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Koskinen, Ari M.P.; Brunner, Martin

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Biology and Chemistry of Sphingosine-Related Metabolites

Martin Brunner and Ari M. P. Koskinen*

*Laboratory of Organic Chemistry, Helsinki University of Technology, PO Box 6100, FIN-02150 Espoo, Finland.

Abstract: Metabolites of the thermophilic fungi Myriococcum albomyces and Mycelia sterilia such as myriocin (1), the mycestericins (2-8) and the sphingofungins (13-18) structurally resemble the sphingosines, important components of cell membranes. All the compounds revealed in vitro remarkable immunosuppressive activity and their pharmaceutical potential has led to the development of promising novel immunosuppressants. This review article describes the abundance, biology and chemistry of these metabolites.

1. INTRODUCTION

In 1972, two independent research groups isolated virtually simultaneously compounds named myriocin and thermozymocidin, which were metabolites of the culture broth of thermophilic fungi Myriococcum albomyces and Mycelia sterilia, respectively. The compounds were, as the more recently isolated ISP-1 from Isaria sinclairii, identical with 1 and are homologues of the sphingosines. The sphingosines are important components of all mammalian membranes, where some members play structural roles (e.g. sphingomyelin), whereas others appear to be important in (sphingosine. cellular regulation ceramide. glycosphingolipids). Although myriocin is historically the legitimate and most commonly used name for 1, the terms thermozymocidin and especially ISP-1 are used frequently and may cause confusion at times.

The initially discovered immunosuppressant 1 is the most thoroughly investigated one of the sphingosine-related metabolites and its biological profile is best known. Several total syntheses with a variety of different approaches have been accomplished during the last two decades. The more recently isolated mycestericins (2 - 8) and sphingofungins (13 - 18) with similar biological properties have attracted a much lesser interest. In recent years, some related compounds as such as falvovirin (19) and malonofungin (20) were also discovered.

The discovery of these novel immunosuppressants with completely different activity profiles lead to the development of new highly potential drug candidates. The close relation of the metabolites to the sphingosines suggests an important role in regulatory processes of eukaryotic cell membranes. It is no surprise that there has been considerable synthetic interest for these compounds.

2. ISOLATION AND BIOLOGY OF SPHINGOSINE-ANALOGUES

2.1. Isolation and Biology of Myriocin

Kluepfel and co-workers from the Ayerst Research Laboratories isolated myriocin (1) for the first time in 1972 from the fermentation broth of the thermophilic fungus Myriococcum albomyces, an ascomycete, and the compound showed strong in vitro antifungal, but no significant antibiotic activity. Acute oral toxicity of 1 in mice was 50 times lower than intraperitoneal toxicity, probably due to poor absorption. Subcutaneous injection of 0.25 mg/kg in dog resulted in death of the animals after 2-3 days (Fig. (1)) [1].

In the same year, Aragozzini and co-workers reported the discovery of thermozymocidin, a highly effective antifungal metabolite found in the thermophilic mould of *Mycelia sterilia*, with an intraperitoneal LD₅₀ of 7.5 mg/kg in mice [2]. Structure and physical properties proved to be identical to those of 1. In both cases, the constitution of 1 with the *E*-double bond was correctly assigned. However, the initially reported relative configuration of the insoluble natural product, revealed from NMR and IR data of derivatives, [3] proved to be wrong and was corrected with the synthesis of the chiral γ -lactone [4] derived from 1, and later confirmed with an X-ray structure of (+)-*N*-acetyl-anhydromyriocin (9) [5]. Eventually, 1 was completely characterised with the absolute configuration obtained from the (-)-enantiomer (10) of (+)-anhydromyriocin (Fig. (2)) [6].

Recently, 1 was isolated from the culture broth of *Melanconis flavovirens*, a non-thermophilic fungus of the class of Pyrenomycetes, [7] and as ISP-1 in *Isaria sinclairii* (also known as *Paecilomyces cicadae* [8]), the imperfect stage of the fungus *Cordyceps sinclairii* [9]. *Cordyceps* is a genus of fungus belonging to the Clavicipitacea family and is parasitic on insects larvae like *Lepidoptera adonata* (butterfly larva) or mature insects. Vegetable wasps and plant worms with the parasitic fungus *Cordyceps sinensis* Sacc. have been used in traditional Chinese Medicine as a nostrum for eternal youth [9].

Myriocin (1) is a very potent immunosuppressive agent in vitro and in vivo, equipotent to tacrolimus (FK506) [10] and approximately 5-100 times more potent than Cyclosporin A (CsA), [11] two clinically prescribed agents. Initial studies have demonstrated that 1 acts mechanistically differently than these commercial immunosuppressive drugs. Compound 1 inhibits the proliferation of a murine cytotoxic T lymphocyte cell line (CTLL-2), but unlike CsA and FK506, not the production of IL-2 [12].

Myriocin binding proteins were identified from CTLL-2. The subunits LCB_1 and LCB_2 were isolated from that cell

^{*}Address correspondence to this author at the Laboratory of Organic Chemistry. Helsinki University of Technology, PO Box 6100, FIN-02150 Espoo, Finland; Tel: (09) 451 2526; Fax: (09) 451 2538; E-mail: ari.koskinen@hut.fi

$$\begin{array}{c} \text{HO}_2\text{C} \\ \text{HO} \\ \text{HO} \\ \text{HO} \\ \text{OH} \\ \text$$

Fig. (1). Metabolites from Mycelia sterilia.

Fig. (2). Andhydromyriocin Derivatives.

line using myriocin derivatives and affinity chromatography, and are mammalian homologues of two yeast proteins that have been genetically linked to sphingolipid biosynthesis [13].

Biosynthetic studies with 14 C-labelled substrates confirmed that 1 is assembled in the mould of *Mycelia sterilia* from acetate building blocks and L-serine [14]. Substitution of the α -proton of L-serine with stearoyl-SCoA proceeded with inversion of configuration. Interestingly, the γ -hydroxyl group, the carbonyl group, and the *E*-double bond of 1 were introduced by the enzymatic systems of the mould after the substitution [15].

In CTLL-2-derived microsomes 1 inhibits serine palmitoyltransferase (SPT), an enzyme that catalyses the first step of the sphingolipid biosynthesis, [16] and in *Saccharomyces cerevisiae* the synthesis of the intermediate ceramide is suppressed by 1 by a rapid and specific decrease in the transportation rate of GPI-anchored proteins to the Golgi apparatus, which blocks cell proliferation [17].

Myriocin (1) prevents the accumulation of free sphingoid bases (sphinganine) caused by the mycotoxins fumonisins through inhibition of ceramide synthase, an enzyme in the *de* novo sphingolipid synthesis. The fumonisins B₁, B₂ and B₃ are produced by the fungi Fusarium moniliforme and Fusarium proliferatum and are found in detectable amounts in most corn containing food and feeds all over the world [18]. Elevated fumonisin concentrations are toxic to animals, but the effect for humans is not clear yet. The accumulation of sphingoid bases in serum, urine, kidney, or liver is correlated with outbreaks of equine leukoencephalomalacia (ELEM) and other farm animal diseases [19]. ELEM is usually mortal to horses, but temporary reduction of sphinganine concentration by 1 induced SPT inhibition could be exploited for promising therapeutic application in animals [20].

2.2. Isolation and Biology of Mycestericins A to G

Re-examination of the culture broth of *Mycelia sterilia* yielded the novel immunosuppressants mycestericin A (2) to G (8), which are structurally and pharmacologically closely related to 1. All mycestericins have an equivalent quaternary carbon with identical absolute configuration [21]. Moreover, the polar ends of mycestericin A(2), B(3), and C(4) have identical hydroxyl group substitution patterns as 1, but 4

lacks the double bond and 2 and 3 have different alkyl chains. Mycestericin D/E (5/6) and F/G (7/8) have only one secondary hydroxyl group and form two pairs of diastereomers (Fig. (1)).

In order to reduce toxicity and to improve physicochemical properties, structure-activity relationship studies (SAR) of derivatives of 1 to 8 in mouse allogeneic mixed lymphocyte reaction (MLR) in vitro [22] led to the development of symmetrical 2-amino-1,3-propanediols, the essential structures for immunosuppressive activity [23] and, hence, the discovery of the very effective immunosuppressant FTY720 (11) (Fig. (3)) [24].

