the system. A 15-86 minute start-up stabilization time was allowed by circulating retentate and permeate at the 87 beginning of each cycle before collecting the permeate. Then, diluted cream (100 kg) was 88 filtered until the mass of the retentate reached the initial amount of cream (10 kg). The resultant 89 retentate was collected and the procedure was repeated for three cycles.

The microfiltration temperature was kept at 50–55 °C throughout the experiments and the filtration was operated at 0.6 bar UTP. Samples were withdrawn after each cycle as retentate and permeate and stored at 8 °C before analysis. Three replicate filtrations were performed for each experimental condition.

94 2.3 Centrifugal separation

95 Centrifugal separation was performed in three washing cycles similarly to microfiltration.
96 Briefly, 10 kg of cream was diluted with deionized water (1:10 ratio) and separated into cream
97 and skim fractions using disk stack centrifugal separator (Frau CN2S, Italy) equipped with 51
98 disks (outer radius 54.5mm, height 51mm, angle 52°, inner radius 15mm). Three separation
99 cycles were concluded and samples were collected after each one. The temperature was kept
100 between 50 and 55 °C during the whole process. The experiments were performed in triplicate.
101 2.4 Analyses

102 **2.4.1 Composition.** Dry matter content was analysed gravimetrically by drying the samples 103 overnight at 105 °C [24]. Total fat content was analysed according to the extraction method of 104 Röse-Gottlieb (ISO 1211, IDF 1:2010). The Kjeldahl method was used to determine the total 105 nitrogen and a conversion factor 6.38 was applied to calculate the protein content in the sample, 106 according to the standards (ISO 8968-1:2014, IDF 20-1:2014).

107

2.4.2 Protein analysis by gel electrophoresis. Proteins in the unprocessed cream, 108 microfiltered cream and separated cream were analysed by reducing tricine-sodium dodecyl 109 sulphate polyacrylamide gel electrophoresis (SDS-PAGE). The cream samples were diluted in 110 water to a protein concentration $\sim 40 \,\mu\text{g/mL}$ and mixed (3:1) with tricine sample buffer (0.1 M 111 Tris-HCl buffer pH 6.8, 20% glycerol, 4% SDS, 10% beta-mercaptoethanol, 0.02% 112 bromophenol blue). The proteins were denaturated at 100 °C for 5 min. After cooling down to 113 room temperature, the samples were centrifuged and loaded in a Criterion 4-20% TGX gel (Bio-114 Rad Laboratories Inc., US). The amount of protein deposited per well was 10 µg. The running 115 buffer contained trizma base (3.03 g/L), glycine (14.42 g/L) and SDS (1.0 g/L). The 116 electrophoretic migration for the gel was performed at 200 V and 100 mA over 65 min. The 117 proteins were stained with Coomassie blue over 60 min with stirring. After staining, the gel was 118 destained with a solution containing 40% ethanol and 10% acetic acid for 15 min and then 119 overnight with a solution containing 5% ethanol and 7.5% acetic acid. After washing, the gel was 120 stored in deionized water and scanned using a Gel Doc EZ Imager (Bio-Rad Laboratories Inc., 121 US). The molecular mass of proteins present in the different bands was determined from the 122 migration of molecular mass markers (All Blue Precision Plus Protein standards; Bio-Rad

