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Effect of negative valve overlap in a heavy-duty methanol-diesel dual-fuel engine: A pathway to improve efficiency

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- EGR
- High efficiency

ABSTRACT

Methanol (MeOH) is a promising low-carbon liquid fuel to provide global energy security with a potential to achieve net-zero greenhouse gas emissions in transport sector. However, its utilization in diesel engines at high MeOH substitution ratios (MSR) suffers from misfire or high pressure rise rates owing to its distinct physio-chemical properties. This issue is addressed in the present study by adopting negative-valve overlap (NVO) and hot residual gases from the previous cycle. Experiments are performed in a single-cylinder heavy-duty CI engine for a constant MSR (90% energy based) and an engine speed of 1500 rpm. The aim of the study is to investigate the effects of 1) NVO period, 2) charge-air temperature (Tair), 3) MeOH lambda (λMeOH) on the MeOH-diesel dual-fuel (DF) combustion in NVO mode, and 4) to demonstrate the implications of NVO in yielding high net-indicated efficiency (ηind) together with low pollutant emissions at a wide range of engine operating loads (40–90%). The results show that the hot residual gases from the previous cycle enhance the reactivity of the fresh MeOH-air mixture by inducing slow oxidation processes before TDC. The slow pre-flame oxidation processes are disruptive or oscillatory in nature, wherein NVO period, Tair and λMeOH can be used to control these processes and their induced reactivity enhancing capability. It is noticed that the pre-flame oxidation processes and the main combustion have a direct correlation between them. Based on the control strategy, the MeOH-diesel combustion in the NVO mode produced on average ηind of approx. 53% accompanied with very low NOx emission of 1.1 g/kWh at a wide range of engine operating loads (40–90%). Additionally, on average the combustion phasing (CA50) is maintained at ~ 2 °CA aTDC, while the combustion stability remains high (COVIMEP ~ 3.5%).

1. Introduction

EU’s commitment to the global climate action under Paris agreement has led the governments to adopt ambitious targets to achieve net-zero greenhouse gas emissions by 2050 [1,2]. In this effort, where electrification of light-duty passenger vehicles is gaining its momentum [3,4], the plans of utilizing alternative- and renewable fuels in heavy-duty, marine, and power generation sectors with improved efficiency are also getting attention [5,6]. Methanol (MeOH), among many alternative fuels, is the simplest low-carbon and a clean burning fuel [7,8]. It can potentially contribute to achieve global net-zero emission targets when produced from biomass and the hydrogenation of atmospheric carbon dioxide (CO2) using renewable energy sources [9,10]. Additionally, its single molecule morphology as a hydrogen carrier and the characteristic state of being liquid at standard conditions makes it a suitable transport fuel solution for storing and fuel transport infrastructure with minimal losses [9,10]. However, the efficient burning of MeOH in internal combustion engines (ICE) is one of the main challenges besides renewable MeOH supply that curbs its applicability. It is, therefore, important to further investigate and develop MeOH ICEs for improved efficiency and fuel consumption.

In the past, MeOH has been extensively studied in ICEs as a pure or blended fuel by many researchers [11–14]. In SI engines (dedicated MeOH or flex fueled M85 engines), it has been demonstrated to increase engine efficiency and power by 7–10% with lower unburned hydrocarbons (THC) and nitrogen oxides (NOx) compared to counterpart gasoline engines [7,15]. MeOH leverages the benefits of octane boosting in SI engines, which enables the use of relatively high compression ratios and extends the limits of high-load engine operations. This allows the SI MeOH engines to achieve brake thermal efficiency (BTE) comparable to conventional diesel engines (~40%) [7,16]. However, the utilization of MeOH in diesel engines brings a distinctive set of challenges, which include extremely low cetane number and high heat of vaporization. In general, high-octane number indicates low cetane number. The cetane number of MeOH lies in the range of 2–3 [7,8,17], which makes its
autoignition difficult, particularly, when cylinder temperatures are also dropped due to latent heat of vaporization. In this scenario, combustion of MeOH in diesel engines requires either dedicated engines with high compression ratios [18] or advanced combustion technologies for existing diesel engines. Furthermore, there is more interest of improving diesel engines because heavy-duty applications typically employ diesel engines.

Dual-fuel (DF) is an advanced combustion concept, which uses two fuels of different reactivities [19-21]. Typical examples of DF engines are already available in the market in which the DF concept is realized either by adopting port-fuel injection (PFI), direct injection (DI) or reactivity-controlled compression ignition (RCCI) combustion technologies. In MeOH-diesel DF combustion, MeOH is ignited by pilot diesel close to the top-dead center. Wang Q. et al. [22] investigated MeOH-diesel PFI-DF combustion at varying engine loads (6% to 100%) with various MeOH substitution ratios (MSR). Only at medium loads, they could use a maximum MSR of 76%. They reported a restriction in using such high MSR at low or high loads due to partial burn, misfire, or knocking combustion. They concluded that the maximum employable MSR is quite load dependent. On the other hand, in all their experiments, they showed BTE below 38%. In a later publication, Wang B. et al. [23] attempted to further explore high-load (75% and 100%) engine operations by employing exhaust-gas recirculation (EGR) and controlled pilot injection timings at various MSR. They utilized maximum MSR of 80%. The results depicted a misfire occurrence at 1660 rpm when using MSR beyond 80% with 17% EGR. However, they reported a significant reduction in NOx and soot emissions with high MSR, nonetheless, also with high emissions of THC and carbon monoxide (CO). Similarly, Tutak et al. [24] investigated MeOH-diesel PFI-DF combustion at varying loads with various MSR, where they could employ maximum MSR of 90% only at full load (24 kW). They also reported a reduction in NOx and soot emissions, however, with an increase in THC and CO at increasing MSR. They reported BTE below 35% where at part loads (16 kW and 8 kW), combustion efficiency was demonstrated to drop drastically with increasing MSR.

RCCI-DF combustion (either PFI or DI mode of MeOH) allows an explicit control of mixture’s in-cylinder reactivity stratification by adopting various fuel injection strategies. This is generally considered an advantageous procedure for improving combustion efficiency together with low NOx and soot emissions [25-28]. Duraisamy et al. [29] studied single pilot (SOI = –10 CA ATDC) MeOH-diesel PFI-RCCI combustion to explore the effects of MSR and hot/cold EGR. They used two engine loads with maximum 90% MSR only at high load. They summarized the study with similar conclusions as above. That is, with increasing MSR, NOx and soot emissions reduce with increasing emissions of THC and CO. Also, cooled EGR is more beneficial than hot EGR for improving efficiency. Nevertheless, in their experiments, they reported maximum BTE < 35%. Jia et al. [30] performed a comparative RCCI (split diesel injection) study at varying loads with 50–76% MSR for three different MeOH injection strategies i.e., PFI, DI during the intake stroke, and DI during the compression stroke. They also employed EGR with optimized injection timings to control combustion for the best possible efficiency gains. In their experiments, a maximum indicated efficiency < 48% is reported. However, they concluded that the PFI strategy is more efficient than the DI strategies, which also maintains ultra-low NOx and soot at high loads.