RO
$$\begin{array}{c}
NH_2 \\
HO
\end{array}$$

$$\begin{array}{c}
\mathbf{11} (R = H) \\
\mathbf{12} (R = PO_4H_7)
\end{array}$$

Fig. (3). FTY720 and Phosphorylated Derivative.

Phosphorylated FTY720 (12), a sphingosine-1-phosphate (S1P) receptor agonist, protects organ grafts by reducing the recirculation of lymphocytes from lymphatics to blood and inflammatory tissues, and has a different mode of action than the commonly used calcineurin-inhibitors CsA and FK506, or the macrolides rapamycin and RAD. Unlike 1, this aminoalcohol did not inhibit SPT [25]. Drug candidate 11, which is currently in phase III clinical trials, is expected to become, in combination with the known calcineurininhibitors cyclosporin A (CsA) or FK506, the drug of the future for the therapeutic areas of transplantation and autoimmunity. Aminoalcohol 11 was recently found to effectively suppress cellular infiltration and tissue necrosis in a model of acute myocarditis. Clinical phase II studies with humans showed high efficacy in kidney transplantation for the combination of 11 and the novel macrolide RAD [26].

Fig. (4). Sphingofungins.

2.3. Isolation and Biology of Sphingofungins

A series of compounds called sphingofungins (13 - 18) were found during a screening program for antifungal inhibitors in the Merck, Sharp & Dohme Laboratories (Fig. (4)). Sphingofungin A (13) to D (16) were isolated from a strain of the thermotolerant fungus Aspergillus fumigatus, [27] and sphingofungin E (17) and F (18) from the fermentation broth of the thermophilic fungus Paecilomyces variotii (Fig. (4)) [28]. The metabolites 13 to 16 have the same backbone skeleton with congruent configuration with the typical quaternary carbon of myriocin and mycestericins missing. It is possible to transform sphingofungin C (13) sequentially into sphingofungin D (14). E (15) and F (16) [29]. The previously unknown absolute configuration of the C14 hydroxyl group of 15 was eventually assigned from ozonolysis products [30].

Sphingofungin E (17) and F (18) resemble myriocin (1) and the mycestericins (2 - 8), and bear an equivalent quaternary carbon centre with the opposite absolute configuration, respectively.

All sphingofungins are SPT inhibitors [31] and potent antifungal agents against various *Candida* species, but are essentially inactive against filamentous fungi and bacteria. Metabolite 16 is much less potent than 13 to 15 [27,28].

2.4. Isolation of Miscellaneous Sphingosine Analogues

Further examples of sphingosine-related metabolites with a quaternary carbon centre are flavovirin (19) and malonofungin (20). The antibiotic 19 was found in the pyrenomycete *Melanconis flavovirens* in the course of isolation of 1 and is 5 to 10 times more active than 1 against filamentous fungi, but is ineffective against bacteria [32].

As part of a screening programme for microbial metabolites with growth inhibitory activity against phytopathogenic fungi, the antifungal and antibacterial metabolite 20 was isolated from fermentations of a fungus originating from Jamaican *Panicum maximum* leaves and identified as *Phaeoramularia fusimaculans*. Malonofungin (20) shows broad growth inhibitory effects against a range of fungi belonging to the genera Botrytis, Pyricularia,

Fig. (5). Miscellaneous Sphingosine Analogues.

Conditions: a) 21. *n*-BuLi, THF; b) *m*-CPBA, cy: c) LiP(Ph₃)₂, MeI, THF; d) (*n*-Bu)₄NF·H₂O, PhH, THF; e) DMSO/Ac₂O; f) NaCN. NH₄CI, sat. NH₅/MeOH; g) HCl (g), H₂O/MeOH.

Fig. (6).

Fusarium, and Penicillium, but not against bacteria or yeast (Fig. (5)) [33].

3. TOTAL SYNTHESES OF SPHINGOSINE ANALOGUES

3.1. Introduction

Early synthetic attempts towards myriocin (1) were made when Kuo and co-workers synthesised a chiral lactone derived from myriocin and, at the same time, corrected the relative configurations of the 2-amino and 3,4-hydroxyl groups to syn, syn [4]. Payette and co-workers synthesised the (-)-enantiomer (10) of (+)-anhydromyriocin, the y-lactone derived from 1, using L-arabinose [6]. They explored a variety of synthetic routes to elongate the pentose moiety and introduced the long alkyl chain tail already at an early stage of the synthesis. However, the Wittig reaction of the alkyl chain ylide 21 with the aldehyde 22 derived from Larabinose gave predominantly the undesired Z-isomer 23 (Z:E 95:5) in 67% yield, which had to be isomerised to the corresponding E-isomer 24 laboriously by the method of Vedejs and Fuchs: epoxidation of the double bond, followed by reaction with lithium diphenylphosphide, quaternisation with methyl iodide, and decomposition of the betaine intermediate gave 24. The aminocyanation of 25 to install the quaternary amino acid carbon centre of the final product gave an equivalent mixture of diastereomeric cyanoamines, which had to be separated by HPLC. Diastereomer 26 (31% yield) was hydrolysed to yield lactone 10, the optical antipode of the γ -lactone of 1 (Fig. (6)) [6].

3.2. Total Syntheses of Myriocin

In the following all published total syntheses of natural occurring (+)-myriocin (1) are presented in chronological order with focus on the utilised chiral or achiral starting material. The synthesis routes are not in all cases completely described as the present review article concentrates basically on the key parts of each of the described synthetic strategy.

3.2.1. D-Fructose (Scolastico 1982)

The first total synthesis by Scolastico and co-workers used D-fructose (27) as the chiral starting material, which was transformed into a diastereomeric mixture of cyanoamines following the Kuhn-procedure. This desired 2S-epimer 28 was obtained as a 1:3 mixture of diastereomers in favour of the undesired 2R-epimer (80% yield), which was partially converted into soluble 28 by equilibration in EtOH/H₂O in the presence of an excess of liquid hydrogen cyanide. The 2S-epimer 28 was hydrolysed and the amino

Conditions: a) p-TosNH₂; b) HCN, EtOH/H₂O; c) HCl; d) H₂/Pd, HCl; e) BzCl, Py; f) TEA, MeOH; g) acetone, H₂SO₄; h) BzCl, Py; i) H⁷/H₂O; j) NalO₄; k) NaBH₃(CN). THF; l) TsCl; m) 32; n) hydrolysis.

Fig. (7).

Conditions: a) DMP, PTS, DMF; b) NaBH₄, EtOH; c) *p*-anisyldimethylacetal, *p*-TsOH·H₂O; d) DMSO, (COCl)₂, TEA, CH₂Cl₂; e) CH₂Cl₂, LDA, THF; f) NaN₃, 15-crown-5, HMPA; g) NaBH₄, EtOH, h) MOMCl, Hünig's base, CH₂Cl₂; i) H₂, Pd/C, EtOH; j) BzCl, Py.

Fig. (8).

tolyl protection cleaved by hydrogenolysis to give 2-amino-2-deoxy-2-hydroxymethyl-D-mannonic acid Lactonisation of the amino-benzovlated 29, temporary acetalisation with acetone and subsequent benzylation of the remaining free hydroxyl groups finally yielded alcohol 30. Diol 30 was oxidised to the unstable aldehyde 31 with sodium periodate, then, reduced and tosylated. Subsequent nucleophilic substitution of the tosylate with vinyl cuprate 32 in Et₂O with HTMP as cosolvent established fullyprotected myriocin (28% yield). The E-divinylcuprate reagent 32 was generated by standard reactions from 1-morpholinocyclohexene in 9 linear steps. Global deprotection completed the synthesis of 1 in 13 linear steps (Fig. (7)) [34].

3.2.2. 2-Deoxy-D-Glucose (Yoshikawa 1994)

The group of Yoshikawa and co-workers started their stereoselective synthesis with 2-deoxy-D-glucose 33 and

used ketone 34 after double acetal protection for a modified Darzen's reaction to obtain the addition product 35 (68% yield for two steps). Treatment with sodium azide in HMPA in the presence of a crown ether as catalyst stereoselectively yielded quantitatively the aldehyde 37 via chloroepoxide 36. The aldehyde was gradually reduced to the amino acid derivative 38 (Fig. (8)).