123 Laboratories Inc., US) and compared with other reports [4,14,15,25,26].

124	2.4.3 Lipid extraction. Lipids from the permeate and skim fractions were extracted by
125	lyophilizing 15–30 mL of the sample and weighting the dry material (~20 mg) into 10 mL
126	Kimax [®] tubes with PTFE-faced rubber lined caps. The powder was dissolved in 1 mL of 0.9%
127	NaCl, followed by 4 mL dichloromethane-methanol (2:1) addition. The tubes were vortexed for
128	15 s and shaken vigorously (300 rpm) for 30 min at room temperature. Then, the phase boundary
129	was sharpened by centrifugation (1600 g for 5 min), and the organic layer at the bottom was
130	transferred to a clean $Kimax^{\ensuremath{\mathbb{R}}}$ tube and evaporated under N_2 stream. The aqueous layer was re-
131	extracted with dichloromethane-methanol (20:1) and the organic layer was combined with the
132	first extract. The extraction of the lipids from the cream fractions was carried out
133	correspondingly, except 100 μ L of sample volume was applied and supplemented with 900 μ L of
134	0.9% NaCl.
135	2.4.4 Lipid class separation. Thin layer chromatography (TLC) was employed to
120	
136	separate lipid classes. The extracted lipid samples from permeate and skim fractions were
136	separate lipid classes. The extracted lipid samples from permeate and skim fractions were supplemented with 10 μ L of the standard lipid containing 1,2-Dinonadecanoyl-L- α -
136 137 138	separate lipid classes. The extracted lipid samples from permeate and skim fractions were supplemented with 10 μ L of the standard lipid containing 1,2-Dinonadecanoyl-L- α - glycerophosphorylcholine (Larodan, Sweden) 11.4 mg/mL dissolved in chloroform and 200 μ L
136 137 138 139	separate lipid classes. The extracted lipid samples from permeate and skim fractions were supplemented with 10 μ L of the standard lipid containing 1,2-Dinonadecanoyl-L- α - glycerophosphorylcholine (Larodan, Sweden) 11.4 mg/mL dissolved in chloroform and 200 μ L of dichloromethane-methanol (100:1). Then, the samples were applied to the TLC plate
136 137 138 139 140	separate lipid classes. The extracted lipid samples from permeate and skim fractions were supplemented with 10 μ L of the standard lipid containing 1,2-Dinonadecanoyl-L- α - glycerophosphorylcholine (Larodan, Sweden) 11.4 mg/mL dissolved in chloroform and 200 μ L of dichloromethane-methanol (100:1). Then, the samples were applied to the TLC plate (Silicagel 60, glass plates, 20x20, Merck, Germany) and let to develop in the chamber containing
 136 137 138 139 140 141 	separate lipid classes. The extracted lipid samples from permeate and skim fractions were supplemented with 10 μ L of the standard lipid containing 1,2-Dinonadecanoyl-L- α - glycerophosphorylcholine (Larodan, Sweden) 11.4 mg/mL dissolved in chloroform and 200 μ L of dichloromethane-methanol (100:1). Then, the samples were applied to the TLC plate (Silicagel 60, glass plates, 20x20, Merck, Germany) and let to develop in the chamber containing elution solution petroleum ether-diethylether-acetic acid (80:30:1) for 50 min. The plate was
 136 137 138 139 140 141 142 	separate lipid classes. The extracted lipid samples from permeate and skim fractions were supplemented with 10 μ L of the standard lipid containing 1,2-Dinonadecanoyl-L- α - glycerophosphorylcholine (Larodan, Sweden) 11.4 mg/mL dissolved in chloroform and 200 μ L of dichloromethane-methanol (100:1). Then, the samples were applied to the TLC plate (Silicagel 60, glass plates, 20x20, Merck, Germany) and let to develop in the chamber containing elution solution petroleum ether-diethylether-acetic acid (80:30:1) for 50 min. The plate was sprayed with 0.001% Rhodamine 6G aqueous solution and the fat containing spots were
 136 137 138 139 140 141 142 143 	separate lipid classes. The extracted lipid samples from permeate and skim fractions were supplemented with 10 μ L of the standard lipid containing 1,2-Dinonadecanoyl-L- α - glycerophosphorylcholine (Larodan, Sweden) 11.4 mg/mL dissolved in chloroform and 200 μ L of dichloromethane-methanol (100:1). Then, the samples were applied to the TLC plate (Silicagel 60, glass plates, 20x20, Merck, Germany) and let to develop in the chamber containing elution solution petroleum ether-diethylether-acetic acid (80:30:1) for 50 min. The plate was sprayed with 0.001% Rhodamine 6G aqueous solution and the fat containing spots were visualized under UV light and scraped off the plate. The polar lipids were dissociated off the
 136 137 138 139 140 141 142 143 144 	separate lipid classes. The extracted lipid samples from permeate and skim fractions were supplemented with 10 μ L of the standard lipid containing 1,2-Dinonadecanoyl-L- α - glycerophosphorylcholine (Larodan, Sweden) 11.4 mg/mL dissolved in chloroform and 200 μ L of dichloromethane-methanol (100:1). Then, the samples were applied to the TLC plate (Silicagel 60, glass plates, 20x20, Merck, Germany) and let to develop in the chamber containing elution solution petroleum ether-diethylether-acetic acid (80:30:1) for 50 min. The plate was sprayed with 0.001% Rhodamine 6G aqueous solution and the fat containing spots were visualized under UV light and scraped off the plate. The polar lipids were dissociated off the silica by extraction with dichloromethane-methanol-H ₂ O (8:4:3) as described above and stored
 136 137 138 139 140 141 142 143 144 145 	separate lipid classes. The extracted lipid samples from permeate and skim fractions were supplemented with 10 μ L of the standard lipid containing 1,2-Dinonadecanoyl-L- α - glycerophosphorylcholine (Larodan, Sweden) 11.4 mg/mL dissolved in chloroform and 200 μ L of dichloromethane-methanol (100:1). Then, the samples were applied to the TLC plate (Silicagel 60, glass plates, 20x20, Merck, Germany) and let to develop in the chamber containing elution solution petroleum ether-diethylether-acetic acid (80:30:1) for 50 min. The plate was sprayed with 0.001% Rhodamine 6G aqueous solution and the fat containing spots were visualized under UV light and scraped off the plate. The polar lipids were dissociated off the silica by extraction with dichloromethane-methanol-H ₂ O (8:4:3) as described above and stored frozen in N ₂ atmosphere until further processing.