Recently, Huang et al. [31] investigated the same RCCI-DI MeOH strategy with split diesel injection for varying MSR in a range of 30–60%. They called this mode as intelligent-charge compression ignition (ICCI) combustion whereby they realized high efficiency and low emissions. All presented results at IMEP = 8 bar showed indicated efficiency below 48% together with low NOx emissions, however, with rather high THC and CO emissions. Nevertheless, they reported high indicated efficiency of 53.5% only at IMEP = 12 bar using 50% MSR. Apart from this, Dong et al. [32] studied non-premixed DI-DF combustion at varying loads with various MSR in the range of 40–95%. They reported an improved indicated efficiency and combustion stability with increasing load and MSR together with decreased CO and THC emissions. However, in this study, the maximum indicated efficiency was attained < 45% for a full range of experiments. Also, the burning of 100% MeOH in DI-PPC mode produced indicated efficiency below 45% [33].

Based on the above literature and other review articles [7,8,34], it is apparent that using high MSR is more desirable to achieve net-zero emission targets at a wide range of engine operations. However, the phenomenon like misfire or high pressure-rise rates limits its applicability. Moreover, high MSRs are suggested to be favorable only for optimized high-load engine operations with maximum achievable indicated efficiency < 50%. Therefore, in this study, we attempt to investigate diesel-methanol PFI-DF combustion in a single-cylinder heavy-duty engine using a constant MSR of 90%. For this purpose, we utilize an internal EGR (iEGR) strategy with its counteracting effects of high temperature and dilution using negative valve overlap (NVO).

NVO is a well-known variable-valve actuation (VVA) strategy for improving lean operations of a single fuel type HCCI combustion concepts [35,36]. However, it has been scarcely reported for DF combustion. Only few studies [37-39] have used NVO in DF concept with an aim to improve low-load engine operations by leveraging fuel reforming [37,38] with an additional fuel injection at the end of the exhaust stroke. In this study, we address this significant research gap by investigating the PFI-DF combustion in NVO mode, however, without any additional
fuel injection. In addition, we employ a one-dimensional (1D) GT-power based engine model to estimate iEGR percentage and in-cylinder mixture temperatures. Thus, this study aims to:

1) fundamentally investigate the MeOH-diesel DF combustion in NVO mode at varying a) NVO period, b) charge-air temperature, and c) MeOH lambda,

2) demonstrate the ability of the DF combustion in NVO mode to achieve high net-indicated efficiency ($\eta_{\text{ind}} \sim 50\%$), together with ultra-low exhaust emissions at varying engine loads (40–90%) with a constant MSR of 90%.

2. Experimental setup and methods

2.1. Research engine setup

Like our previous studies [20,40], a compression ignition (CI) single-cylinder heavy-duty engine is used here as a DF research engine. Fig. 1 shows a schematic of the laboratory engine setup. The specifications of the research engine are outlined in Table 1. The engine is equipped with necessary auxiliary systems and several monitoring devices to flexibly control engine parameters. A 45 kW process performance electric motor coupled with a frequency converter handles the load and desired engine speed. Of special importance to the present study, an electro-hydraulic-valve-actuation (EHVA) [41] hardware operates the valve train with variable-valve timings and lifts. Hence, with EHVA, various VVA strategies including NVO can be controlled in a flexible manner while running the engine.

A PID-based charge-air system supply fresh air into the engine. It consists of a compressor, air-heater, control valve, and RHM-08 Coriolis mass-flow meter (Rheonik Messtechnik GmbH, Accuracy < 0.1% of flow [40]) controlling the desired pressure, temperature, and mass-flow rate of the fresh air. Similarly, a port-fuel injection system consisting of a low-pressure pump, fuel-cooling system, weighing scale (Accuracy < 0.5 mg) and a port-fuel injector (Bosch, GmbH) delivers a desired mass of MeOH into the cylinder at a desired injection timing. On the other hand, a piezo injector directly delivers diesel pilot into the cylinder with an ability to control injection timing and duration [20,40]. In addition, the engine is equipped with an in-cylinder pressure sensor (Kistler 6125C) and a charge amplifier (Kistler 5011b) to measure pressure in-side the cylinder at a resolution of 0.2 crank angles.

The engine is provided with exhaust-gas analyzing system containing various emission analyzers for measuring THC, CO, CO$_2$, oxygen (O$_2$) and NO, NO$_2$, NOx. A suction pump extracts a sample of raw emission gases from the exhaust runner via a heated probe and delivers it to the analyzers. The details of the analyzers are summarized in Table 2 [40]. An engine control unit monitors all devices and auxiliary systems, which also enables a high-speed data acquisition at 40 MHz and a flexible control of engine operating parameters.

2.2. Test fuels

In the present study, 99.9% pure MeOH and commercial diesel complying with the European EN590 standard are employed as the test fuels. MeOH is utilized as the main fuel, which is a simple low-carbon

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**Table 1**

<table>
<thead>
<tr>
<th>Engine type</th>
<th>4 Stroke, modified single-cylinder DF engine</th>
</tr>
</thead>
<tbody>
<tr>
<td>Piston geometry</td>
<td>Re-entrant bowl</td>
</tr>
<tr>
<td>Displacement volume</td>
<td>1.4 L</td>
</tr>
<tr>
<td>Bore × Stroke</td>
<td>111 × 145 mm</td>
</tr>
<tr>
<td>Geo. compression ratio</td>
<td>16.7:1</td>
</tr>
<tr>
<td>No. of valves</td>
<td>4</td>
</tr>
<tr>
<td>In-cylinder swirl ratio</td>
<td>2.7 [42]</td>
</tr>
<tr>
<td>Injector no. of holes × diameter</td>
<td>3 × 0.160 mm (symmetric)</td>
</tr>
<tr>
<td>Firing TDC (TDCf)</td>
<td>0 °CA aTDC</td>
</tr>
<tr>
<td>Gas exchange TDC (TDCex)</td>
<td>360 °CA aTDC</td>
</tr>
</tbody>
</table>