The *p*-methoxybenzylidene group (PMB) of **38** was reductively opened to the *p*-methoxybenzyl (MPM) protected primary and free secondary hydroxyl group. Then, migration of the isopropylidene group and oxidation of the liberated primary hydroxyl group to the acid with subsequent esterification with diazomethane yielded the α,α -disubstituted amino acid derivative **39**. The MPM protection-group was cleaved; the resulting primary hydroxyl group oxidised to the corresponding aldehyde, and directly reacted with the phosphonium salt **40**, which was obtained in 8 linear steps from cyclooctanone, [6] in a Wittig reaction

Conditions: a) NaBH₃CN, TMSCl, CH₃CN; b) DMSO, (COCl)₂, TEA, CH₂Cl₂; c) NaClO₂, NH₂SO₃H, dioxane/H₂O; d) CH₂N₂, Et₂O; e) DDQ, CH₂Cl₂/H₂O, f) DMSO. (COCl)₂, TEA, CH₂Cl₂; g) 40, *n*-BuLi, *t*-BuOH/ THF, h) hv, PhSSPh, cy; i) PTS, aq. EtOH; j) 1N NaOH.

Fig. (9).

Conditions: a) $MgBr_2$, TEA, CH_3CN ; b) TBDMSOTf, 2,6-Lutidine. CH_2Cl_2 ; c) DIBALH, CH_2Cl_2 ; d) HCI, MeOH; e) NaOH. $MeOH/H_2O$; f) IRC-50 (H type).

Fig. (10).

to give predominantly the undesired Z-alkene 41 ($E:Z\sim1:3$) in 65% yield. Photochemical isomerisation in the presence of diphenylsulfide changed the E:Z-ratio to 4:1 (92% yield). The isomers were separated by HPLC and global deprotection of the E-isomer 41 afforded myriocin (1) in 20 steps in an overall yield of 5% from 33 (Fig. (9)) [35].

3.2.3. D-Valine (Nagao 1995)

Nagao and co-workers used for the first time highly diastereoselective aldol additions of amino acid derivatives as the key steps in the synthesis of 1. Unnatural D-valine was transformed into Schöllkopf's bis-lactimether 42 and then added to the linear C-17 aldehyde 43 to yield the fully protected myriocin 44 (47% for two steps). Reduction of the bis-lactimether and stepwise removal of protecting groups gave 1 in 13 linear steps (Fig (10)).

The long-chain aldehyde 43 was synthesised starting with a highly diastereoselective aldol addition of the enolate of thiazolidinethione 45 to aldehyde 46 to give the chiral aldehyde 47 in 93% de and 79% yield. Subsequent reductive cleavage of the thiazolidinethione auxiliary in 47 yielded the aldehyde 48 with the right absolute configuration of the

protected secondary hydroxyl group. Highly *E*-selective Schlosser-type Wittig reaction of the phosphonium salt 49 with 48 then provided the protected diol 50 (*E:Z* 96:4; 82% yield), which was, after deprotection of the primary hydroxyl group, oxidised to aldehyde 43 (Fig (11)) [36].

3.2.4. Achiral Starting Material (Hatakeyama 1997)

Recently, Hatakeyama and co-workers published an enantioselective synthesis based on the chemistry of 1-trimethylsilylbuta-2,3-dienes. The utilisation of achiral starting material required stereoselective reaction steps. Catalytic Katsuki-Sharpless asymmetric epoxidation of the *E*-allylic alcohol obtained from reduction of the mono protected 2-butyn-1,4-diol 51 gave the enantiomerically pure epoxide 52 (92% yield). Reduction of 52 followed by a series of protection and deprotection steps gave alcohol 53. Addition of 1-trimethylsilylbuta-2,3-diene (54) to the aldehyde obtained by Swern oxidation of 53 occurred in a highly diastereoselective manner in 78% yield over two steps. The ensuing dienol 55 was diastereoselectively epoxidised and converted to an epoxytrichloroacetamidate. Then, Lewis-acid catalysed cyclisation with inversion to

Conditions: a) Sn(OTf)₂, N-ethylpiperidine, CH₂Cl₂; b) MeOCH₂Cl, Hünig's Base, CH₂Cl₂; c) DIBALH, THF; d) PhLi, **49**, THF/Et₂O;e) PhLi, f) Silica gel; g) Li, liq. NH₃, THF; h) DMSO, (COCl)₂, CH₂Cl₂.

Fig. (11).

Conditions: a) PMBCI, KOH, DMSO; b) Red-Al * , Et $_2$ O; c) diisopropyl L-tartrate, Ti(Oi-Pr) $_4$, i-BuO $_2$ H, 4 Å MS, CH $_2$ Cl $_2$; d) Red-Al * , THF; e) i-BuCOCI, Py, CH $_2$ Cl $_2$; f) TMSCI, TEA, THF; g) PhCHO, TESH, TMSOTf, CH $_2$ Cl $_2$; h) (COCI) $_2$, DMSO, TEA, CH $_2$ Cl $_2$; i) 54. TiCl $_4$, CH $_2$ Cl $_2$; j) VO(acac) $_2$, i-BuO $_2$ H, CH $_2$ Cl $_2$; k) CCl $_3$ CN, DBU, 4 Å MS, CH $_2$ Cl $_2$; l) Et $_2$ AlCl, CH $_2$ Cl $_2$; m) HCl, THF; n) (MeO) $_2$ CMe $_2$, PTS, CH $_2$ Cl $_2$; o) O $_3$. CH $_2$ Cl $_2$, then Me $_2$ S: p) NaClO $_2$, NaH $_2$ PO $_4$, 2-methyl-2-butene, i-BuOH/H $_2$ O; q) CH $_2$ N $_2$, Et $_2$ O; r) NaOMe, MeOH; s) (COCI) $_2$. DMSO, TEA, CH $_2$ Cl $_2$; t) 49. i-BuLi, THF; u) NaOH/MeOH; v) Li, THF, liq. NH $_3$; w) HCl/MeOH; x) Ac $_2$ O, DMPA, Py; y) hv, PhSSPh, PhH; z) NaOH/MeOH, then Amberlite IRC-76.

Fig. (12).

oxazoline 56 stereoselectively established the amino group in 69% yield from 55. Acid hydrolysis and sequential diol protection, ozonolysis, oxidation, esterification and Swern oxidation gave the aldehyde 57. A Wittig reaction joined 57 with the phosphonium salt 49 (Fig. (11)), to yield the fully protected 58 as a 92:8 Z:E-mixture (48% yield). Conversion of the modified 58 into the corresponding γ -lactone and photochemical isomerisation as in Yoshikawa's synthesis completed, after global deprotection, the synthesis for 1 in 26 linear steps (Fig. (12)) [37].

3.2.5. D-Mannose (Chida 2001)

Chida and co-workers used aldohexoses for a long linear total synthesis of 1 and, similarly, sphingofungin E (17, see below), where an Overman rearrangement was the key

reaction. Compound 1 was synthesised starting from D-mannose (59). The α -methyl furanoside 60 was obtained in six steps and 61% overall yield. After Swern oxidation, Homer-Wadsworth-Emmons olefination under the Masamune-Roush conditions (E:Z 15:1) and reduction, the allylic trichloroacetimidate 61 was subjected to Overman rearrangement to give the allylic amide 62. Ozonolysis, followed by oxidation and esterification afforded intermediate 63 (Fig. (13)).

Isomerically pure *E*-allyl alcohol **64** was obtained by a Wittig reaction of acetal **63** and the stabilised phosphor ylide derived from ethyl bromoacetate (53% yield over two steps). The coupling of the alkyl sulfone **66** with the highly functionalised allyl bromide **65** yielded, after saponification and subsequent Birch reduction, the carboxylic acid **67**.

Conditions: a) H_2SO_4 /acetone; b) $(Me_2N)_3P/CCl_4$ then Li, NH₃, THF; c) NaH; BnBr; d) $Hg(OAc)_2$, THF/ H_2O , then K1, NaBH₄, THF/ H_2O ; e) AcOH/ H_2O , then AcCl, MeOH; f) n-Bu₂SnO, tol, then MPMCl, CsF, DMF; g) $(COCl)_2$, DMSO, CH₂Cl₂, TEA; h) $(MeO)_2$ -POCH₂CO₂Me, LiBr, DBU, CH₃CN, 90% from 60 (E:Z 15:1); i) DIBALH, tol; j) Cl₃CCN, DBU, CH₂Cl₂; k) K_2CO_3 , o-xylene, 90% for 2 steps; l) O_3 , CH₂Cl₂, then Mc₂S; m) NaClO₂, NaH₂PO₄, HOSO₂NH₂, t-BuOH/t-PO, then TMSCHN₂, MeOH.