146 The amount of the cream samples required for polar lipid analysis exceeded the capacity 147 of TLC-based lipid class separation procedure. Therefore, solid phase extraction (SPE) was 148 applied. The lipid samples were supplemented with 10 µL of the standard lipid containing 1,2-149 Dinonadecanoyl-L- α -glycerophosphorylcholine 9.98 mg/m dissolved in 200 μ L of 150 dichloromethane-methanol (2:1). Then, the samples were applied to the SPE tubes (3 mL 151 Supelclean[™] LC-SI SPE Tubes, Supelco, PA, USA), preconditioned with hexane. 152 The neutral lipids were eluted with 2 mL hexane-diethylether (4:1) and 2 mL hexane-153 diethylether (1:1). The polar lipids were eluted with 2 mL methanol and 2 mL dichloromethane-154 methanol-H₂O (3:5:2). The eluted fractions were evaporated under N₂ stream and stored frozen 155 in N₂ atmosphere until further processing.

1562.4.5 Determination of polar lipid concentration. The amount of polar lipid class was157analysed by gas chromatography (GC) after converting the fatty acids to methyl esters. Fatty158acids from the polar lipid fractions were methylated by adding 500 μ L of boron trifluoride in15914% methanol (Sigma-Aldrich, USA) into polar lipid samples. After flushing with N₂, the tubes160were incubated for 90 min in the boiling water bath. Then, 600 μ L of hexane-

161 methyltertbutylether (1:1) and 1.5 mL of deionized water was added into the cooled samples and 162 shaken vigorously (300 rpm). The aqueous phase was removed and the organic phase was 163 washed with 3 mL of 0.01% (w/w) NaOH. To sharpen the phase boundary, the tubes were 164 centrifuged for 15 min (1600 g). The organic phase was dried with anhydrous sodium sulphate 165 and transferred to the GC vial. To concentrate the samples, the solvent was evaporated under 166 nitrogen stream and the methyl ester residue was dissolved in 100 μ L of hexane.

167 The methyl esters were separated on Zebron ZB-FAME column (60 m x 250 μm x 0.2
168 μm) in an Agilent 7890 A GC system. The temperature in the oven was raised gradually to 280

°C and the gas flow in the detector was 350 mL/min, 30 mL/min and 35 mL/min for air, H₂ and
N₂, respectively. Split ratio 20:1 was adopted. The concentration of the polar lipids was
calculated by comparing the peak areas of the sample methyl esters to the peak area of the
methyl ester of the internal standard lipid.

2.4.6 Particle sizing and zeta potential. Particle size distribution was measured for creams
before and after the processes with a Mastersizer 2000 (Malvern Instruments, Malvern, UK)
based on light scattering according to [27] and [23]. A refractive index of 1.458 was adopted for
the dispersed phase and the samples were analysed in triplicate.

177 The zeta potential of the milk fat globules was measured on the following day after 178 processing using an electrophoretic mobility instrument (Zetasizer, Nano series ZS90, Malvern 179 Instruments, Malvern, UK) according to Michalski et al. [23]. The samples were diluted 500-fold 180 in 10mM saline (pH 6.8) and measured in a Malvern Dip Cell at 20 °C. Three readings were 181 collected for each individual sample and the measurements were run in triplicate.

182 2.4.7 Shear analysis. The effect of flow conditions on milk fat globules was evaluated 183 based on estimates for the average energy dissipation rate $\langle \varepsilon \rangle$. The highest shear forces in disk-184 stack centrifuges are typically attributed to the feed zone wherein the suspension is accelerated to 185 high rotational velocities prior to its injection to the stack [28,29]. The local energy dissipation 186 rate can vary considerably depending on the feed zone geometry [28,29]. In this study, a direct 187 suspension feed mechanism into the disk-stack was employed resulting in the inlet tube 188 essentially comprising the feed zone. Therefore, the separation processes shared a similarity in 189 that they took place under high Reynolds number flow (Re $\approx 2 \times 10^4$ in microfiltration; axial/tangential Re $\approx 4 \times 10^3$ and 9×10^4 in centrifuge inlet) within an axisymmetric geometry. 190

191 For centrifugal separation, $\langle \varepsilon \rangle$ was approximated by the power *P* per unit mass *m* needed to 192 accelerate a fluid volume of a size equivalent to the inlet region V_{in} from rest to the wall 193 tangential velocity ωR_{in} during a residence time $T_{in} = V_{in}/Q$:

194
$$\langle \varepsilon \rangle = \frac{P}{m} = \frac{\frac{1}{2}Q\rho\omega^2 R_{\rm in}^2}{\rho V_{\rm in}} = \frac{\omega^2 R_{\rm in}^2}{2T_{\rm in}},\tag{1}$$

195 where *Q* is the volumetric flow rate, ρ is the density, R_{in} is the inlet channel radius and ω is the 196 angular velocity [30]. For microfiltration, $\langle \varepsilon \rangle$ was estimated by assuming a simple axial turbulent 197 duct flow that neglects the filtrate flux over the channel wall [31,32]. Utilizing a force balance 198 for a microfiltration channel, the wall shear stress was expressed using the measured axial 199 pressure drop (ΔP) by $\tau_w = d\Delta P/(4L)$, where *d* is the channel diameter and *L* is its length 200 [31,32]. The mean energy dissipation rate was determined from:

201
$$\langle \varepsilon \rangle = \frac{2f\langle v \rangle^3}{d},$$
 (2)

where $f = 2\tau_w/(\rho \langle v \rangle^2)$ is the Fanning friction factor and $\langle v \rangle$ is the measured mean axial flow velocity. To evaluate how the turbulent flow conditions affect particle flocculation and coalescence, the size of the smallest turbulent eddies (Kolmogorov microscale) and their characteristic response time (Kolmogorov timescale) were calculated according to Eqs. (3) and (4) by using the continuous phase kinematic viscosity (v):

207
$$\eta = \left(\frac{\nu^3}{\langle \varepsilon \rangle}\right)^{1/4} \tag{3}$$

208 and

209
$$\tau_{\eta} = \left(\frac{\nu}{\langle \varepsilon \rangle}\right)^{1/2}.$$
 (4)

210 Furthermore, the Stokes number

211
$$St = \frac{\tau_{\eta}}{\tau_{p}}$$
(5)

was derived using the particle response time, $\tau_p = D^2 \rho_p / (18\mu)$, where *D* and ρ_p are the particle diameter and density, respectively, and μ is the continuous phase dynamic viscosity [33].

214

215 **3 Results and discussion**

216 **3.1 Process efficiency**

217 Filtration performance was evaluated based on the permeate flux and permeation of non-fat 218 solids. The permeate flux decreased with increasing filtration cycles, as explained by 219 concentration polarization. Moreover, the concentration of the feed increased at the end of each 220 filtration cycle, making the filtration more arduous (Fig. 1). Nevertheless, the permeate flux 221 remained much higher compared to that previously reported for cream microfiltration [22]. 222 Evidently, the dilution of cream with water decreased the concentration and increased the 223 permeate flux. Note: it should be noted that the current setup used for cream microfiltration was 224 not optimal in terms of water consumption and process efficiency but it was chosen to reproduce 225 the conditions sued in the centrifugal separation process (i.e. three washing and separation cycles 226 combined with drainage instead of diafiltration to achieve valid comparison).





Figure 1. Effect of processing time or cycles during microfiltration and centrifugal separations using a three-cycle cream washing. Microfiltration performance is expressed in terms of average permeate flux (left axis) for the three repetitions and the area indicated in grey corresponds to the standard deviation around the averaged profile. Centrifugal separation is plotted in terms of separation mass, right axis

233

The flux was not measured from the centrifugal separation process. However, the

235 processing time for each separation cycle remained constant, indicating no reduction in process

efficiency between the separation cycles (Fig. 1). The average total processing time for

237 microfiltration was ~5.5h, whereas it was 3.5h for the centrifugation process. Hence,

238 microfiltration involved a longer exposure to shear compared to centrifugal separation, leading to

239 possible damages in the fat globules and MFGM.

The potential microbial spoilage was monitored by pH measurements upon processing andno reduction in pH was observed in any of the experiments (data not shown).

242 **3.2 Composition**

243 3.2.1 Chemical composition. The effect of processing on cream composition and MFGM 244 integrity was evaluated via protein, dry matter and lipid analyses. Despite the differences in 245 separation methods and processing times, similar cream compositions were obtained in the 246 products. Both microfiltration and centrifugal separation reduced 85%, 92% and 93% of the 247 cream protein after cycles 1, 2 and 3, respectively. The fat content of the resultant cream varied 248 between 35% and 38%. The permeate and the skim fractions contained no detectable dry matter 249 nor protein after cycles 2 and 3, indicating that everything was separated, except for the proteins 250 that were possibly bound to the fat globules. The amount of protein remaining in the resultant 251 cream was 0.17% and 0.14% after microfiltration and centrifugal separation, respectively.