**Table 2**

<table>
<thead>
<tr>
<th>Emission type</th>
<th>Analyzer and measuring principle</th>
<th>Measuring range</th>
<th>Measurement Uncertainty</th>
</tr>
</thead>
<tbody>
<tr>
<td>THC</td>
<td>A heated-flame ionization detection (HFID) based analyzer Model VE-7 (J.U.M Engineering GmbH)</td>
<td>0–3000 ppm</td>
<td>&lt;30 ppm (&lt;45 ppm )</td>
</tr>
<tr>
<td>CO, CO$_2$</td>
<td>SIDOR non-dispersive infrared spectroscopy (NDIR) based analyzer (Sick AG)</td>
<td>CO: 0 – 3000 ppm, CO$_2$: 0 – 30 vol%</td>
<td>CO: ±3 ppm, CO$_2$: &lt;0.05 vol%</td>
</tr>
<tr>
<td>O$_2$</td>
<td>A paramagnetic dumbbell (OXOR-P) analyzer (Sick AG)</td>
<td>0–25 vol%</td>
<td>&lt;0.05 vol%</td>
</tr>
<tr>
<td>NO, NO$_2$, NOx</td>
<td>CLD-822Sh (ECO Physics AG) analyzer based on chemiluminescence</td>
<td>NO, NO$_2$: 0 – 2000 ppm</td>
<td>&lt;20 ppm</td>
</tr>
</tbody>
</table>

* for non-oxygen species.
* for oxygen containing species.
liquid fuel with a boiling point of 65 °C. However, it’s high latent heat of vaporization establishing a challenge for a reliable combustion in CI engines. Additionally, it’s extremely low cetane number and high auto-ignition temperature add up a difficulty to its ignition. Therefore, diesel is utilized as pilot to ignite the MeOH-air mixture. In addition, charge-air energy (TFE) is supplied by MeOH, whereas only 10% (TFE based) diesel is utilized as pilot to ignite the MeOH-air mixture. In addition, charge-air mass flow \((\dot{m}_{air})\), temperature \((T_{air})\), and boost pressure \((P_{int})\) are adjusted according to the experimental test matrix, as described in Section 2.4. Furthermore, as a pre-engine-run condition, the engine cooling water is heated up to 70 °C via external water heating system to evenly heat up the engine block, cylinder head and the liner. The test fuels are injected only after achieving the stable initial operating conditions.

After injecting test fuels, the engine is continuously run for 5 min in order to record the test data reliably including the averaged exhaust emissions data. Total hydrocarbons (THC) including oxygenated components (e.g., derivatives of aldehydes, mainly formaldehyde) produced from incomplete combustion of MeOH are measured by flame-ionization detection (FID) based analyzer. It should be noted that FID analyzer is calibrated with propane (non-oxygenated fuel), which may induce uncertainty in the measurements. Additionally, FID technique offers lower sensitivity for oxygenated hydrocarbons compared to non-oxygenated hydrocarbons [46]. Regardless, this systematic uncertainty in

### Table 3

<table>
<thead>
<tr>
<th>Properties</th>
<th>Diesel (EN590)</th>
<th>Methanol [44,45]</th>
</tr>
</thead>
<tbody>
<tr>
<td>Molecular formula</td>
<td>C_{12}-C_{20}</td>
<td>CH_{3}OH (MeOH)</td>
</tr>
<tr>
<td>Cetane number</td>
<td>≥ 51.0</td>
<td>2.3</td>
</tr>
<tr>
<td>Density [kg/m³]</td>
<td>820-845</td>
<td>791</td>
</tr>
<tr>
<td>Viscosity [mm²/s]</td>
<td>2-4.5</td>
<td>0.56</td>
</tr>
<tr>
<td>Boiling point [°C]</td>
<td>200-360</td>
<td>65</td>
</tr>
<tr>
<td>Lower heating value [MJ/kg]</td>
<td>43.2</td>
<td>19.9</td>
</tr>
<tr>
<td>Energy density [MJ/m³]</td>
<td>35260-36335</td>
<td>15503</td>
</tr>
<tr>
<td>Standard heat of vaporization [kJ/kg]</td>
<td>225-280</td>
<td>1165</td>
</tr>
<tr>
<td>Auto-ignition temperature [°C]</td>
<td>230</td>
<td>464</td>
</tr>
<tr>
<td>Heat capacity ratio ((γ = c_p/c_v)) at 77 °C</td>
<td>–</td>
<td>1.203</td>
</tr>
<tr>
<td>C-content [wt. %]</td>
<td>86.5</td>
<td>37.48</td>
</tr>
<tr>
<td>H-content [wt. %]</td>
<td>13.5</td>
<td>12.58</td>
</tr>
<tr>
<td>O-content [wt. %]</td>
<td>0</td>
<td>49.93</td>
</tr>
<tr>
<td>Water content [wt. %]</td>
<td>-0.02</td>
<td>&lt;0.01</td>
</tr>
<tr>
<td>Air-to-fuel ratio</td>
<td>14.7</td>
<td>6.47</td>
</tr>
</tbody>
</table>

\(\gamma\) is utilized as pilot to ignite the premixed MeOH-air mixture close to the top-dead center (TDC). The properties of the test fuels are listed in Table 3.

#### 2.3. Operating conditions and procedure

In this study, experiments are carried out at a constant engine speed of 1500 rpm in the NVO mode. Fig. 2 illustrates the concept of NVO using a representative cylinder pressure, intake, and exhaust valves. The NVO is attained by closing the exhaust valve earlier and opening the intake valve later than the gas-exchange TDC (TDC_{ex}). Thus, NVO represents a crank-angle period between exhaust-valve closing (EVC) and intake-valve opening (IVO) timings. Table 4 outlines the valve timings and other constant operating conditions. With NVO, a fraction of exhaust gases is trapped inside the cylinder, which is referred as residual gases. These gases undergo a re-compression and -expansion process around the TDC_{ex} (see Fig. 2). Later during the intake stroke, the intake valve opens at 420 °CA aTDC delayed enough to avoid any backpressure into the intake port. Meanwhile, methanol is port injected at SOI_{MeOH} = 420 °CA aTDC and supplied into the cylinder along with fresh air as soon as the intake valve opens. The fresh MeOH-air charge may experience pre-combustion processes during the early cycle due to the presence of high-temperature residual gases from the previous cycle. Later at the end of the compression stroke, diesel pilot is injected at SOIpilot = -6 °CA aTDC to ignite the premixed MeOH-air mixture close to TDCf.