Fig. (13).

Conditions: a) aq. HCI/THF, then Ph₃P=CHCO₂Et, tol, 71%; b) DIBALH, THF/tol, 75%; c) DBU, CH₂Cl₂; d) MsCl. TEA, CH₂Cl₂, then LiBr, acetone; e) *n*-BuLi, THF, then 66, 80%; f) LiOH, aq. MeOH; g) Li, liq. NH₃, THF; h) aq. HCI/THF; i) aq. NaOH/MeOH; j) Ac₂O, Py; k) aq. NaOH/MeOH.

Fig. (14).

Hydrolysis of the carbamate and the acetal protection group in 67 yielded, via anhydromyriocin, 1 in 24 steps and 5% overall yield (Fig. (14)) [38].

3.2.6. L-Serinol (Ham 2002)

Most recently, Ham and co-workers published a convergent total synthesis employing the diastereoselectively formed oxazoline 70 as a key intermediate, which was prepared from L-serinol derivative 68. Oxidation of 68 and subsequent Grignard reaction afforded a 1:1 syn/anti-mixture of allyl alcohol 69 in 70% yield over two steps. Acetyl protection of 69 followed by highly diastereoselective Pd(0)-catalysed intramolecular cyclisation yielded only trans-oxazoline 70 in 60% yield from 69. Hydroxymethylation of the α -amino carbon of 70 gave the desired anti adduct 71 in a diastereomeric ratio of 20:1 in favour of the desired isomer in 52% for two steps. Ozonolysis of the terminal double bond and MgBr₂-promoted allylic stannane addition of the

obtained unstable aldehyde was followed by another ozonolysis and iodomethylation to give the vinyl iodide 73 in 51% yield over four steps. Negishi-coupling of alkyl chain 72 as an organozinc reagent with the halide 73 and final hydrolysis finished the synthesis in 19 steps from 68 and 4.6% overall yield (Fig. (15)) [39].

3.3. Formal Syntheses of Myriocin from D-Glucose (Rao 1993, Olesker 1994)

The first and longer formal synthesis of (1) by Rao and co-workers utilised the Trost Pd-catalysed *cis*-hydroxyamination of chiral vinyl epoxides as the key step. The 5-ulose derivative 74, readily obtained from D-glucose, was alkylated with vinylmagnesium bromide. The alkene 75 was selectively converted to the epoxide 76 under Mitsunobu conditions. The resulting vinyl epoxide 76 underwent *cis*-hydroxyamination with *p*-methoxyphenyl-isocyanate and oxazolidinone 77 was obtained as the single product in 76%

.

Conditions: a) Dess-Martin periodinane, CH_2CI_2 ; b) $CH_2CHMgBr$, THF, 70% for two steps; c) Ac_2O , Py, CH_2CI_2 ; d) $Pd(PPh_3)_4$, K_2CO_3 , CH_3CN , 60%; e) TBAF, THF; f) $RuCI_3$, $K_2S_2O_8$, NaOH, CH_3CN ; g) CH_2N_2 , EI_2O ; h) HCHO, CHMDS, CHMDS

Fig. (15).

Conditions: a) $CH_2CHMgBr$, THF, 80%; b) Bu_4NF , THF; c) PPh_3 , DEAD, tol; d) TosNCO, $Pd(PPh_3)_4$, THF, $(CH_3)_2CHO)_3P$, 76%; e) O_3 , CH_2Cl_2 , $NaBH_4$; f) NaH, BnBr, THF; g) CAN, CH_2CN/H_2O ; h) NaOH, THF; i) Bz_2O , MeOH; j) $RuCl_3$, $NaIO_4$, H_2O , CCl_4 , CH_3CN ; k) CH_2N_2 , Et_2O ; l) H_2/Pd , MeOH, 35 psi, 85%; m) NH_3 , MeOH.

Fig. (16).

yield. After a series of transformations the key intermediate 79 for both of the formal syntheses was achieved in 13 steps from 74. The final known aldehyde 31 (Fig. (7)) was obtained after few transformations (Fig. (16)) [40].

The related formal synthesis by Olesker and co-workers used the same 5-ulose derivative as the starting material, but reduced the number of reaction steps to 6 for the route from 74 to 79 (Fig. (17)) [41]. Instead of the alkylation/cishydroxyamination sequence, the key intermediate 79 was obtained through a highly stereoselective Darzen's condensation with chloromethyl p-tosylsulfone. An S_N2 reaction of the α , β -epoxy chloride 80 in the presence of azide ion gave the α -azido aldehyde 81 with the desired (S)-configuration in 55% yield over 2 steps.

3.4. Total Syntheses of Mycestericins

The scientific focus in total syntheses of mycestericin A to G (2 - 8) was mainly on the two pairs of diastereomers, D/F (5 7) and E/G (6/8), since 7 and 8 are easily obtained after reduction of the double bond of 5 and 6 respectively. Here, we present the published syntheses in chronological order.

3.4.1. L-Threonine Aldolase (Mycestericin D 1996 and F 2000)

An enzymatic aldol reaction was used in the synthesis of mycestericin D (5) and F (7). L-Threonine aldolase from *Candida humicola* catalysed the reaction of

Conditions: a) LDA, CH₂Cl₂, THF; b) NaN₃, DMPU, 15-crown-5-ether, 55%; c) H₂/Pd, abs. EtOH; d) BzO₂, MeOH, 46% for two steps; e) (COCl)₂, DMSO, TEA, CH₂Cl₂, 86%; f) TBAF, THF, 97%.

Fig. (17).

BnO CHO a-c MeO₂C OBn
$$\frac{d}{OH}$$
 MeO₂C OBn $\frac{d}{OH}$ MeO₂C OBn $\frac{d}{OH}$ MeO₂C OBn $\frac{d}{OH}$ MeO₂C OBn $\frac{d}{OH}$ 85

Conditions: a) L-Threonine aldolase, glycine; b) Ac_2O , TEA, CH_2Cl_2 ; c) CH_2N_2 , Et_2O ; d) Chromatography; e) HCl, MeOH; f) PhC(NH)OMe, Et_2O/H_2O ; g) DBU, $(CH_2O)_n$, DMF; h) Ac_2O , Py, 83% for two steps; i) AlCl₃, NaI, CH_3CN , 88%; j) PCC, CH_2Cl_2 , 92%.

Fig. (18).

benzyloxybutanal **83** and glycine to a mixture of *threo/erythro* products **84** with high diastereoselectivity in favour of the *erythro* product. Pure **85** was obtained after derivatisation and chromatography and used for the subsequent formation of *cis*-oxazoline **86** with methyl benzimidate in 76% yield and with >99% *ee*. The addition of formaldehyde occurred stereoselectively and cleavage of the benzyl ether with AlCl₃ and oxidation of the primary alcohol stereoselectively gave the aldehyde **87** in 67% yield from **86** (Fig. (18)). Addition of the known phosphonium salt **49** (Fig. (11)) and subsequent photochemical isomerisation [35] established the backbone of the mycestericin D (5) and F (7). Finally, global deprotection (5) and double bond reduction (7) gave the natural products [42].

3.4.2. L-Serine (Mycestericin E and G 1995)

Mycestericins E (6) and G (8) were synthesised with the methodology of self-regeneration of stereocentres (SRS) for the formation of the quaternary stereocentres [43]. Addition of the enolate of oxazolidine 88, obtained selectively in three steps from unnatural D-serine, to acid chloride 89 gave the β -keto ester 90 in 46% yield. Reduction with NaBH₄ resulted in a 85:15-mixture of the 1'R and 1'S hydroxyl esters, which could be separated by chromatography. The compounds were

unstable on silica gel and, especially, the undesired S-isomer decomposed easily. Mycestericin E (6) was obtained after global deprotection of R-isomer 91 and reduction of the double afforded mycestericin G (8) (Fig. (19)) [44].