252 3.2.2 Protein profiling. An electrophoretic analysis was performed to (i) identify the 253 residual proteins in the creams; (ii) to visualize the presence of membrane associated proteins 254 and, (iii) to compare the relative amount of MFGM proteins and milk serum proteins. The results 255 showed a reduction in caseins with increasing processing cycles (Fig 2). Moreover, the processed 256 creams had more intense bands for high-molecular-weight proteins compared to the unprocessed 257 one. The intensities of the bands between 40 and 80 kD, identified as MFGM proteins, increased 258 with increasing processing cycles indicating that they were concentrated by the washing instead 259 of being released from the MFGM surface. Similarly, bands corresponding to xanthine oxidase (XO) were intensified with the processing cycles for both microfiltered and separated cream, 260 261 indicating that XO was not released from MFGM. Contrarily, Holzmuller et al., [17] reported a 262 decrease in XO upon cream washing and centrifugal separation explained by high shear,

263 coalescence and incorporation of air upon processing. High-molecular-weight protein

264 MUC1/Myocin was identified in the microfiltered cream samples but less in cream or separated

cream samples.

kD	Std	Rf	MF1	MF2	MF3	C1	C2	C3		Rf	MF1	MF2	MF3	C1	C2	C3
250 -	_							1	Muc1/Myocin	0.2	1.1	1.6	2.1	0.8	0.1	0.2
150 -	_		-	-	-		-	+	XO	1.5	2.9	5.4	6.6	3.7	7.1	8.5
100 -	_							1	LP, LF, CD36	2.0	2.3	0.3	1.6	1.0	1.3	4.1
75 -	-	_		-	1		-	1	BTN	4.5	7.1	8.6	9.1	6.8	12.7	12.9
50 -	-		-	=			-	Ŧ	Adipo.	0.5	1.4	1.7	2.9	0.7	2.2	3.6
37 -	_	-		1	1	_		-	PAS6/7	2.3	4.6	10.0	8.4	4.9	8.6	9.4
25 -	_		-					+	- CNs	61	51	39	34	49	28	18
20 -	-	_	_	-		-	-		B-Lg PP3	20	26	31	35	29	37	39
15 - 10 -	_	=							aLa	7.2	4.1	2.9	0.4	4.5	3.1	3.8





It is worth noting that a "tailing" of the bands was observed in cycle 3 cream samples (Fig. 2), which might have affected the intensity measurements. This was most likely caused by the residual lipids in the samples after cycle 3 that were not diluted because of their very low protein concentration. 279 The results indicate that the cream became more concentrated with high-molecular-280 weight proteins while the amount of caseins decreased. The relative amount of membrane-281 associated proteins, such as XO, butyrophilin (BTN), cluster of differentiation 36 (CD36), 282 adipophilin and PAS 6/7 increased with increasing washing cycles for both processes, suggesting 283 that the proteins remained in the MFGM. Moreover, myocin/mucin 1 was visualized in 284 microfiltered creams with increasing intensity within washing cycles. Such concentration was 285 not observed in centrifugal separation, indicating that some loosely attached proteins might have 286 been released. Similar results for microfiltration were reported by Steffen et al. [34]. 287 The centrifugal process resulted in a steeper decrease in caseins after the three washing

cycles compared to microfiltration. However, incorporation of a-La was more prominent in centrifugal separation. While the method adopted here has limitations in protein quantification, nonetheless, it gives a good indication of the protein distribution and supports the findings from the other analyses discussed herein.

292 Steffen et al., [34] recently elucidated the effect of pasteurization on protein adsorption 293 upon cream microfiltration. As expected, casein and beta-lactoglobulin (B-Lg) were found to 294 adsorb on MFGM when pasteurization was done prior to microfiltration. In contrast, no casein or 295 B-Lg were present in MFGM material if pasteurization was applied after microfiltration. Hence, 296 it is reasonable to propose that the shear applied during microfiltration did not necessarily cause 297 protein adsorption since caseins and B-Lg were already adsorbed before filtration (and 298 centrifugal separation). Nevertheless, the process was able to remove most of the caseins. 299 Overall, pasteurizing the cream after filtration is expected to maximize the separation efficiency. 300 3.3.3 Polar lipid composition. The mass balance of polar lipids was carried out for both 301 microfiltration and centrifugation to further estimate the effect of shear on MFGM and fat