It should be noted that in this study the period of NVO is varied by sweeping the EVC timing. As EVC timing advances with respect to TDC_{ex}, the period length increases, and vice versa. The employed EVC sweeps along with representative EVC are presented in Fig. 2. On the other hand, during experiments a constant fraction i.e., 90% of total-fuel energy (TFE) is supplied by MeOH, whereas only 10% (TFE based) diesel fuel is utilized as pilot to ignite the MeOH-air mixture. In addition, charge-air mass flow \((\dot{m}_{air})\), temperature \((T_{air})\), and boost pressure \((P_{int})\) are adjusted according to the experimental test matrix, as described in Section 2.4. Furthermore, as a pre-engine-run condition, the engine cooling water is heated up to 70 °C via external water heating system to evenly heat up the engine block, cylinder head and the liner. The test fuels are injected only after achieving the stable initial operating conditions.

Fig. 2. A schematic concept description of negative valve overlap (NVO) and sweeps of exhaust valve closing (EVO) timing. NVO represents a crank-angle period between EVC and IVO. For a comparison, the baseline standard valve timing (SVT) is presented in Fig. A1 in Appendix A.
...calculations using Equations 1 and 2. Additionally, net- and gross-indicated mean effective pressures (nIMEP, gIMEP) are calculated for intervals −180 to 540 and −180 to 180, respectively [20,40]. The pumping mean effective pressure (PMEP) is calculated as a difference between nIMEP and gIMEP. Here, combustion stability is determined by the coefficient of variability of gIMEP (COV_{IMEP}), which is estimated using the standard deviation method. Other important performance parameters such as MeOH lambda (λ_{MeOH}), total-fuel energy (TFE), combustion efficiency (η_{comb}), net-induced efficiency (η_{ind}) and thermal efficiency (η_{th}) are calculated according to Equations (3) to (7).

\[
HRR = \frac{dQ}{d\theta} = \frac{\gamma}{\gamma - 1} P \frac{dV}{d\theta} + \frac{1}{\gamma - 1} V \frac{dP}{d\theta}
\]

(1)

\[
AccQ = \int \frac{dQ}{d\theta} \, d\theta
\]

(2)

\[
λ_{MeOH} = \frac{\rho_{air}/ρ_{MeOH}}{(AFR_{MeOH})_{stoichiometric}}
\]

(3)

\[
TFE = LHV_{MeOH} \times \rho_{MeOH} + LHV_{diesel} \times \rho_{diesel}
\]

(4)

\[
η_{comb} = \frac{\eta_{ind}}{\frac{T_{out}}{T_{in}} \times 100}
\]

(5)

\[
η_{ind} = \frac{nIMEP \times V_{displacement}}{TFE} \times 100
\]

(6)

\[
η_{th} = \frac{gIMEP \times V_{displacement}}{η_{comb} \times TFE} \times 100
\]

(7)

2.3.1. 1D GT power simulations

A 1D engine model based on GT-power by Gamma technologies is used here to mainly estimate the amount of trapped residual gases and mean mixture temperatures for the DF combustion in NVO mode. First, an engine model was developed for the single-cylinder research engine...
with all possible components including piping, tanks, valves, and ports. Fig. 3(A) shows the validation of the engine model against motored (no fuel) cylinder pressure at the measured initial conditions. This represents an adequately calibrated engine model for the intake and exhaust gas dynamics.

Later, the model was further developed based on the full cycle, Three Pressure Analysis (TPA) [49] method and introduced with DF fuel injection systems. The TPA method uses the measured cylinder pressure data to calculate apparent burn rate and imposes it to the simulation cycle for the combustion prediction. TPA is an iterative method until the simulated cylinder pressure matches the measured one. Fig. 3(B) shows the validation of the TPA method with the experimental cylinder pressure data of MeOH-diesel DF combustion. It should be noted that the engine conditions for Fig. 3(A) and (B) are the same. On the other hand, measured emissions of THC (C1 carbon based), CO and NO are imposed in the combustion model to estimate $\eta_{\text{comb}}$. However, 1D model-based $\eta_{\text{comb}}$ is used only for validation purposes, where we observed an error of <2% for all studied cases. Furthermore, the percentage of iEGR is defined as the ratio of the trapped mass of residuals ($m_{\text{residual}}$) to the total mass of mixture inside the cylinder ($m_{\text{total}}$), according to Equation 8.

$$\text{iEGR} = \frac{m_{\text{residual}}}{m_{\text{total}}} \times 100 \quad (8)$$

2.4. Test matrix

This study aims to investigate the characteristics of MeOH-diesel DF combustion in the NVO mode, and to demonstrate the ability to control a wide range of engine operating loads with an adequate combustion stability, improved emissions, and positive efficiency gains. However, first we compare NVO with standard-valve timing (SVT) briefly to emphasize the benefits of using NVO mode. Then, to proceed with the NVO mode investigation, the experimental test matrix is categorized in four test cases as, 1) effect of negative-valve overlap (NVO) period, 2) effect of charge-air temperature (CAT), 3) effect to methanol lambda (ML), and 4) effect of varying engine load (EL). The experimental test points and corresponding performance parameters are summarized in Table 5. Here, all experiments are performed under design-of-experiment approach to achieve $\eta_{\text{comb}} \geq 90\%$.

In each test case, the corresponding test points are identified from their characteristic name. For example, NVO120 indicates a test point with NVO period of length 120 oCA and so on. Similarly, CAT40, ML32, and EL40 represent the test points of charge-air temperature of 40 °C, methanol lambda of 3.2, and engine load of 40%, respectively. It should be noted that here engine load is defined as a ratio of 90% of the TFE to the maximum rated engine power per cylinder. The test cases #1–3 are performed for constant operating parameters except for the parameter under investigation. However, in the test case of engine load variations, all parameters are adjusted to achieve engine performance results above a set threshold. These thresholds include 1) $\text{CA50} \approx 2$ oCA aTDC, 2) $\eta_{\text{comb}} \geq 90\%$, 3) $\eta_{\text{ind}} \sim 50\%$, and 4) COVIMEP < 5%.