3.4.3. Achiral Starting Material (Mycestericin E 2001)

A second enantioselective synthesis of mycestericin E (6) used a *Cinchona* alkaloid-catalysed asymmetric Baylis-Hillman reaction where hexafluoroisopropyl acrylate was added to aldehyde 89 (X = O) in the presence of catalyst 92 (yield 47%) [45]. The resulting allylic hydroxyester 93 was epoxidised and converted to the epoxytrichloroacetimidate 94 with trichloroacetonitrile in the presence of DBU in 42% yield from 93. The nitrogen was stereoselectively installed by a Lewis-acid-promoted cyclisation of 94, a methodology already used by the same group in the synthesis of 1 (Fig. (20)) [37].

3.5. Total Syntheses of Sphingofungins

Only the sphingofungins E (17) and F (18) have the characteristic quaternary carbon present in myriocin (1) or mycestericins (2-8). The remaining sphingofungins A to D (13-16) have the same substitution pattern as the sphingosines. A formal synthesis of sphingofungin D (16)

Conditions: a) CH₃CH₂COCl, LDA, THF, 46%; b) NaBH₄, MeOH,70%.

Fig. (19).

89
$$(X = O)$$

a

R

OH

OCF3

 OH
 $OTBS$
 OH
 OH

Conditions: a) 92, CH₂CHCO₂CH(CF₃)₂, DMF/CH₂Cl₂; b) NaOMe, MeOH, then Dowex-50 (H *), 95%; c) Ti(O*i*-Pr)₄, *t*-BuO₂H, 4 Å MS. CH₂Cl₂, 73%; d) NaBH₄. THF/MeOH, 79%; e) TBSCl, DMAP, TEA, CH₂Cl₂, 95%; f) DBU, CCl₃CN, CH₂Cl₂, 81%.

Fig. (20).

was already published in 1994 [46]. It was shown that A (13), B (14), and D (16) are synthetically accessible from the most abundant sphingofungin C (15) [29].

3.5.1. Kobayashi (Sphingofungin B 1996 and F 1997)

Kobayashi and co-workers published the first asymmetric total synthesis of sphingofungin B (14) in 1996. The chiral Lewis-acid-controlled (CLAC) aldol addition of 2-benzyloxy-1-trimethylsiloxy-1-phenoxyethene and trimethylsilylpropynal afforded the phenyl ester 95 in a syn/anti ratio of 97:3 (87% yield). In general, it is possible to synthesise all four stereoisomers of 95 and, hence, also of 14, based on the CLAC synthesis. Reduction of the phenyl ester in 95 and protection of the resulting diol was followed by desilvlation and acetylene 96 was obtained. Compound 96 was coupled with the bromide 97 to afford ether 98 in 86% yield. An initial Sn(II)-catalysed asymmetric aldol reaction based on the CLAC methodology was also the initial step in the synthesis of 97. Ether 98 was hydrolysed and converted in 5 steps to the aldehyde 99 (63% yield from 98), which in turn underwent the key aldol reaction with the zinc enolate of 100 to give the desired adduct in 93% yield and 67% ds. Final hydrolysis, and sequential removal of the protecting groups

afforded 14 in 17 linear steps and 12% overall yield from trimethylsilylpropynal (Fig. (21)) [47].

The first total synthesis of sphingofungin F (18) in 1997 was based on an analogous methodology, but the formation of the quaternary amino acid carbon was accomplished utilising Schöllkopf's bis-lactimether method [48]. Here, the diastereoselective Sn(II)-catalysed aldol reaction of bislactimether 103, an alanine equivalent, and keto-aldehyde 102, obtained after oxidation of 99, gave the major diastereomer 104 (*dr* 70:25:5:0) in 83% yield and, eventually, 18 was obtained after a series of deprotection steps (Fig. (22)) [49].

3.5.2. Lin (Sphingofungin E 2001 and F 2000)

An efficient and convenient approach to sphingofungin F (18) started from L-(+)-tartaric acid and employed the methodology developed by Hatakeyama and co-workers used for the total synthesis of mycestericin E (6) [45]. Aldehyde 105 underwent a Wittig reaction with ylide 106 to give the α,β -unsaturated ester 107. After reduction of the methyl ester in 107, a Katsuki-Sharpless asymmetric epoxidation (AE) gave a 3:1 mixture of the desired diastereomer 108 (73% yield), which was joined with the known phosphonium salt

Conditions: a) Sn(OTf)₂, (R)-1-methyl-2-[(N-1-naphtylamino)methyl]pyrrolidine, SnO, C₂H₃CN, 87%, 91% *ee* (*syn*); b) DIBALH, CH₂Cl₂, 83%; c) DMP, PTS, DMF, 97%; d) TBAF, CH₂Cl₂, 89%; e) 97, BuLi, THF, HMPA, 87%; f) HCl, MeOH, 96%; g) MMTrCl, TEA, DMAP, CH₂Cl₂: h) LAH, THF; i) TBSCl, imid., DMF, 98% for three steps; j) HCO₂H, Et₂O, 83%; k) (COCl)₂, DMSO, TEA, CH₂Cl₂, 95%; l) LDA, 100, ZnCl₂. THF, 93%, 67% *ds*; m) (Boc)₂O, CH₂Cl₂.

Fig. (21).

Conditions: a) 103, BuLi, SnCl₂, THF, 83%; b) TBAF, THF, 94%; c) PTS, THF/H₂O; d) NaOH, MeOH/H₂O 58% for two steps; e) BCl₃, CH₂Cl₂, 7276.

Fig. (22).

49 (Fig. (11)) in another Wittig reaction to yield 109 (85%). Application of the Hatakeyama procedure gave the oxazoline 110, which was transformed, via ester 111 and photoisomerisation [35] with phenyl disulfide, to 18 in total 22 linear steps and 3.7% overall yield (Fig. (23)) [50].

The same method was used for the first total synthesis of sphingofungin E (17), which was achieved in 19 steps with 8% overall yield (Fig. (24)) [51]. Here, the known aldehyde 105 underwent a Baylis-Hillman reaction with methylacrylate to afford the methyl ester 112 in 70% yield. Dihydroxylation with substrate control and protection of the primary hydroxyl group gave the 2,3-syn-triol 113 as the sole product. Mesylation of the secondary hydroxyl group in 113 and ring closure was followed by reduction and protection to afford epoxide 114. Then, Hatakeyama's

methodology produced oxazoline 115, with all stereogenic centres of the final product, which was transformed to the key intermediate carbamate 116.

3.5.3. Trost (Sphingofungin E 2001 and F 1997)

The utilisation of *gem*-diacetates as carbonyl surrogates by Trost and co-workers presented two more asymmetric total syntheses of sphingofungins E (17) and F (18). The key step to establish two of the four stereocentres is a Pd(0)-catalysed asymmetric allylic alkylation reaction (AAA) of the *gem*-diacetate 117 with azlactone 118. The desired alkylation product 120 is obtained as the major diastereomer in a ratio of 11:1 in 70% isolated yield. Alkylated azlactone 120 was converted to the corresponding methyl ester and dihydroxylation exclusively gave the lactone 121 in 86%

BnO

CHO

a

CO2Me

b-f

OTBS

R

OTBS

$$k-q$$
 $k-q$
 $k-$

Conditions: a) 106, CH₂Cl₂ (E/Z 17:1); b) DIBALH, CH₂Cl₂, 90%; c) TBHP, L-(+)-DIPT, Ti(Oi-Pr)₄, 4 Å MS, CH₂Cl₂, 73%; d) TBSCl, TEA, DMAP, CH₂Cl₂, 80%; e) Pd(OH)₂/C, H₂, EtOAc/MeOH, 97%; f) (COCl)₂, DMSO, TEA, CH₂Cl₂, 87%; g) 49, BuLi, THF, 85%; h) TBAF, THF, 95%; i) Cl₃CCN, DBU, CH₂Cl₂; j) BF₃Et₂O, CH₂Cl₂, 80% for two steps; k) CO(OCCl₃)₂, Py, CH₂Cl₂; l) K₂CO₃, MeOH, 82% for two steps; m) PDC, DMF; n) CH₂N₂. Et₂O, 78% for two steps; o) PhSSPh, hv, cy. 1,4-dioxane, 95%; p) PTS. EtOH, 81%; q) NaOH, EtOH, 51%.