302	globule disruption. The initial creams in microfiltration and centrifugal separation contained
303	1.58±0.27 mg/g (16.43±2.81 g) and 1.54±0.13 mg/g (16.02±1.38 g) polar lipids, respectively,
304	which are in good agreement with those reported previously [35] and [36]. The microfiltration
305	process yielded 13.1 g, 9.6 g and 8.5 g polar lipids in the cream after washing cycle 1, 2 and 3,
306	respectively (Fig. 3A). Clearly, the smallest fat globules (diameter < 1.4 μ m) permeated during
307	the first cycle, causing a decrease in the polar lipid mass. Reduction in the polar lipid mass was
308	less prominent after cycle 2 and only 2% of the cream polar lipids permeated after cycle 3.
309	The centrifugal separation process yielded 12.4 g, 9.8 g and 9.3 g polar lipid after cycle 1,
310	2 and 3, respectively; very similar results compared to those from the microfiltration process.
311	However, more polar lipids permeated especially after cycle 3 (6%) during centrifugation (Fig.
312	3B). Therefore, the total polar lipid permeation was slightly higher in centrifugal separation
313	compared to that of microfiltration.





Figure 3. Polar lipid mass balance for cream after microfiltration (A) and centrifugation (B)
processes, and relative amounts of polar lipids from fat in fractions from microfiltration (C) and
centrifugation (D). Percentages in A and B indicate the permeation of polar lipids in each three
washing cycles and in cumulative total.

319

320	Rombaut et al. [36] reported partitioning of the smallest fat globules upon skimming of
321	milk to significantly decrease the content of milk polar lipids. In particular, raw milk was
322	reported to contain 0.04% polar lipids, which contributes less than 1% of the total fat. Skim milk
323	contained 0.1% fat (less than 1 μ m in size combining 25% of fat globule numbers) from where
324	polar lipids were reported to represent 19% [36]. Hence, small fat globules comprise large
325	surface area (relatively more polar lipids) and, therefore, among them, up to half of the original
326	milk polar lipids may have permeated [37].

327 To estimate whether the permeated fat globules were intact or damaged, the ratio of polar 328 lipid-to-fat was examined. After cycle 1, microfiltration and centrifugal separation induced 0.9% 329 and 1.0% permeation of fat, from where 7% and 6% comprised of polar lipids, respectively (Fig. 330 3 C,D). Presumably, this is due to permeation of small fat globules enriched with polar lipids. 331 Theoretically, filtration with 1.4µm membrane cut-off removes 5% and 32% of the volume and 332 numbers of milk fat globules (globules smaller than $1.4\mu m$), respectively [37]. If the initial 333 skimming of milk is taken into account, then microfiltration of cream is expected to remove fat globules of sizes between 1 μ m and 1.4 μ m. This yields a ~7% reduction in the number of fat 334 335 globules. Hence, it may be argued that the loss of small fat globules contributed to the loss of 336 polar lipid material upon the first washing cycle. The ratio of polar lipid-to-fat decreased after 337 cycle 2 and 3 indicating a decrease in small fat globule permeation and possible leakages of the 338 larger milk fat globules. Noticeable, the surface active proteins also permeated at this stage, 339 leaving the affected fat globules with no surface coverage and making them prone to 340 coalescence. The ratio of polar lipid to fat in microfiltration permeate decreased linearly with 341 increasing washing cycles (Fig. 3C). Contrarily, in the skim fraction the ratio remained the same

after cycle 2 and 3 (Fig. 3D), indicating that more polar lipids may have been disintegrated from
the native fat globules upon centrifugal separation compared to that of microfiltration. A loss of
total polar lipid material in the mass balance calculations is noted (Fig. 3), owing to the variation
among raw materials, methods and analysis.

346 It has been noted in previous reports that an increased number of washing cycles in 347 centrifugal separation cause depletion of MFGM polar lipids [18,35]. The loss in polar lipid 348 material may cause an imbalance in the original ratio between individual polar lipids and loss of 349 certain polar lipids. The polar lipids arrange on MFGM surfaces into ordered lipid domains and 350 liquid disordered phase, depending on their conformation and properties. The ordered domains 351 usually comprise of sphingomyelin and cholesterol and have higher resistance to rupture 352 [18,38,39]. Previously, Zheng et al. [18] indicated the desorption of PE and PC upon centrifugal 353 separation, given that these lipids are loosely bound. Hence, it can be expected that both, 354 microfiltration and centrifugation, induced changes in the MFGM, e.g., both released some of the 355 loosely attached polar lipids. The effect of microfiltration on individual polar lipid desorption 356 remains a subject for elucidation.