3. Results and discussion

3.1. Comparison of NVO and SVT

Prior to the detailed investigation, MeOH-diesel DF combustion is compared here using two different valve configurations at constant operating conditions, as listed in Table 5. $\text{SOL}_{\text{MeOH}}$ is adjusted according to the corresponding IVO timing. The compared NVO and SVT configurations, labelled as representative intake and exhaust valves, are illustrated in Fig. 2 and Fig. A1, respectively. The obtained results of cylinder pressure and HRR are presented in Fig. 4. The combustion with NVO is observed to produce significantly higher cylinder pressure and HRR compared to SVT. Also, NVO assists producing higher $\eta_{\text{comb}}$ of 93% compared to 33% as produced with SVT. Despite the constant conditions, the reason for attaining such high efficiency with NVO may be associated with implicitly higher compression pressure, temperature and internal EGR (iEGR). Therefore, for validation, additional experiments were performed to investigate the combustion with SVT at increased boost pressure ($P_{\text{air}} \sim 2$ bar) and charge-air temperature ($T_{\text{air}} \sim 150$ °C), while keeping other conditions constant. For brevity, the
results are not presented here, nevertheless, the $\eta_{comb}$ was observed to improve to approx. 80%. However, the early start of combustion and its phasing were unrealistic. The use of cold EGR could be advantageous for controlling the phasing of the start of combustion.

In an attempt to improve the combustion with SVT, it is also found that SOI$_{pilot}$ earlier than $12^\circ$CA aTDC (RCCI mode at 90% MeOH) is unfavorable at such high temperatures. Moreover, the implementation of the split injection strategy is restricted due to the small pilot quantity (10%) and the operational limits of the injector. In brief, compared to NVO, the combustion with SVT requires drastically high charge-air temperatures and a complex control of parameters to reliably attain $\eta_{comb} \geq 90%$. Therefore, in this study, 10/90 diesel-methanol DF combustion is investigated and controlled at varying loads using NVO method. In addition, in this way part of waste heat from exhaust gases can be recovered.

Here, it should be noted that in this study NVO method is used as a combination of early EVC together with late IVO timing, which is considerably more efficient VVA strategy compared to using early EVC timing alone. With late IVO, most of the work spent on the re-compression of the residual gases can be recovered during the re-expansion process, while only some of the work is lost (pumping loss) as heat transfer to the cylinder walls. In this study, moderate levels of boost pressures (1.5 – 2.0 bar) are also utilized, which compensates the lost work and influences the nIMEP positively, as shown in Table 5. More details about the potentials of NVO compared to various VVA strategies (including PVO) can be found in a publication by Rodriguez J.F. et al. [50].

### 3.2. Effect of NVO period

The period of NVO regulates the amount of residual gases trapped inside the cylinder. In this study, it is controlled by varying EVC timing, as described in Section 2.3. To understand the effect of trapped residual gases, the combustion is investigated at four various NVO periods under constant operating conditions, as outlined in Table 5. The obtained results of cylinder pressure and HRR for test points NVO120-NVO135 are presented in Fig. 5. The combustion improves with an increase in the residual gas amount. It can be observed that the peak cylinder pressure and peak HRR increases as NVO period prolongs from NVO120 to NVO135. A greater amount of hot-residual gases retains inside the cylinder when the NVO period is extended. Consequently, the retained residual-gas mass enhances the in-cylinder thermodynamic state. Fig. 6 presents mean mixture temperatures for the studied test points, as estimated using the GT-power based TPA method explained in Section 2.3.1.

It can be observed that trapped mass of residual gases has a pronounced effect on the simulated mean mixture temperatures. In the re-compression and -expansion process around TDC$_{ex}$, the peak temperature increases by ~ 40 ºC between NVO120 and NVO135, which is roughly a gain of ~ 5%. This thermal gain nevertheless sustains also during the intake stroke, although the temperature ranges are lower. Interestingly, at IVC timing, the simulated mean mixture temperature retains at ~ 190 ºC despite the latent heat absorbed by MeOH during the early intake stroke (after IVO). It is estimated that at IVO timing, fresh MeOH is injected into bulk ambient temperature of ~ 300–320 ºC, which gradually drops to ~ 185–195 ºC (at IVC). Here, it is interesting to note that such a thermodynamic state facilitates the thermal cracking or slow partial oxidation of the fresh MeOH in the presence of residual gases, which may lead to the production of reactive species and thus to earlier autoignition of the in-cylinder mixture.

Aniokhe et al. [51] investigated such slow oxidation of methanol at similar temperature and pressure conditions in a constant volume stirred reactor. They reported that HO$_2$ is the active specie that leads to CH$_3$OH → CH$_2$OH → CH$_2$O → CHO → CO as a result of partial oxidation of MeOH in the pre-flame regime. However, they also reported a heterogeneous termination of HO$_2$ chain branching at low temperature and pressure (below 650 K at 700 torr or here below 200 ºC at 1.5 bar). This is, however, less important at elevated conditions during compression stroke or close to TDC when the in-cylinder mixture temperature rises to an autoignition trigger point or pilot diesel is injected. At high temperatures, HO$_2$ enhances the production of OH radicals via direct dissociation of H$_2$O$_2$ that initiates high temperature (main) combustion around TDC$_{ex}$.

In summary, the high temperatures (300–320 ºC) during MeOH in-
increased with the increase in early main-combustion stages. On the other hand, we observe that during the late cycle because of dilution induced by greater fuel burned during the early stage of the main combustion and 2) slow burning the combustion at richer conditions is found to be highly sensitive to rates, early autoignition and strong acoustic oscillations. In addition, the investigation at richer conditions is limited by high pressure-rise MeOH-diesel DF combustion in the NVO mode. It is worth noting that Therefore, a leaner condition (\( \lambda \)) is selected here and only is varied, while keeping other operating conditions constant, as outlined in Table 5. In this case study, several performance parameters and emission characteristics are analyzed, however, for brevity only the relevant cylinder pressure and HRR data is presented in Fig. 7.

Temperature is an important parameter for determining the Arrhenius rate of chemical reactions. In general, high temperatures accelerate the rates of combustion by increasing the kinetic energy and collision probability of the fuel/oxygen molecules within a mixture. Thus, high temperatures can be advantageous for the pre-flame slow oxidation and promoting the autoignition of the main combustion. It is observed from Fig. 7(A) that the peak cylinder pressure and peak HRR increase with the increase in \( T_{\text{air}} \) from CAT40 to CAT70. In particular, the pre-flame slow oxidation processes (reactivity) are enhancing from CAT40 to CAT70 and producing a larger amount of reactive species (as discussed in Section 3.2). Whereby these promote early autoignition of the MeOH-air mixture at respectively higher compression temperatures. As an evidence, CAT60 and CAT70 depict autoignition even earlier than pilot diesel injection timing. Nevertheless, CO emissions decrease because of completed combustion with monotonically higher exhaust-gas temperatures. The increased temperature of residual gases has also a pronounced effect of promoting the slow pre-flame oxidation processes before ignition occurs.