Fig. (23).

Conditions: a) methylacrylate, DABCO, 70%; b) OsO₄, NMO, acetone/ H_2O ; c) TBSCI, TEA, DMAP, 89% for two steps; d) MsCI, Py, CH_2Cl_2 ; e) K_2CO_3 , MeOH, 91% for two steps; f) DIBALH, CH_2Cl_2 , g) NaBH₄, MeOH, 90% for two steps; h) Cl_3CCN , DBU, CH_2Cl_2 ; i) El_2AICI , CH_2Cl_2 , 97% for two steps; j) $CO(OCCl_3)_2$, Py, CH_2Cl_2 ; k) K_2CO_3 , MeOH, 81% for two steps; l) MOMCI, DIPEA, CH_2Cl_2 , 87%.

Fig. (24).

isolated yield. The acetyl protected hydroxyl group in 121 was transformed into a tosylate and an intramolecular nucleophilic substitution yielded oxazoline 122 with the desired absolute configuration of the final product. Removal of the silyl protecting group, oxidation of the primary alcohol and iodomethylenation afforded the coupling partner for the Suzuki cross coupling with borane 123 and the sphingofungin derivative 124 was obtained in very good yield. The natural product 18 was obtained after deprotection and hydrolysis in 15 steps and 17% overall yield. Analogously, the synthesis of 17 was accomplished in 17 steps and 5.1% overall yield (Fig. (25)) [52].

3.5.4. Shiozaki (Sphingofungin E 2001)

Shiozaki and co-workers used a combination of known methods for a long total synthesis of sphingofungin E (17). The known benzylidene derivative 125 was oxidised and

chloromethylated to give alcohol 126, which in turn was treated according to the method of Olesker [41] to afford azide 127. The absolute configuration of the secondary hydroxyl group in 128 was inverted with a combination of oxidation and selective reduction in a ratio of >95:5 in favour of the desired azide 128. The key intermediate 130, obtained after hydrolysis and a series of deprotection steps, was transformed to the final product 17 in 29 steps and 1% overall yield with the above-mentioned procedure by Trost, consisting of iodoolefination and Suzuki-Miyaura cross-coupling [52] (Fig. (26)) [53].

3.5.5. Chida (Sphingofungin E 2002) and Ham (Sphingofungin F 2002)

Sphingofungin E (17) and F (18) were synthesised with the same methodology as used for the total synthesis of 1. Chida and co-workers synthesised 17 from D-glucose in 26

Conditions: a) 106, CH_2CI_2 (E/Z 17:1); b) DIBALH, CH_2CI_2 , 90%; c) TBHP, L-(+)-DIPT, $Ti(Oi\text{-Pr})_4$, 4 Å MS, CH_2CI_2 , 73%; d) TBSCI, TEA, DMAP, CH_2CI_2 , 80%; e) $Pd(OH)_2/C$, H_2 , EtOAc/MeOH, 97%; f) ($COCI)_2$, DMSO, TEA, CH_2CI_2 , 87%; g) 49, $BuLi_1$, THF, 85%; h) TBAF, THF, 95%; i) CI_3CCN , DBU, CH_2CI_2 ; j) BF_3 Et_2O , CH_2CI_2 , 80% for two steps; k) $CO(OCCI_3)_2$, Py, CH_2CI_2 ; l) K_2CO_3 , MeOH, 82% for two steps; m) PDC, DMF; n) CH_2N_2 , Et_2O , 78% for two steps; o) PhSSPh, hv, cy. 1,4-dioxane, 95%; p) PTS, EtOH, 81%; q) NaOH, EtOH, 51%.

Fig. (25).

Conditions: a) (COCl)₂, DMSO, TEA, CH₂Cl₂, 76%; b) LiCHCl₂. THF, 70%; c) DBU, DMSO; d) NaN₃, 15-crown-5. HMPA; e) NaBH₄. MeOH, 84% for three steps; f) TBDPSCl, imid, DMF; g) CSA, MeOH, 86% for two steps; h) TBSCl, imid, DMF; i) PMBCl, NaH, DMF, 64% for two steps; j) [Ir(COD)(PMePh)₂)PF₆, THF; k) NBS, H₂O/THF, 82% for two steps; l) Dess-Martin periodinane, CH₂Cl₂; m) H₂NMe. MeOH, 94% for two steps; n) (COCl)₂, DMSO, TEA, CH₂Cl₂; o) L-Selectride, THF, 82% for two steps; p) SEMCl., DIPEA, DCE, 85%; q) Mel, NaH, DMF, 97%; r) DDQ, H₂O, CH₂Cl₂; s) PPTS, tol, 62% for two steps; t) NaBrO₃, NaHSO₃, H₂O/EtOAC, 69%; u) Pd, H₂, EtOAc; v) PhCOCl, TEA, CH₂Cl₂, 80% for two steps; w) aq. H₂SO₄, acetone, 87%.

Fig. (26).

linear steps with the known strategy from myriocin (1) [38] with an Overman rearrangement of an allylic trichloro-acetimidate as the key step. The primary hydroxyl group of known diol 131, prepared from D-glucose in three steps, was protected and the secondary hydroxyl group oxidised to the corresponding ketone. Wittig reaction with the stabilised ylide (Ph₃P=CHCO₂Et) yielded, after reduction, a mixture of allyl alcohol 132 (E:Z 1:4). The stereoselective Overman rearrangement of the separated Z-isomer resulted in a mixture of isomers (R:S 64:14). The (R)-isomer 133 was transformed via reductive ozonolysis and subsequent reduction to the alcohol 134. Acid hydrolysis removed the MOM and the

acetonide protecting group to give furanose 135, and after another Wittig reaction with the previously used stabilised ylide (Ph₃P=CHCO₂Et) the unsaturated ester 136 was formed exclusively. The hydroxyl groups of 136 were simultaneously protected and, then, the obtained diacetonide 137, analogously to 1, transformed to the final product 17. (Fig. (27)) [54].

Ham and co-workers applied their method of Pd(0)-catalysed diastereoselective oxazoline formation used in the total synthesis of 139 to accomplish a total synthesis of 18. The additional secondary hydroxyl group was introduced via MgBr₂-promoted γ-alkoxy allylic stannane addition of 140

Conditions: a) n-Bu₂SnO, tol; b) CsF, BnBr, DMF, 88% for two steps; c) Swern oxidation; d) Ph₃P=CHCO₂Et, tol, 98% for 2 steps, E:Z=1:4; e) DIBALH, tol, 73%; f) Cl₃CCN, DBU, CH₂Cl₂; g) xylene, 140 °C, sealed tube, 64% for two steps; h) O₃, CH₂Cl₂, then Me₂S; i) Zn(BH₄)₂, Et₂O, 91% for two steps; k) aq. HCl, THF; l) Ph₃P=CHCO₂Et, CH₂Cl₂; m) Me₂C(OMe)₂, CSA, 46% for three steps.

Fig. (27).

Conditions: a) MeI, KHMDS, HMPA, THF, 73%; b) O_3 , MeOH, then Me_2S ; c) 140, MgBr₂Et₂O, CH₂Cl₂, 78% for two steps; d) O_3 , MeOH, then Me_2S ; e) CrCl₂, CHI₃, THF, 63% for two steps; f) 72, t-BuLi, then ZnCl₂, Pd(PPh₃)₄, THF, 68%, g) HCI, THF, 80%; h) NaOH, 79%.

Fig. (28).

to the aldehyde obtained from ozonolysis of 139 (Fig. (28)) [55].

CONCLUSIONS

The antifungal and immunosuppressive metabolite myriocin (1) was simultaneously discovered in 1972 by two independent research groups. The first group isolated 1 from the culture broth of the thermophilic fungus *Myriococcum albomyces* and named it myriocin while the second group found the compound in *Mycelia sterilia* and named it thermozymocidin. More recently, 1 was also isolated from the culture mash of the non-thermophilic organism *Melanconis flavovirens* and, termed as ISP-1, from the thermophilic fungus *Isaria sinclairii*. Myriocin binding

proteins were identified from mammalian homologues of two yeast proteins that have been genetically linked to sphingolipid biosynthesis. Myriocin (1) inhibits in CTLL-2-derived microsomes SPT, an enzyme which catalyses the first step of the sphingolipid biosynthesis. In horses the accumulation of free sphingoid bases (sphinganine) serum, urine, kidney, or liver caused by the mycotoxins fumonisins is suppressed by inhibition of ceramide synthase. The accumulation of sphingoid bases is correlated with outbreaks of mortal equine leukoencephalomalacia (ELEM) and other farm animal diseases.