551

357 **3.3 Estimation of shear-induced damage**

358 3.3.1 Particle size and zeta potential. The shear-induced damage was estimated based on 359 particle size and zeta potential measurements. The particle size was expected to increase due to 360 permeation of the smallest fat globules and casein micelles [23]. However, microfiltration 361 decreased the mean fat globule size slightly after the first cycle (Table 1) after which the particle 362 size remained constant. Thus, the actual reduction in particle size was presumably much 363 stronger, indicating possible rupture of the milk fat globules. However, the particle size did not 364 decrease after cycle 2 and 3 suggesting that such size reduction did not dependent on the shear or

365 shearing time. Bullon et al., [40] reported a reduction in fat droplet size after 30 minutes 366 microfiltration, after which the size remained constant. This was explained by the breaking of the 367 emulsion flocs due to shearing forces during the first minutes of the filtration. In contrast to 368 microfiltration, centrifugal separation increased the mean fat globule size after each separation 369 cycle, due to separation of the smallest particles and possible process-induced damage and 370 coalescence. Similar results were reported, for instance by Kathriarachchi et al., [41] and 371 Holzmuller et al., [17], who found an increase in milk fat globule size and coalescence with 372 increasing separation cycles upon cream washing. An open question remains whether these 373 effects caused a difference in cream composition and disruption of milk fat globule membrane 374 components.

375

376 **Table 1.** The effect of microfiltration and centrifugal separation processing cycles on milk fat

Particle size µm,	D [4. 3]
4.1±0.12	
3.84±0.10	
3.78±0.07	Microfiltration -0.3
3.77±0.05	
4.26±0.03	
4.49±0.10	+0.8
4.87±0.14	
	Particle size μm, 4.1±0.12 3.84±0.10 3.78±0.07 3.77±0.05 4.26±0.03 4.49±0.10 4.87±0.14

377 globule particle size.

The surface charge of the milk fat globules were estimated by means of electrophoretic mobility and zeta potential measurements. The zeta potential of the fat globules decreased upon process for both separation methods (Fig. 4). The decrease in zeta potential was more pronounced in centrifugation after cycles 2 and 3 compared to that of microfiltration, which reached a plateau after cycle 1. The results for zeta potential coincided with those of particle size, indicating that the given processing modified the surface of the milk fat globules and possibly released MFGM components due to shear stress and change in the equilibria.



386

Figure 4. The effect of cream washing cycles (1-3) upon microfiltration (triangle symbols) and
 centrifugal separation (circle symbols) on milk fat globule surface charge as measured by the
 zeta potential.

390

A more negative zeta potential of milk fat globule upon microfiltration was
 reported previously by Jukkola et al., [23] and was explained by a decrease in mineral

393 composition. Furthermore, Bourlieu et al. [42] reported an accumulation of free fatty acids on the 394 MFGM that made the zeta potential more negative (and increased the electrostatic repulsion). In 395 addition, incorporation of air and washing were shown to release MFGM components and lower 396 the zeta potential [16]. Various factors can be considered to influence the reduced zeta potential. 397 After cycle 1, in both processes, the lower mineral concentration decreased the counterion 398 concentration, which made the fat globules more negatively charged (Fig. 4). After cycles 2 and 399 3, all the minerals were permeated. Hence, the zeta potential remained constant in the 400 microfiltration process. The lower zeta potential in centrifuged cream after cycles 2 and 3 are 401 explained by the formation of free fatty acids, coalesced fat droplets and possible release of 402 MFGM components.

403 In order to evaluate the effect of shear on milk fat globules upon microfiltration and 404 centrifugal separation, the flow conditions of both process need to be estimated. Moreover, 405 contradictory information has been reported about shear-induced damage on milk fat globules 406 caused by processing. In particular, microfiltration has been documented to cause less damage to 407 shear-sensitive milk components compared to centrifugal separation [21,43]. Nevertheless, long 408 shear exposure in microfiltration has been shown to decrease the mean fat droplet size in cream 409 MF [22] and several studies reported a decrease in droplet size of an oil-in-water emulsion upon 410 microfiltration [40,44,45], suggesting that microfiltration can disrupt native milk fat globules. 411 Moreover, membrane surface properties and droplet size/pore diameter ratio was reported to play 412 an important role in the passage of oil droplets during filtration and droplet breaking phenomena 413 [45,46]. In addition, hydrophobicity of the filtration membrane influences the filtration of an 414 O/W emulsion: hydrophobic membranes are prone to partially break an emulsion, whereas 415 hydrophilic membranes are able to retain the properties of the emulsion [40]. Centrifugal

separation was reported to cause high shear and disruption of milk fat globules and MFGM in
various studies [17,18,41]. For instance, membrane protein-based Periodic acid and Shiff (PAS)
6/7 was found to diminish by 80% after the first washing cycle in cream separation process [17];
it is well known that proteins and phospholipids may be released from MFGM surface upon high
shear, causing fat globule coalescence [18,35]. Such partitioning could contribute to the lowering
of zeta potential in the present study.