In this case, the main-combustion duration (interval between CA2 and CA90) is found to increase \(-2^\circ\text{CA} \) per 10 \(^\circ\text{C}\) increase in \( T_{\text{air}} \) although the start of combustion (\(-CA2\)) also advances \(-2^\circ\text{CA} \) per 10 \(^\circ\text{C}\). This particular behavior is attributed to 1) a faster burning rate during the early stage of the main combustion and 2) slow burning during the later cycle because of dilution induced by greater fuel burned in early main-combustion stages. On the other hand, we observe that gIMEP increases with the increase in \( T_{\text{air}} \) from CAT40 to CAT50, however, for CAT60 to CAT70 gIMEP remains unchanged (see Table 5). This is due to early combustion phasing with respect to TDC and negative work produced.

In brief, increased \( T_{\text{air}} \) enhances the reactivity or pre-flame slow oxidation processes, which consequently promotes autoignition and improves the overall combustion. We found that at the lean condition, increased \( T_{\text{air}} \) advances the combustion phasing (CA10 and CA50, however, delays CA90), helps to increase \( \eta_{\text{accQ}}, \eta_{\text{IMEP}} \), \( \eta_{\text{sec}} \) and \( \eta_{\text{comb}} \) accompanied with significantly reduced THC and CO emissions.

3.3. Effect of charge-air temperature (\( T_{\text{air}} \))

Next, we demonstrate the impact of \( T_{\text{air}} \) on the performance of MeOH-diesel DF combustion in the NVO mode. It is worth noting that the investigation at richer conditions is limited by high pressure-rise rates, early autoignition and strong acoustic oscillations. In addition, the combustion at richer conditions is found to be highly sensitive to \( T_{\text{air}} \) and therefore, a leaner condition (\( \lambda_{\text{MeOH}} = 3.8 \)) is selected here and only \( T_{\text{air}} \) is varied, while keeping other operating conditions constant, as outlined in Table 5. In this case study, several performance parameters and emission characteristics are analyzed, however, for brevity only the relevant cylinder pressure and HRR data is presented in Fig. 7.

Temperature is an important parameter for determining the Arrhenius rate of chemical reactions. In general, high temperatures accelerate the rates of combustion by increasing the kinetic energy and collision probability of the fuel/oxygen molecules within a mixture. Thus, high temperatures can be advantageous for the pre-flame slow oxidation and promoting the autoignition of the main combustion. It is observed from Fig. 7(A) that the peak cylinder pressure and peak HRR increase with the increase in \( T_{\text{air}} \) from CAT40 to CAT70. In particular, the pre-flame slow oxidation processes (reactivity) are enhancing from CAT40 to CAT70 and producing a larger amount of reactive species (as discussed in Section 3.2). Whereby these promote early autoignition of the MeOH-air mixture at respectively higher compression temperatures. As an evidence, CAT60 and CAT70 depict autoignition even earlier than pilot diesel injection timing. Nevertheless, CO emissions decrease because of completed combustion with monotonically higher exhaust-gas temperatures. The increased temperature of residual gases has also a pronounced effect of promoting the slow pre-flame oxidation processes before ignition occurs.

In this case, the main-combustion duration (interval between CA2 and CA90) is found to increase \(-2^\circ\text{CA} \) per 10 \(^\circ\text{C}\) increase in \( T_{\text{air}} \) although the start of combustion (\(-CA2\)) also advances \(-2^\circ\text{CA} \) per 10 \(^\circ\text{C}\). This particular behavior is attributed to 1) a faster burning rate during the early stage of the main combustion and 2) slow burning during the later cycle because of dilution induced by greater fuel burned in early main-combustion stages. On the other hand, we observe that gIMEP increases with the increase in \( T_{\text{air}} \) from CAT40 to CAT50, however, for CAT60 to CAT70 gIMEP remains unchanged (see Table 5). This is due to early combustion phasing with respect to TDC and negative work produced.

In brief, increased \( T_{\text{air}} \) enhances the reactivity or pre-flame slow oxidation processes, which consequently promotes autoignition and improves the overall combustion. We found that at the lean condition, increased \( T_{\text{air}} \) advances the combustion phasing (CA10 and CA50, however, delays CA90), helps to increase \( \eta_{\text{accQ}}, \eta_{\text{IMEP}} \), \( \eta_{\text{sec}} \) and \( \eta_{\text{comb}} \) accompanied with significantly reduced THC and CO emissions.

3.4. Effect of MeOH lambda (\( \lambda_{\text{MeOH}} \))

Relative air/fuel ratio is one important parameter for characterizing combustion. It indicates the quality of air-to-fuel mixture, particularly, the flammability and reactivity stratification of the in-cylinder mixture. To understand how relative air/fuel ratio affects the performance of MeOH-diesel DF combustion in the NVO mode, the experiments (ML32 - ML38) are performed for constant operating conditions at varying \( \lambda_{\text{MeOH}} \) as listed in Table 5. It should be noted that \( \lambda_{\text{MeOH}} \) is increased by increasing \( P_{\text{air}} \), which correspondingly increases the charge-air quantity. For brevity, only the relevant results of cylinder pressure and HRR are presented in Fig. 8.

It can be seen from Fig. 8(A) that as \( \lambda_{\text{MeOH}} \) increases from ML32 to ML38, the peak cylinder pressure and peak HRR decrease although the compression pressure increases. This may be due to over-lean mixtures, which subsequently leads to an incomplete combustion, a reduced combustion efficiency and lower exhaust-gas (or residual gas) temperature. For instance, the measured exhaust gas temperature decreases from 235 \(^\circ\text{C}\) to 195 \(^\circ\text{C}\) when \( \lambda_{\text{MeOH}} \) is increased from 3.2 to 3.8. However, here it is interesting to note that despite the respective lean mixtures, the start of combustion timing (\(-CA2\)) is maintained from ML32 to ML38. This may be attributed to increased amount of reactive species produced during the pre-flame slow oxidation processes with the increased \( \lambda_{\text{MeOH}} \) value.