Six total syntheses of naturally occurring (+)-myriocin (1) have been published since its discovery and characterisation. Despite the fact that 1 is a α -substituted

amino acid, most of the syntheses have used carbohydrate derivatives as chiral starting materials because of the predisposition of the carboxy and hydroxyl groups. The alternative choice of an amino acid proved to be versatile in several of syntheses, though, in these cases all stereocentres of the final product had to be established during synthesis. The two formal synthesis of 1 published so far started from D-glucose and followed similar synthetic strategies.

Mycestericin A to C (2-4), structurally and biologically closely related to 1, have been of minor synthetic interest so far, and only a few total syntheses are published. Scientific focus was mainly on the two pairs of diastereomers, mycestericins D/F (5/7) and E/G (6/8).

The very potent immunosuppressive activity of 1 - 8 *in vitro* and *in vivo* with a completely different mechanism than the commercial immunosuppressive tacrolimus (FK506) CsA led to the development of symmetrical 2-amino-1,3-propanediols and, hence, the very effective immunosuppressant FTY720 (11). Aminoalcohol 11 is currently in phase III clinical trials and is expected to become in combination with the known calcineurin-inhibitors CsA or FK506 a novel potent drug for the therapeutic areas of transplantation and autoimmunity.

From the sphingofungin family only metabolites E (17) and F (18) have the characteristic quaternary carbon present in 1 - 8. Therefore, 17 and 18 caught the main synthetic interest and a series of total synthesis with completely different methodologies have been published. In a few cases just a modified synthesis of 1 was applied to the sphingofungins.

Although much effort in the synthesis of sphingosinrelated metabolites has been done, many challenges in research still remain. The interest of the pharmaceutical industry in derivatives of theses immunosuppressive metabolites and their need for optical active compounds might lead to the development of short and highly stereoselective methods and syntheses to obtain the natural compounds and their chiral derivatives.

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REFERENCES

[2]

- [1] Bagli, JF.; Kluepfel, D. Advan. Antimicrob. Antineoplastic Chemother., Proc. Int. Congr. Chemother., 7th (1972), Meeting Date 1971, I(1), 257.
 (b) Kluepfel, D.; Bagli, J.; Baker, H.; Charest, MP.; Kudelski, A.;
 - Sehgal, S.N.; Vézina, C. J. Antibiot., 1972, 25, 109. Craveri, R.; Manachini, P.L.; Aragozzini, F. Experientia, 1972, 28,
 - (b) Aragozzini, F.; Manachini, P.L.; Craveri, R.; Rindone, B.;
 Scolastico, C. Experientia, 1972, 28, 881.
 (c) Aragozzini, F.; Manachini, P.L.; Craveri, R.; Rindone, B.;
 - (c) Aragozzini, F.; Manachini, P.L.; Craveri, R.; Kindone, B.; Scolastico, C. *Tetrahedron*, **1972**, *28*, 5493.
- [3] Bagli, JF.; Kluepfel, D.; St-Jacques, M. J. Org. Chem., 1973, 38, 1253.
- [4] Kuo, C.H.; Wendler, N.L. Tetrahedron Lett., 1978, 211.
- [5] Destro, R.; Colombo, A. J. Chem. Soc., Perkin Trans. 2, 1979, 896.
- Just, G.; Payette, D.R. Tetrahedron Lett., 1980, 21, 3219.
 (b) Payette, D.R.; Just, G. Can. J. Chem., 1981, 59, 269.

- [7] Šašek, V.; Sailer, M.; Vokoun, J.; Musílek, V. J. Basic. Microbiol., 1989, 29, 383.
- [8] Samson, R.A. Studies in Mycology, 1974, Vol 4. pp. 1-119.
- [9] Fujita, T.; Inoue, K.; Yamamoto, S.; Ikumoto, T.; Sasaki, S.; Toyoma, R.; Chiba, K.; Hoshino, Y.; Okumoto, T. J. Antibiot., 1994, 47, 208.
- [10] Kino, T.; Goto, T. Ann. N.Y. Acad. Sci., 1993, 685, 13.
 (b) Peters, D.H.; Fitton, A.; Plosker, G.L.; Faulds, D. Drugs, 1993, 46, 746.
- [11] Borel, JF. History of Cyclosporin A and Its Significance in Immunology. In *Cyclosporin A. Ed.* JF. Borel, Elsevier Biochemical Press, Amsterdam, 1982, pp 5-17.
 (b) Borel, JF. *Pharmacol. Rev.*, 1990, 41, 259.
- [12] Nakamura, S.; Kozutsumi, Y.; Sun, Y.; Miyake, Y.; Fujita, T.; Kawasaki, T. J. Biol. Chem., 1996, 271, 1255.
- [13] Chen, J.K.; Lane, W.S.; Schreiber, S.L. Chem. Biol., 1999, 6, 221.
- [14] Aragozzini, F.; Manachini, P.L.; Craveri, R.; Beretta, M.G.; Rindone, B.; Scolastico, C. Ann. Microbiol. Enzym., 1972, 22, 29.
 (b) Aragozzini, F.; Beretta, M.G.; Ricca, G.S.; Scolastico, C. Chem. Commun., 1973, 788.
- [15] Banfi, L.; Beretta, M.G.; Colombo, L.; Gennari, C.; Scolastico, C.; Aragozzini, F. Gazz. Chim. Ital., 1982, 112, 51.
- [16] Miyake, Y.; Kozutsumi, Y.; Nakamura, S.; Fujita, T.; Kawasaki, T. Biochem. Biophys. Res. Commun., 1995, 211, 396.
 (b) Hanada, K.; Nishijima, M.; Fujita, T.; Kobayashi, S. Biochem. Pharmacol., 2000, 59, 1211.
- [17] Horvath, A.; Sütterlin, C.; Manning-Krieg, U.; Movva, N.R.; Riezman, H. EMBO J., 1994. 13, 3687.
- [18] Riley. RT.; Showker, J.L.; Owens, D.L.; Ross, PF. Environ. Toxicol. Pharmacol., 1997, 3, 221.
- [19] Riley, R.T.; Voss, K.A.; Norred, W.P.; Bacon, C.W.; Meredith, F.I.: Sharma, R.P. Environ. Toxicol. Pharmacol., 1999, 7, 109.
- [20] Ross, P.F.; Rice, L.G.; Reagor, J.C.; Osweiler, G.D.; Wilson, T.M.; Nelson, H.A.; Owens, D.L.; Plattner, R.D.; Harlin, K.A.; Richard, J.L.; Colvin, B.M.; Banton, M.I. J. Vet. Diagn. Invest., 1991, 3, 238.
 - (b) Plumlee, K.H.; Galey, F.D. J. Vet. Intern. Med., 1994, 8, 49.
- [21] Sasaki, S.; Hashimoto, R.; Kiuchi, M.; Inoue, K.; Ikumoto, T.; Hirose, R.; Chiba, K.; Hoshino, Y.; Okumoto, T.; Fujita, T. J. Antibiot., 1994, 47, 420.
 (b) Fujita, T.; Hamamichi, N.; Kiuchi, M.; Matsuzaki, T.; Kitao, Y.; Inoue, K.; Hirose, R.; Yoneta, M.; Sasaki, S.; Chiba, K. J. Antibiot., 1996, 49, 846.
- [22] Fujita, T.; Inoue, K.; Yamamoto, S.; Ikumoto, T.; Sasaki, S.; Toyama, R.; Yoneta, M.; Chiba, K.; Hoshino, Y.; Okumoto, T. J. Antibiot., 1994, 47, 216.
- [23] Fujita, T.; Hirose, R.; Hamamichi, N.; Kitao, Y.; Sasaki, S.; Yoneta, M.; Chiba, K. Bioorg. Med. Chem. Lett., 1995, 5, 1857.
 (b) Fujita, T.; Yoneta, M.; Hirose, R.; Sasaki, S.; Inoue, K.; Kiuchi, M.; Hirase, S.; Adachi, K.; Arita, M.; Chiba, K. Bioorg. Med. Chem. Lett., 1995, 5, 847.
 (c) Fujita, T.; Hirose, R.; Yoneta, M.; Sasaki, S.; Inoue, K.; Kiuchi,
 - M.; Hirase, S.; Chiba, K.; Sakamoto, H.; Arita, M. J. Med. Chem., 1996, 39, 4451.