422 3.3.2 Influence of flow conditions on milk fat globules. Based on the sigma model [47], 423 the particle diameter with 50% separation efficiency is ca. 800 nm in the centrifugal separation, 424 which suggests that the increase in mean particle size was not likely caused by an inadequate 425 separation efficiency but by coalescence and flocculation promoted by the flow conditions [48-426 50]. To compare the potential for droplet breakage during centrifugal separation and 427 microfiltration, the average energy dissipation rate $\langle \varepsilon \rangle$ for both processes was estimated 428 according to equations 1 and 2, respectively. For centrifugal separation, $\langle \varepsilon \rangle$ resulted ≈ 60 W/kg (Eq. 1), corresponding to a turbulent eddy size of $\eta \approx 13 \ \mu m$ (Eq. 3) and response time $\tau_{\eta} \approx$ 429 140 µs (Eq. 4). For microfiltration, the average energy dissipation rate $\langle \varepsilon \rangle \approx 1200$ W/kg was 430 431 found much higher than for centrifugation, implying a smaller eddy size $\eta \approx 6 \,\mu\text{m}$ and a faster response time $\tau_{\eta} \approx 30 \,\mu s$. The large difference in $\langle \varepsilon \rangle$ comparing microfiltration and centrifugal 432 433 separation is partly anticipated if one considers the dimensions of the channels, flow 434 characteristics and the effect of axial rotation on turbulent flow, for example, in reducing the 435 friction factor and inducing flow laminarization [51-53]. 436 Although the collision rate can be enhanced by turbulence, coalescence was not observed 437 in microfiltration. Moreover, the particle pair relative velocity differences in collisions do not

438 likely contribute to an increased orthokinetic coalescence due to the small value of the Stokes

139 number (St << 1, Eq. 5), i.e. the particles follow streamlines under both flow conditions [54]. In 140 turn, droplet breakup and deflocculation are more pronounced in microfiltration due to its larger 141 $\langle \varepsilon \rangle$ as the droplets are subject to larger hydrodynamic stresses that scale according to $\langle \varepsilon \rangle^{1/2}$ [55-142 57]. Conversely, the larger turbulent eddy size of centrifugal separation could be less effective in 143 disintegrating flocs, while the particle size polydispersity could increase flocculation and 144 collision rate by differential settling at the stack [48].

445 Hence, unlike previously assumed, the data implies that the shear stress during 446 centrifugation is much lower compared to that for microfiltration. However, the flow conditions 447 of the centrifugal separation cause fat globules to flocculate and coalesce whereas microfiltration 448 promotes deflocculation. The authors note that the pilot systems used here may differ 449 significantly from the systems used in industrial scale. For instance, the pilot microfiltration 450 system was operated at high crossflow velocities, much higher compared to those used in the 451 industrial plants, possibly affecting the outcome of the shearing. This together with the use of 452 diafiltration remains to be elucidated in the course of scale-up.

453 Overall, the experimental data supports our finding for microfiltration, which decreases
454 milk fat globule particle size owing to shearing and deflocculation phenomena. Conversely,
455 centrifugal separation increases the particle size as a result of flocculation and coalescence,
456 owing to the streamlined geometry of the centrifuge.

457

458 Conclusions

459 The separation efficiency of centrifugal and microfiltration separation were compared as
460 far as cream separation efficiency and MFGM yield. Both processes yield significant protein

removal, up to 93%, which is thus far, to the best of our knowledge, the highest reported forcream microfiltration.

463 Despite the efficiency of the processes with regard to protein separation, significant 464 changes are observed in the milk fat globules and MFGM. Based on particle sizing, centrifugal 465 separation favours flocculation/coalescence, as also supported by its estimated lower average 466 energy dissipation rate. Some polar lipid material is released from the MFGM, which is more 467 evident in the centrifugation process. The release of polar lipids increase with the number of 468 processing cycles, indicating that the components are loosely bound and sensitive to turbulent 469 flow conditions and to changes in the serum equilibria. Such effects play a role in the formation 470 of free fatty acids as well as in the physical properties of the cream and the (reduced) MFGM 471 vield.

472 Overall, microfiltration and centrifugation produce MFGM-enriched cream that is 473 separated from milk serum proteins. The MFGM-enriched cream can be considered in butter 474 making, to transform buttermilk side streams into high-value ingredients for novel MFGM-475 enriched foods, such as infant formulas. Compared to centrifugation, microfiltration offers the 476 benefit of a reduced water consumption and controlled selectivity, if optimized and combined 477 with diafiltration.

478 More efforts are needed to optimize the microfiltration process in order to minimize the 479 processing time, energy consumption, loss of small fat globules and desorption of the MFGM 480 components. The exploitation of the MFGM-enriched buttermilk is a subject that will be 481 addressed in a future communication.

482

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