For the same NVO period, increased boost pressure (increased \( \lambda_{\text{MeOH}} \)) leads to retain a greater amount of residual gases inside the cylinder. As noted in Section 3.2, for the constant \( \lambda_{\text{MeOH}} \), the increased residual gases enhance the reactivity of the fresh MeOH-air mixture as a result of pre-flame slow oxidation processes with increasing exhaust-gas temperatures. Likewise, here at varying \( \lambda_{\text{MeOH}} \), the increased residual gases enhance the reactivity however, with decreasing exhaust-gas temperatures. This indicates that the composition of exhaust-gases is also an important factor influencing the ignition timing and determining the characteristics of the pre-flame slow oxidation. However, despite the maintained start of combustion timing, over-lean mixtures (or dilution) prevail around TDC (while MeOH is being injected during the intake stroke, the mixture is rather heterogeneous than homogenous) and worsen the main combustion (reduced \( \eta_{\text{comb}} \)) as \( \lambda_{\text{MeOH}} \) increases from
adverse cooling effects on the cylinder temperature and therefore targeted performance criteria. With an increase in the engine load, more effective cooling is required, which results in increased AccQ, gIMEP, and COVIMEP for the studied test points adequately satisfy the target limit of $\eta_{comb}$. On the other hand, a least amount of residual gases (EVC6 ~ NVO105) is utilized for EL40, which is necessary to avoid dilution of the in-cylinder mixture. As mentioned earlier in Section 3.2, the DF combustion around TDC improves when reactivity (pre-flame slow oxidation processes) of the fresh MeOH-air mixture also utilizes the trapped mass of residuals, adapting thermodynamic state (essentially cylinder pressure) of the exhaust gases at EVO. This is also confirmed using the TPA method, as mentioned in Section 2.3.1. For EL50, ~30 mg lower mass of residuals is estimated at IVC compared to EL40 for the same NVO period and charge-air quantity. Furthermore, the estimated iEGR (Equation 8) values for the test points EL40 – EL90 lie in the range of 20–30%. The validation of the simulated cylinder pressures is presented in Fig. A2 (Appendix A).

Above discussion shows that a certain engine load requires a certain combination of $\lambda_{MeOH}$, $T_{air}$, and amount of residual gases to yield the best target-specific combustion characteristics. Therefore, compared to EL40, the $T_{air}$ requirement for EL50 reduces to 80 °C due to increased mixture flammability at richer conditions. It should be noted that when engine load increases at the same NVO period and charge-air quantity, the trapped mass of residuals decreases correspondingly due to higher thermodynamic state (essentially cylinder pressure) of the exhaust gases at EVO.

3.5. Effect of engine load

3.5.1. Combustion characteristics

In this test case, the DF combustion is studied to demonstrate the ability of NVO mode to control a wide range of engine operating loads (EL40 – EL90) with an adequate combustion stability, improved emissions, and positive efficiency gains. For this purpose, target-specific performance criteria are set for the experiments including 1) CA50 ≤ 2° CA aTDC, 2) $\eta_{comb}$ ≥ 90%, 3) $\eta_{ind}$ ≥ 50%, and 4) COVIMEP ≤ 5%. The obtained combustion characteristics for the test points from EL40 to EL90 are presented in Figs. 9 - 11. It can be seen that the peak HRR and AccQ increase with an increase in the engine load. In addition, CA50, $\eta_{comb}$, $\eta_{ind}$, and COVIMEP for the studied test points adequately satisfy the target-specific performance criteria. With an increase in the engine load, more MeOH amount is to be injected into the cylinder, which may cause adverse cooling effects on the cylinder temperature and therefore deteriorate the ignition of the in-cylinder mixture. However, based on the insight from the previous sections (Sections 3.2-3.4), several possible parameters are adjusted including $\lambda_{MeOH}$, $T_{air}$, and NVO period, as shown in Table 5. The adjustments of these parameters enabled us to reach the target-specific performance.

Realization of the target-specific performance from the MeOH-diesel DF combustion is a challenging procedure, especially at low-load conditions when a small pilot quantity is utilized. EL40 is operated at the leanest methanol lambda of 4.0 ($\lambda_{MeOH} = 4.0$), which is intrinsically implemented due to the minimum operational limits of the charge-air system. At such lean condition, EL40 requires a high $T_{air}$ of 100 to meet the target limit of $\eta_{comb}$. On the other hand, a least amount of residual gases (EVC6 ~ NVO105) is utilized for EL40, which is necessary to avoid dilution of the in-cylinder mixture. As mentioned earlier in Section 3.2, the DF combustion around TDC improves when reactivity (pre-flame slow oxidation processes) of the fresh MeOH-air mixture also improves as a result of increased residual gas amount. However, for EL40 at $\lambda_{MeOH} = 4.0$, a contrary trend is observed i.e., the DF combustion around TDC worsens as residual gas amount increases. This is the same combustion behavior described in Section 3.4, when lean mixtures dominate despite a maintained start of combustion timing (an enhanced pre-flame process). This causes a decrease in the overall $\eta_{comb}$. Nevertheless, EL40 fairly well satisfies the performance criteria except for CA50, which is purposefully adjusted to achieve $\eta_{ind} ≈ 50%$.

At the medium-load range, more residual gases can be utilized with a reduced $T_{air}$, as also richer conditions can be applied to the fresh MeOH-air mixture. The use of greater amount of residual gases without compromising combustion performance is an advantageous procedure. This is because it allows greater waste heat recovery from the exhaust gases, which consequently results in increased indicated and thermal efficiency. Thus, for EL60 and EL70 at relatively rich conditions and with increased mass of residuals, adapting $T_{air}$ is utilized to meet the performance criteria. However, at high-load conditions, using a greater
mass of residuals (longer NVO periods) promotes autoignition and early start of combustion with high pressure-rise rates causing acoustic oscillations. Therefore, in this study, a least mass of residuals with low $T_{air}$ are utilized to control the combustion characteristics of EL80 and EL90. Moreover, high-load conditions are intrinsically operated at richer conditions due to the maximum operating limits of the charge-air system.

From Fig. 11 (as well as Table 5), it can be observed the maximum $\eta_{comb}$ is reaching to ~ 98% and $\eta_{ind}$ to ~ 54%. All test points (EL40 – EL90) show $\eta_{comb}$ and $\eta_{ind}$ way above the set limits. Interestingly, for such high efficiencies, the DF combustion is reliably controlled (CA50 ~ 2°CA aTDC) and has reasonable combustion stability. Many researchers [47,52,53] have declared COVIMEP = 3–5% as a safe range for engine operations. Here, the studied test points (for gIMEP = 5–11 bar) show COVIMEP below 5%, as estimated over 200 combustion cycles. Overall, we note that the DF combustion in NVO mode is potentially controllable and capable of yielding noticeable high-performance characteristics at a wide range of engine operating loads.

3.5.2. Emission characteristics

Next, we discuss the emission characteristics of the DF combustion at varying engine loads. Fig. 12 presents emission data for the studied test points EL40 – EL90. In NVO mode, the combustion is observed to produce quite low THC, CO, and NOx emissions despite the challenges associated with low/high load operations, methanol’s high octane-number and its latent heat of vaporization. This is because of the hot residual gases from the previous cycle that enhance the overall combustion efficiency, as discussed above. Hot residuals help to increase the reactivity of fresh MeOH-air mixture by possibly commencing pre-flame slow oxidation processes before TDCf. It is worth mentioning that here the molecular weight of THC (containing mostly oxygenated hydrocarbons e.g., formaldehydes) is adjusted for oxygen part so that 50% higher THC emissions could be estimated (also remarked in Section 2.3).