 Adachi, K.; Kohara, T.; Nakao, N.; Arita, M.; Chiba, K.; Mishina,
- [24] Adachi, K.; Kohara, T.; Nakao, N.; Arita, M.; Chiba, K.; Mishina, T.; Sasaki, S.; Fujita, T. Bioorg. Med. Chem. Lett., 1995, 5, 853.
 (b) Kiuchi, M.; Adachi, K.; Kohara, T.; Minoguchi, M.; Hanano, T.; Aoki, Y.; Mishina, T.; Arita, M.; Nakao, N.; Ohtsuki, M.; Hoshino, Y.; Teshima, K.; Chiba, K.; Sasaki, S.; Fujita, T. J. Med. Chem., 2000, 43, 2946.
- [25] Brinkmann, V.; Pinschewer, D.D.; Feng, L.; Chen, S. Transplantation, 2001, 72, 764.
- [26] Brinkmann, V.; Lynch, K.R. Current Opinion in Transplantation, 2002. 14, 569.
- [27] VanMiddlesworth, F.; Giacobbe, R.A.; Lopez, M.; Garrity, G.: Bland, J.A.; Bartizal, K.; Fromtling, R.A.; Polishook, J.; Zweerink, M.; Edison, A.M.; Rozdilsky, W.; Wilson, K. E.; Monaghan, R.L. J. Antibiot., 1992, 45, 861.
- [28] Horn, W.S.; Smith, J.L.; Bills, G.F.; Raghoobar, S.L.; Helms, G.L.;

- Kurtz, M.B.; Marrinan, J.A.; Frommer, B.R.; Thornton, R.A.; Mandala, S.M. J. Antibiot., 1992, 45, 1692.
- [29] VanMiddlesworth, F.; Dufresne, C.; Wincott, F.E.; Mosley, R.T.; Wilson, K.E. Tetrahedron Lett., 1992, 33, 297.
- [30] Chida, N.; Ikemoto, H.; Noguchi, A.; Amano, S.; Ogawa, S. Nat. Prod. Lett., 1995, 6, 295.
- [31] Zweerink, M.M.; Edison, A.M.; Wells, G.B.; Pinto, W.; Lester, R.L. J. Biol. Chem., 1992, 267, 25032.
- [32] Sailer, M.; Šašek, V.; Sejbak, J.; Bud čšínký, M.; Musílek, V. J. Basic Microbiol., 1989, 29, 375.
- [33] Berova, N., Breinholt, J.; Jensen, G.W.; Kjær, A.; Lo, L.C.; Nakanishi, K.; Nielsen, R.I.; Olsen, C.E.; Pedersen, C.; Stidsen, C.E. Acta Chim. Scand., 1994, 48, 240.
- [34] Banfi, L.; Beretta, M.G.; Colombo, L.; Gennari, C.; Scolastico, C. J. Chem. Soc. Chem. Commun., 1982, 488.
 (b) Banfi, L.; Beretta, M.G.; Colombo, L.; Gennari, C.; Scolastico, C. J. Chem. Soc. Perkin Trans. 1, 1983, 1613.
- [35] Yoshikawa, M.; Yokokawa, Y.; Okuno, Y.; Murakami, N. Chem. Pharm. Bull., 1994, 42, 994.
 (b) Yoshikawa, M.; Yokokawa, Y.; Okuno, Y.; Murakami, N. Tetrahedron, 1995, 51, 6209.
- [36] Sano, S.; Kobayashi, Y.; Kondo, T.; Takebayashi, M.; Maruyama, S.; Fujita, T.; Nagao, Y. Tetrahedron Lett., 1995, 36, 2097.
- [37] Hatakeyama, S.; Yoshida, M.; Esumi, T.; Iwabuchi, Y.; Irie, H.; Kawamoto, T.; Yamada, T.; Nishizawa, M. *Tetrahedron Lett.*, 1997, 38, 7887.
- [38] Oishi, T.; Ando, K.; Chida, N. Chem. Commun., 2001, 1932.
 (b) Oishi, T.; Ando, K.; Inomiya, K.; Sato H.; Iida, M.; Chida, N. Bull. Chem. Soc. Jpn., 2002, 75, 1927.
- [39] Lee, K.Y.; Oh, C.Y.; Kim, Y.H.; Joo, J.E.; Ham, W.H. Tetrahedron Lett., 2002, 43, 9361.
- [40] Rao, A.V.R.; Gurjar, M.K.; Devi, T.R.; Kumar, K.R. Tetrahedron Lett., 1993, 34, 1653.
- [41] Deloisy, S.; Thang, T.T.; Olesker, A.; Lukacs, G. Tetrahedron Lett., 1994, 35, 4783.

- (b) Deloisy, S.; Thang, T.T.; Olesker, A.; Lukacs, G. Bull. Soc. Chim., 1996, 133, 581.
- [42] Shibata, K.; Shingu, K.; Vassilev, V.P.; Nishide, K.; Fujita, T.; Node, M.; Kajimoto, T.; Wong, C.H. Tetrahedron Lett., 1996, 37, 2791.
 (b) Nishida K.; Shibata K.; Fujita T.; Kajimata T.; Wang, C.H.;
 - (b) Nishide, K.; Shibata, K.; Fujita, T.; Kajimoto, T.; Wong, C.H.; Node, M. *Heterocycles*, **2000**, *52*, 1191.
- [43] Seebach, D.; Sting, A.R.; Hoffmann, M. Angew. Chem. Int. Ed. Engl., 1996, 35, 2708.
- [44] Fujita, T.; Hamamichi, N.; Matsuzaki, T.; Kitao, Y.; Kiuchi, M.; Node, M.; Hirose, R. *Tetrahedron Lett.*, **1995**, *36*, 8599.
- [45] Iwabuchi, Y.; Furukawa, M.; Esumi, T.; Hatakeyama, S. Chem. Commun., 2001, 2030.
- [46] Formal Syntheses for Sphingofungin D: (a) Mori, K.; Otaka, K. *Tetrahedron Lett.*, 1994, 35, 9207.
 (b) see Ref. [30].
 (c) Otaka, K.; Mori, K. Eur. J. Org. Chem., 1999, 1795.
- [47] Kobayashi, S.; Hayashi, T.; Iwamoto, S.: Furuta, T.; Matsumura, M. Synlett, 1996, 672.
- [48] Schöllkopf, U. Pure Appl. Chem., 1983, 55, 1799.
 (b) Schöllkopf, U., Hartwig, W.; Groth, U.; Westphalen, K. Liebigs Ann. Chem., 1981, 696.
- [49] Kobayashi, S.; Matsumura, M.; Furuta, T.; Hayashi, T.; Iwamoto,
 S. Synlett, 1997, 301. Kobayashi, S.; Furuta, T.; Hayashi, T.;
 Nishijima, M.; Hanada, K. J. Am. Chem. Soc., 1998, 120, 908.
 (d) Kobayashi, S.; Furuta, T. Tetrahedron, 1998, 54, 10275.
- [50] Liu, D.G.; Wang, B.; Lin, G.Q. J. Org. Chem., 2000, 65, 9114.
- [51] Wang, B.; Yu, X.M.; Lin, G.Q. Synlett., 2001, 904.
- [52] Trost, B.M.; Lee, C.B. J. Am. Chem. Soc., 1998, 120, 6818.
 (b) Trost, B.M.; Lee C.B. J. Am. Chem. Soc., 2001, 123, 12191.
- [53] Nakamura, T.; Shiozaki, M. Tetrahedron Lett., 2001, 42, 2701.
 (b) Nakamura, T.; Shiozaki, M. Tetrahedron, 2002, 58, 8779.
- [54] Oishi, T.; Ando, K.; Inomiya, K.; Sato, H.; Iida, M.; Chida, N. Org. Lett., 2002, 4, 151.
- [55] Lee, K.Y.; Oh, C.Y.; Ham, W.H. Org. Lett., 2002, 4, 4403.