In general, THC and CO are produced due to an incomplete combustion, whereas CO2 is produced as a result of complete combustion. During a combustion process, NOx usually forms as thermal NOx due to locally high in-cylinder temperatures. Therefore, in this study, as more fuel burns completely, more CO2 and NOx emissions tend to produce
with correspondingly lower THC and CO emissions. THC is regarded as the unburned fuel and CO produces as a result of partial oxidation of the fuel depending on the in-cylinder thermodynamic conditions and combustion mechanisms. For instance, EL60 produces higher CO, however, lower THC, CO2 and NOx emissions compared to EL50. This indicates that the slightly lower n_{comb} of EL60 than EL50 is mainly owing to the dilution of the in-cylinder mixture rather than unburned fuel. Apparently, some of the fuel of EL60 was partially oxidized but was not able to fully oxidize to produce higher CO2 and NOx compared to EL50. Typically, CO emissions can be reduced by using high T_{air} and richer conditions, as well as avoiding dilution.

In summary, we found here that at a wide range of engine operating loads, the high fractioned MeOH DF combustion in NVO mode can certainly achieve high efficiency targets (\( \eta_{\text{comb}} \sim 50\% \)) together with low emissions of THC, CO, and NOx. Additionally, the production of low NOx emissions despite the high efficiencies is owing to inherent ability of MeOH to burn with low temperatures. Furthermore, in this study, we utilized the AVL smoke meter (using filter paper method according to ISO 10054) to determine filter smoke number (FSN) and soot concentration in the exhaust. We observed no trace of particles or soot on the filter paper left by exhaust gases in all the test points in the present study.

4. Conclusions

In this study, high fractioned (90\% on energy base) methanol (MeOH) dual-fuel (DF) combustion is experimentally investigated in a single-cylinder research engine. A constant 10\% (energy based) pilot diesel ignites the premixed MeOH-air mixture. Experiments are performed at a constant engine speed of 1500 rpm in the negative-valve overlap (NVO) mode to investigate the effects of varying: 1) NVO period, 2) charge-air temperature (CAT), and 3) methanol lambda (ML), as well as 4) engine load (EL). The combustion at a wide range of engine loads demonstrates the relevance of NVO mode to control and yield high indicated efficiency (\( \eta_{\text{ind}} \sim 50\% \)) together with low exhaust emissions. Furthermore, 1D GT-power model based on Three Pressure Analysis (TPA) method is adopted to estimate particularly mean mixture temperature and amount of residual gases at the measured engine conditions. The main findings of the present study are concluded in the following.

1) NVO mode enables the use of hot-residual gases from the previous cycle, which facilitate the MeOH-diesel DF combustion to achieve high combustion (\( \eta_{\text{comb}} > 90\% \)) and indicated (\( \eta_{\text{ind}} \sim 50\% \)) efficiency more readily than that of the standard valve timing (SVT).

2) Hot residual gases induce a slow pre-flame oxidation processes in the fresh MeOH-air mixture, which generates a pool of reactive species. This increases the reactivity of the in-cylinder mixture to a point that the high-temperature autoignition may occur even earlier than the pilot injection timing. In other words, the slow pre-flame oxidation processes have a direct influence on the reactivity/autoignition of the in-cylinder mixture.

3) The slow pre-flame oxidation processes and the subsequent main combustion are found to be dependent on composition, trapped mass, and thermal state of the residual gases. It is concluded that: a. with an increase in the residual gas amount either by increasing the NVO period or methanol lambda (\( \lambda_{\text{MeOH}} \), boost pressure), the slow pre-flame oxidation is possibly always enhancing regardless of the utilized lean conditions. However, there are negative effects on the main combustion above a certain lean-mixture limit. b. with an increase in charge-air temperature (T_{air}), both the pre-flame oxidation and the subsequent main combustion are improved.

4) For an engine operating load, a certain combination of NVO period, \( \lambda_{\text{MeOH}} \), and T_{air} helps to yield high efficiency together with low pollutant emissions. For instance, at a wide range of operating loads (gIMEP = 5–11 bar), in general, a \( \eta_{\text{comb}} \) of \~ 98\% and \( \eta_{\text{ind}} \) of up to 53\% is achieved together with low exhaust emissions of THC \~ 2 g/kWh (despite 50\% higher estimate), CO = 2 g/kWh, and NOx = 1.1 g/kWh.

5) NVO mode helps to maintain high combustion stability despite a high pressure-rise rate at high loads, and ultra-lean conditions at low loads. On average, the coefficient of variability of gIMEP (COV_{IMEP}) is observed to be maintained at low values around 3.5\%.

In summary, we observe here that despite the utilization of 90\% methanol, the MeOH-diesel DF combustion in NVO mode has the potential to produce high efficiency together with positive environmental impacts. It is believed that with more precise tuning of engine parameters further 0.5–1\% efficiency gains can be achieved. On the other hand, the high efficiency gains based on the partial oxidation of methanol in pre-flame regime provide a pathway for further detailed evaluations of interactions between hot residual gases and fresh MeOH. We believe that this pre-flame oxidation process has a close resemblance to the moderate- or -intense low-oxygen dilution (MILD) combustion, generally also known as the flameless combustion.

CRediT authorship contribution statement

Zeeshan Ahmad: Conceptualization, Investigation, Software,
Validation, Methodology, Data curation, Formal analysis, Visualization, Writing – original draft, Writing – review & editing. **Ossi Kaario:** Resources, Supervision, Writing – review & editing. **Cheng Qiang:** Investigation, Validation. **Martti Larmi:** Supervision, Funding acquisition, Writing – review & editing.

**Declaration of Competing Interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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**Appendix A**

**Fig. A1.** A schematic illustration of the standard valve timing (SVT) configuration and positive valve overlap (PVO). IVO = 356 °CA aTDC, IVC = -155 °CA aTDC, EVO = 150 °CA aTDC, EVC = 380 °CA aTDC. SOI\textsubscript{MeOH} = 365 °CA aTDC, SOI\textsubscript{pilot} = -6 °CA aTDC.

**Fig. A2.** Validation of GT power model with experimental cylinder pressures to calculate the mass of trapped residuals and the fraction of internal EGR (iEGR) in negative valve overlap (NVO) mode.

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